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EVALUATION OF REMEDIATION OF GROUNDWATER BY NATURAL ATTENUATION

L. E. CARPENTER AND COMPANY WHARTON, NEW JERSEY SRP NO. NJD002168748

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Section 1 Introduction

1.1 Background

The L.E. Carpenter facility is located at 170 North Main St., Borough of Wharton, Morris County, New Jersey (see Figure 1). The site history has been summarized in the 1992 Final Supplemental Remedial Investigation Addendum Report (Weston, 1992a), and is summarized briefly here. The site was operated as a manufacturing facility for vinyl wall coverings from 1943 to 1987. Subsequently, it has been used as a warehouse and manufacturing facility. Key compounds used in the manufacturing process or plant operations included a plastisol compound, coal, #6 fuel oil, xylenes, and methyl ethyl ketone. Some waste materials were disposed into an unlined, on-site surface impoundment. In addition, the site had an operating iron mine and forge in the late 1800's to early 1900's. In 1982, an Administrative Consent Order (ACO) was entered into with the NJDEP, followed by a 1983 Addendum, and a 1986 additional ACO.

A map of current conditions at the site is presented as Figure 2. Site remediation activities began in 1982, and have included, but are not limited to, removal of 4,000 cubic yards of waste from the surface impoundment, removal of underground and aboveground storage tanks, floating product recovery, and building demolition and removal. From 1984 to 1992, more than 5,000 gallons of floating free product were removed by a series of recovery wells (Weston, 1992a). Free product recovery continued, and since 1997, enhanced fluid recovery (EFR) wells have been used to recover an additional 2,000 gallons of free product. Currently, the remaining free product covers an area of approximately 100 feet by 600 feet on-site, with typical apparent product thickness values in wells of less than a foot, but ranging up to several feet in a few wells.

Groundwater in the vicinity of the free product zone contains three main constituents of concern that exceed New Jersey groundwater quality standards (NJGQS): ethylbenzene, xylenes, and diethylhexyl phthalate (DEHP), also known as bis-2-ethylhexyl phthalate (NJDEP, 1994). Typically, concentrations of constituents of concern in the groundwater decrease substantially over relatively short distances from the free product area, such that, within 100 to 300 feet downgradient of the free product zone, concentrations are commonly nondetectable.

The pattern of substantial decreases in concentrations of constituents of concern over relatively short distances from the free product indicates that some degree of natural attenuation may be occurring in the groundwater system.

1.2 Objective

The objective of this report is to evaluate the extent to which natural attenuation is reducing the mass and concentration of constituents of concern in the groundwater at the site downgradient of the free product plume. This evaluation will help to assess whether natural attenuation of the constituents of concern could be an effective component in the overall site remediation strategy.

Section 2 Approach for Evaluation Natural Attenuation

Significant documents that present guidelines for evaluating natural attenuation in groundwater include the use of Monitored Natural Attenuation at Superfund, RCRA Corrective Action, and Underground Storage Tank Sites (USEPA, 1999), ASTM Standard Guide for Remediation of Groundwater by Natural Attenuation at Petroleum Release Sites (ASTM, 1998), Technical Protocol for Implementing Intrinsic Remediation with Long-Term Monitoring For Natural Attenuation of Fuel Contamination Dissolved In Groundwater (Wiedemeier, et al., 1995), and Technical Protocol For Evaluating Natural Attenuation of Chlorinated Solvents In Ground Water (Wiedemeier et al., 1998). While dealing mainly with fuel-related constituents and chlorinated solvents, the technical protocols presented in these documents, are not strictly chemical-specific; rather, the approach is broadly applicable to a variety of contaminants where natural attenuation is being evaluated. The general technical protocol, outlined in the 1998 report (Wiedemeier et al., 1998) includes the following elements:

- Develop a preliminary conceptual model of contaminant behavior at the site
- Characterize the site hydrogeologic conditions
- Evaluate chemical indicators of natural attenuation
- Refine the conceptual model, and conduct pre-modeling calculations of contaminant behavior
- Simulate natural attenuation using a solute fate and transport model
- Conduct a receptor exposure pathways analysis
- Evaluate supplemental source removal options
- Prepare a long-term monitoring plan
- Present findings

These elements of a natural attenuation evaluation have been, or are being, addressed as follows. This report will present a preliminary conceptual model of contaminant behavior at the site, and additional data will then be used to refine the conceptual model. The site hydrogeologic conditions have previously been characterized in detail (Weston, 1992a) and are summarized briefly here. Chemical indicators of natural attenuation are evaluated in this report. Premodeling calculations of chemical behavior are presented, and natural attenuation simulations using a solute fate and transport model are presented in this report.

An exposure assessment completed previously for the site indicated no current human exposure pathway completed through groundwater under current conditions, although there was a potential for future use of groundwater identified for the site (Weston, 1992b). Source removal is currently ongoing at the site, in the form of EFR of free product. This source removal action follows removal of contaminated soil from the surface impoundment, and free product removal by extraction wells that was initiated in 1984. Finally, a long-term monitoring plan has been implemented for the site. A revised plan may be considered to track progress in natural attenuation of the contaminants, if natural attenuation becomes an important part of the overall remedial strategy.

The following sections present the results of conceptual model development, evaluation of natural attenuation indicators, premodeling calculations of contaminant behavior, and development of a fate and transport model to evaluate natural attenuation of key contaminants.

Section 3 Preliminary Conceptual Model

The key contaminants of concern in groundwater associated with the free product plume at the site include xylenes, ethylbenzene, and DEHP (NJDEP, 1994). The sources of these constituents are not well known, but leaks from the fuel oil tanks or lines that previously existed in the southern portion of the site are potential sources for xylenes and ethylbenzene. Xylenes were also used in the manufacturing process, and may have leaked from storage tanks or transfer lines located on-site. DEHP is an important component of plasticizers, and it may have been stored and used in the manufacturing process, and may have leaked from storage tanks or lines on-site. The on-site surface impoundment may have been a source of all three of the constituents listed above, since it contained waste materials from the site and was unlined.

Xylenes and ethylbenzene are readily biodegradable (Howard, 1989). Typically, xylenes and ethylbenzene are attenuated over very short distances away from a source. Evidence provided by Rice et al. (1995) from a study of 1,100 leaking underground storage tanks in California indicates that most BTEX plumes are stable or moving slowly relative to groundwater, and are limited by natural attenuation, especially involving biodegradation. Xylene- and ethylbenzene-degrading bacteria utilize electron acceptors for energy, such as (in order of preference) oxygen, nitrate, manganese, iron, and sulfate. Under aerobic (oxygen present) conditions, the biodegradation of ethylbenzene and xylenes will be most rapid; total removal of 7.5 mg/L total BTX (benzene, toluene, and xylenes) in 1.2 years was observed under aerobic conditions in shallow groundwater of one study (Barker et al., 1987). Under anaerobic conditions, xylenes and ethylbenzene are also readily degraded, but at a slower rate (Barker et al., 1987).

DEHP degrades rapidly under aerobic conditions in surface water, with half-lives of 2 to 3 weeks reported (Wolfe et al., 1980; Howard, 1989). Under anaerobic conditions in water sediment mixtures, no biodegradation has been reported in some studies (Howard, 1989). While there are few studies of DEHP biodegradation in groundwater reported in the literature, evidence indicates that, under aerobic conditions, a substantial degree of biodegradation will occur in groundwater, as it does in surface water (Howard, 1989). For example, Graves et al. (1994) reported substantial oxygen-limited biodegradation of DEHP concentrations by indigenous bacteria in laboratory microcosms simulating aerobic groundwater conditions and nutrient additions.

Examination of the distribution of xylenes, ethylbenzene, and DEHP in groundwater at the site shows a consistent pattern, in that concentrations are high close to the free product plume, and

decrease substantially over short distances away from the plume. Highest concentrations tend to be in the shallow groundwater, which is consistent with the free product plume being the source of dissolved constituents in the groundwater.

A preliminary conceptual model of the behavior of ethylbenzene (EH), xylenes (X), and DEHP at the site is that these constituents dissolve out of the free product plume and are carried downgradient in the groundwater. Groundwater flow downgradient of the free product is toward the north and east, toward the drainage ditch that extends east to west along the northern property line of the L.E. Carpenter site. The contaminants migrate laterally largely through the shallow groundwater, and are dispersed somewhat along the flow path. However, substantial decreases in concentration downgradient of the free product occur largely as a result of biodegradation of the three major constituents of concern, as well as other VOCs. The biodegradation occurs first under anaerobic, reducing conditions in the near vicinity of the free product, then under progressively oxidizing conditions downgradient of the free product. Within 200 to 300 feet of the downgradient edge of the free product, concentrations of X, E, and DEHP have decreased to nondetectable levels, as biodegradation has apparently removed the dissolved contaminants from the groundwater. Ten years of chemical data indicate that the plume of dissolved constituents of concern is relatively stable or shrinking over time, as free product is removed and the dissolved constituents are biodegraded along the flow path.

This preliminary conceptual model will be evaluated in consideration of the data and discussions provided in the following section. A refined conceptual model will be presented later in the report.

Section 4 Site Hydrogeologic Conditions

The hydrogeologic conditions of the site have been presented in the 1990 Report of Revised Investigation Findings, L.E. Carpenter & Company, by GeoEngineering, Inc., and in the 1992 Weston report entitled Final Supplemental Remedial Investigation Addendum for L.E. Carpenter and Company. In the western portion of the site, the upper stratum is composed largely of sand and gravel outwash, overlain by a thin layer of fill. In the eastern portion of the site, this outwash unit is overlain by relatively low-permeability silt and clay alluvium from the Rockaway River, that occurs within the upper 10 to 15 feet, based on the 1992 Weston Supplemental Remedial Investigation Report. However, later borings for EFR wells and recent exploratory borings identified a significant amount of sand and gravel in the upper alluvial unit at a number of locations within the free product zone that were previously identified as having silt alluvium. It appears that the upper alluvial unit is variable laterally, including areas of silty sand as well as silt and clay. Within the upper alluvial unit, borehole geophysics conducted at selected locations identified a silt/clay layer of variable thickness up to 2 to 5 feet, at depths ranging from about 5 to 10 feet. The upper alluvial unit of silt, sand, and clay overlies deeper permeable units (up to 170 feet total in thickness) composed of sand and gravel outwash and stratified drift.

Figure 3 shows a conceptual cross section of geologic conditions at the site, along an east-west transect. The cross section was modified slightly from Weston (1992) based on later data and extends through the free product zone, and through the downgradient monitoring locations in the groundwater plume. As shown in the cross section, geologic conditions are variable and complex at the site, with cross-cutting geologic contacts, geologic facies changes over short distances, and significant permeability differences across the site at various depths.

Free product occurs in the upper 15 feet of formation at the site, and in most cases is within 5 to 10 feet of the surface. At some locations, the free product occurs within shallow alluvium or fill, composed mainly of silt and clay, whereas at other locations it occurs within the silty sand.

Shallow groundwater flow is substantially affected by adjacent surface water bodies. Groundwater is recharged from the Rockaway River, and discharges into the drainage ditch that lies to the north and east of the free product zone. As shown on Figure 4, shallow groundwater flows from the free product area to the north and east, toward the drainage ditch that borders the site and the Air Products property. Except for the recharge zone adjacent to the

Rockaway River, vertical gradients are upward toward the shallow groundwater zone and surface water bodies.

The silty sand that occurs immediately beneath the surface (western portion of the site) or beneath the sandy silt/clay alluvium at depths ranging from 0 to 15 feet (eastern portion of the site) has a hydraulic conductivity of approximately 37 feet/day (Weston, 1992a). The hydraulic conductivity of the upper stratum of silt and clay alluvium that occurs in the eastern portion of the site has not been measured, but is likely on the order of 1 foot/day or less, based on boring log descriptions. The horizontal hydraulic gradient varies across the site, but is approximately 0.0016 ft/ft, based on examination of equipotential maps from GeoEngineering (1990), Weston (1992a), and RMT (1999). Assuming a typical effective porosity of 0.3 (Freeze and Cherry, 1979), the horizontal groundwater seepage velocity is approximately 73 ft/year in the shallow portion of the outwash sand.

Section 5

Extent of Groundwater Contamination

The extent of groundwater contamination with respect to free product, total BTEX, and DEHP is presented on Figures 5 and 6. Floating free product (also known as light nonaqueous-phase liquid, or LNAPL) exists over an area that is approximately 500 feet east-west by 100 feet north-south. In the free product area, concentrations of xylenes, ethylbenzene, and DEHP are generally high (several thousand $\mu g/L$); DEHP concentrations exceed the solubility limit of 285 $\mu g/L$ at some locations (e.g., at MW-6R), either as a result of incorporation of free product into the water sample, or because the solubility of DEHP has been raised in the presence of other solvent constituents in the groundwater.

Total BTEX concentrations outside the free product area are generally nondetectable on the western and southwestern (upgradient) sides of the free product area, and are present at concentrations up to 9 mg/L to the east and north (downgradient) of the free product area (see Figure 5). The BTEX consists mainly of xylenes, with lesser concentrations of ethylbenzene. Benzene and toluene are generally not detectable in groundwater samples from the site (RMT, 1999). BTEX concentrations decrease substantially with distance away from the free product area; for example, along an easterly direction from the free product, the concentration of total BTEX decreases from 9 mg/L at MW-22, to 0.014 mg/L at MW-25, to less than 0.001 mg/L at MW-21 and MW-14.

DEHP shows a similar pattern to BTEX (Figure 6), with concentrations decreasing over short distances away from the free product plume. For example, concentrations of DEHP decrease from 0.67 mg/L at MW-22, to less than 0.001 mg/L at MW-25, MW-21, and MW-14I.

Concentrations of the constituents of concern are generally highest in the shallow groundwater, because of the proximity to the free product plume. Also, vertical gradients are generally upward along flowpaths originating from the free product area, which tends to keep the concentrations highest near the water table.

The historical record of contaminant concentrations shows that concentrations fluctuated seasonally over a wide range of values (see Table 1 and Figures 7a, 7b, and 7c). During the period from 1989 to 1992 when the remedial investigation was conducted, concentrations were relatively high in most monitoring wells. Concentrations decreased during the 1995 to 1996 period, then rose substantially during 1997, and have since decreased dramatically in downgradient wells. For example, xylene concentrations shown on Figure 7a show that concentrations in MW-22R decreased from 66,000 to 5,200 since 1997. Similarly, ethylbenzene

and DEHP concentrations have decreased dramatically at downgradient wells MW-22 and MW-25, since initiation of the EFR program late in 1997. The results indicate that the plume has shown seasonal fluctuations, but is now stable or decreasing over time.

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Section 6

Natural Attenuation Chemical Indicators

Biodegradation of organic contaminants involves redox (oxidation/reduction) reactions, whereby an electron is transferred from the organic contaminant to an electron acceptor. The highest energy potential, and thus the most preferred pathway, is obtained from using oxygen as the electron acceptor, which occurs under aerobic conditions. If oxygen is absent, various other electron acceptors can be used to provide energy needed to degrade the contaminant. Table 2 shows various electron acceptors, including oxygen, nitrate, iron, sulfate, and carbon dioxide, in order of energy derived (and preference), from highest to lowest. In nature, usually more than one of the electron acceptors are available in the groundwater at any zone. This is because some of the redox reactions have similar energy values, such that more than one reaction is likely to occur at any point. Also, microenvironments present in the subsurface around particulate carbon, iron minerals, etc., cause a range of redox conditions to exist even in a small zone, such as that sampled by a monitoring well. Generally, there are overlapping ranges of the various electron acceptors, such that groundwater samples usually contain more than one of the electron acceptors.

Indicators of natural attenuation of petroleum hydrocarbons (such as xylenes and ethylbenzene) and DEHP, which are degraded primarily under oxidative pathways, include absence or decreased concentrations of dissolved oxygen, nitrates, and potentially sulfate, depressed electrode millivolt values of redox, increased concentrations of CO₂, nitrite or ammonia, dissolved iron, and potential hydrogen sulfide and methane, relative to background (Wiedemeier et al., 1995).

Table 3 presents the concentrations of indicator parameters of natural attenuation. Figures 8, 9, and 10 show the distribution of dissolved oxygen, redox potential, and ferrous iron in groundwater samples. Laboratory data sheets are presented in Appendix A.

Figure 8 shows the depressed concentrations of dissolved oxygen that occur in the vicinity of the free product, and extending downgradient of the free product, to the east and north. Concentrations below 0.5 mg/L of dissolved oxygen typify the area near to, and within, the free product zone, whereas concentrations above 1 mg/L occur in upgradient locations. Because of the difficulty in preventing diffusion of oxygen into groundwater samples during sample collection, the low values of oxygen shown on the figure are probably maximum values.

The distribution of redox potential values in groundwater over the site is shown on Figure 9. In the upgradient (western) end of the site, values of redox potential are positive. Reduced redox

conditions, indicated by negative redox electrode values, occur in the vicinity of the free product zone, and extend downgradient to the east and north, toward the Drainage Channel and monitoring well MW-25. The negative redox values correspond approximately with values of dissolved oxygen less than 1.0 mg/L.

Ferrous iron values in groundwater are presented on Figure 10. Ferrous iron is the chemically reduced form of iron, which is far more soluble. High ferrous iron concentrations are indicative of reducing conditions in the groundwater. Although few in number, the values are consistent with the information provided by dissolved oxygen and redox potential, with indications of oxidized conditions upgradient of the free product zone, and reduced conditions downgradient (east) of the free product.

Table 3 shows the concentrations of other indicators of natural attenuation, general chemistry, and nutrients for biodegradation. Nitrate, which after oxygen is the most preferred electron acceptor utilized by bacteria under oxidizing conditions, is nondetectable in downgradient monitoring wells MW-6R, WP-B6, WP-B10, MW-2R, MW-22R, and WM-25R, again providing evidence of reducing chemical conditions downgradient of the free product. Sulfate concentrations are also substantially reduced in downgradient monitoring well samples. Sulfate becomes reduced to sulfide under very reducing conditions. Methane, a by-product of microbial respiration, is an indicator of microbial activity and also an indicator of chemically reducing conditions. Concentrations greater than 1 mg/L of methane were detected in groundwater samples from monitoring wells that are located downgradient of the free product plume (MW-6R, WP-B6, WP-B10, MW-2R, MW-22R, and MW-25R). For upgradient wells MW-15S, WP-A3, and MW-17S, methane was nondetectable or was present at a concentration of less than 0.1 mg/L.

Bacterial plate counts of water samples from monitoring wells show that high numbers of heterotrophic bacteria colonies are present downgradient of the free product. Over 1,000 colonies/mL were counted in downgradient monitoring well samples. The results of the bacteria plate counts are presented in Appendix A.

Thus, a number of chemical indicators, including dissolved oxygen, redox potential, ferrous iron, methane, nitrate, and sulfate, show that various electron acceptors were depleted in concentration beneath and downgradient of the free product plume, relative to upgradient locations. Downgradient of the free product, conditions are chemically reducing, whereas oxidizing conditions are present at upgradient locations. Heterotrophic bacteria colonies are present at substantially higher numbers in water samples from downgradient of the free product, than in groundwater from upgradient locations. These patterns are consistent with a conceptual model of depleted concentrations of various electron acceptors, such as oxygen, nitrate, sulfate, and carbon dioxide, that are used for energy as the bacteria degrade xylenes,

ethylbenzene, and DEHP. As the groundwater migrates farther downgradient toward MW-22R and MW-25R, the concentrations of the electron acceptors begin to rise again as the groundwater becomes depleted with the contaminants.

Section 7 Bioscreen Fate and Transport Modeling

Fate and transport modeling was conducted to evaluate the behavior of dissolved constituents of concern in the groundwater. Downgradient of the free product zone, concentrations of three major constituents of concern, xylenes, ethylbenzene, and DEHP, decrease from values in the thousands of $\mu g/L$ to below detectable concentrations, over distances of less than 300 feet. Along the same flowpaths, various indicators of biodegradation, including bacteria colony populations, dissolved oxygen, methane, and others are consistent with conditions that would be expected if significant biodegradation were occurring. In this section, changes in the concentrations of the three major constituents of concern along a flowpath are quantified, to evaluate the rate at which the contaminants are being removed from the groundwater.

The shallow groundwater piezometric surface elevation for the 4th quarter of 1999 is mapped on Figure 4. Local groundwater gradients in the shallow portion of the outwash are from the free product zone eastward and northward, toward the Drainage Channel shown on Figure 4. The direction of the apparent hydraulic gradient varies seasonally, with the gradient being directed more easterly or northerly from one period to another. The actual flowpath of groundwater may be skewed more easterly than the hydraulic gradient, based on the pattern of concentrations observed in monitoring wells between the free product and the drainage ditch. For example, from the MW-6 location, it appears that the groundwater flow may be directed eastward toward MW-22, based on higher concentrations at MW-22 than at MW-2 or WP-B10, which are located north of MW-6. Anisotropy of hydraulic conductivity caused by geologic stratification oriented easterly (along the direction of the Rockaway River) can easily cause a preferential flowpath to be skewed easterly (Freeze and Cherry, 1979).

Based on hydraulic head data and concentration patterns, monitoring wells MW-6, MW-22, and MW-25 lie along a generalized flowline that starts at the free product zone and extends eastward, and then northward, toward the Drainage Channel. The position of these three monitoring wells along the generalized flowline, along with availability of monitoring data, makes them useful for evaluating the rate of change of contaminants in the groundwater. An alternate flowpath, northward between MW-6 and MW-2, was also considered in the evaluation of biodegradation.

Groundwater chemical data that were reported in the Supplemental Remedial Investigation Report (Weston, 1992a) are used in this evaluation, because the three monitoring wells (MW-6, MW-22, and MW-25) that lie along the designated downgradient flowline were sampled at that

time (between 1990 and 1992). Recent groundwater data (from April 2000) were also used in this evaluation, and show that the pattern of decreasing concentrations over distance are consistent over the past decade. As shown in Table 1, in the early 1990s, xylene concentrations decreased from 120,000 μ g/L at MW-6, to 18,000 μ g/L at MW-22, and to less than detection at MW-25. In the same three wells, concentrations of ethylbenzene decreased from 16,000 μ g/L to 3,200 μ g/L to less than detection. Similarly, DEHP concentrations decreased from 62,000 μ g/L at MW-6 to 8 μ g/L at MW-25. No DEHP data were available for MW-22 in the 1990 to 1992 sampling period. A similar pattern of sharp decreases of concentration over distance has continued to the present, as shown in Table 1, for monitoring wells WM-6, MW-22, and MW-25.

7.1 Model Code

The analytical screening model for evaluating natural attenuation, Bioscreen (Newell et al., 1996), was used to estimate degradation rates in the groundwater for the three major contaminants in groundwater downgradient of the free product zone: xylenes, ethylbenzene, and DEHP. The Bioscreen model was developed through a cooperative effort between the U.S. Air Force Center for Environmental Excellence and the USEPA. Bioscreen has the ability to simulate one-dimensional advection, two-dimensional dispersion, sorption, and various types of degradation reactions of contaminants that are evaluated. Bioscreen can simulate solute transport without decay, linear (first-order) rate of degradation, or "instantaneous" degradation by reacting the contaminant with rate-limiting nutrients such as dissolved oxygen at specified ratios. Both aerobic and anaerobic reactions can be simulated.

7.2 Model Input Parameters

It is recognized that there is no unique set of parameters that can be identified definitively for the entire site. In nature, there is always uncertainty associated with parameter estimates for any given site. Two sets of model input parameters were used to simulate the transport and fate of xylenes, ethylbenzene, and DEHP, in order to account for uncertainty in several of the input parameters. By utilizing two sets of input parameters that bracket a likely range of parameter values, the results derived from each parameter set can be compared to determine if the conclusions would change if the model were calibrated to a different set of parameters that were used within a likely range for the parameters.

Measured values of hydraulic conductivity and hydraulic gradient were used to calculate the groundwater seepage velocity for the model, as discussed above. The resulting groundwater seepage velocity is calculated to be 73 feet per year. This value was utilized in both Parameter Sets 1 and 2.

The advective migration of the three dissolved contaminants is governed by the groundwater seepage velocity and by retardation due to sorption to aquifer solids. Because the total organic carbon content of the aquifer was not quantified, a site-specific partition coefficient for each contaminant cannot be defined accurately. An estimated value of 0.1 percent organic carbon in the aquifer soil was used to estimate the degree of sorption, and the retardation of each contaminant. A review of literature values for organic carbon reveals that the value of 0.1 percent carbon is a typical value that is useful for estimating the degree of sorption in the aquifer. The 0.1 percent carbon value, and the resulting values for sorption and retardation for each constituent, were kept constant for Parameter Sets 1 and 2.

In Parameter Set 1, the "source" of each contaminant was conservatively simulated as having a large mass, thus representing a large, slowly diminishing source. This conservative approach assumes that the free product is the source of the xylenes, ethylbenzene, and DEHP, and that the free product will remain as a source for a number of years. This approach is conservative but reasonable, because, although the free product is being actively removed at the site, it is likely that some free product will still remain for several years, and will serve as a source for the dissolved contaminants.

The model was calibrated to historical concentrations of xylenes, ethylbenzene, and DEHP along a flowline located downgradient of the free product on site. The selected flowline starts in the free product zone at MW-6, and extends to MW-22 and then to MW-25.

Model input parameters are listed in Table 4, and on Figures 10a, 11a, and 12a for xylenes, ethylbenzene, and DEHP, respectively. In each case, the groundwater seepage velocity (average linear velocity) along the selected flowline is designated as 73 ft/yr, as discussed in Section 4.

The retardation factor due to sorption was calculated for each of the three constituents that were simulated, based on the following equation:

$$R = 1 + (p_b/n_e)(K_{oc})$$

where R= retardation coefficient, p_b = bulk mass density, n_e = effective porosity, and K_∞ = partitioning coefficient for each contaminant to organic carbon. The retardation values were calculated based on literature values for K_∞ (Wiedemeier et al., 1995; Howard et al., 1990), estimated bulk mass density (Freeze and Cherry, 1979), and an estimated concentration of 0.1 percent organic carbon in the aquifer soil, as discussed in Section 4. Resulting retardation coefficient values were 3.4 for xylenes and 4.0 for ethylbenzene. A calculated retardation coefficient of 601 for DEHP (based on its strong tendency to adsorb) produced resulting concentration distributions that could not be calibrated to observed concentrations; therefore, the retardation coefficient was reduced substantially to 9, in order to calibrate the model to

observed concentrations at the site. This approach is justified because DEHP occurs with organic solvents (notably xylenes and ethylbenzene), which can substantially increase the solubility and decrease the tendency of DEHP to adsorb to aquifer solids (USDHHS, 1993).

The biodegradation first order decay constant was a major calibration parameter that was adjusted during the modeling simulations to derive a best fit of simulated concentrations to the observed data. The decay constants used in the calibrated models for Parameter Sets 1 and 2 varied from 1.2/year for xylenes, to 0.8/year for ethylbenzene, and 0.3/year for DEHP.

The simulation time for the model was set at 30 years. The actual dates or period over which contaminant releases might have occurred is not known; however, it is likely that the releases occurred prior to 1982, and after operations began in 1943.

The source thickness was set at 10 feet, based on the tendency of the highest concentrations to be strongly focused in the upper 10 feet of the of the aquifer. The source concentration at the center of the plume was adjusted during calibration to derive a reasonable fit to observed concentrations especially at the source area (MW-6), after a 30-year simulation time since the release.

The source mass and source half-life were also adjusted during calibration to provide a reasonable fit to the observed data. Source mass values for xylenes, ethylbenzene, and DEHP were conservatively assumed to be infinite for Parameter Set 1, and were set at 70,000, 10,000, and 100,000 kg for Parameter Set 2. The total mass values of xylenes, ethylbenzene, and DEHP in the aquifer include the amount present as free product, and the amount sorbed onto soil. These amounts are not known with accuracy; however, the mass values used in the model were adjusted in conjunction with the source concentration terms and biodegradation reaction rates, to achieve a reasonable fit to the observed concentrations on site. If most of the source mass of the three constituents is present in the free product, the combined source mass of these three constituents is constrained by the volume of free product at the site. Based on estimates conducted recently (RMT, 1999), the free product source mass is estimated to be approximately 44,000 gallons (equivalent to approximately 160,000 kg if the average density is 0.94). Of the total mass of 160,000 kg of free product, some is likely composed of different constituents than the three constituents simulated here. Analyses of free product samples taken in 1995 from monitoring wells MW-1R, MW-11S, and MW-6R indicated that the free product was composed of approximately 35 percent DEHP, 20 percent xylenes, and 4 percent ethylbenzene (Weston, 1995), based on mean concentrations of three samples. Considering these analytical results, and an estimated free product mass of 160,000 kg, the Parameter Set 2 source mass values are reasonable and realistic estimates of the mass of DEHP, xylenes, and ethylbenzene at the source. The simulations that were conducted using Parameter Sets 1 and 2 both conservatively simulate the source mass estimates at the site. In Parameter Set 1, the source mass is assumed to be infinite. In Parameter Set 2, the source mass is set at 70,000 kg for xylenes, 10,000 kg for ethylbenzene, and 100,000 kg for DEHP. Together, the three constituents total a mass of 180,000 kg in Parameter Set 2, which exceeds the estimate for total mass of free product at the site, 160,000 kg.

7.3 Model Results

Results of the Bioscreen modeling of xylenes, ethylbenzene, and DEHP are illustrated on Figures 11a-c, 12a-c, and 13a-c for Parameter Set 1, and Figures 14a-c, 15a-c, and 16a-c for Parameter Set 2. As shown on Figure 11b, Parameter Set 1 values result in a reasonable fit to the observed xylenes concentration data at the source (MW-6, located 0 feet downgradient) and at MW-25, located approximately 300 feet downgradient. The observed concentration of xylenes at MW-22, located approximately 220 feet downgradient, is somewhat higher than predicted by the model, and the higher concentration observed at that location probably reflects a lower rate of biodegradation closer to the source, where anaerobic conditions predominate. It is likely that the rate of biodegradation increases in the downgradient direction (toward MW-25), as conditions become more oxidizing away from the free product. Research has shown that aerobic conditions are necessary for biodegradation of DEHP to occur, and aerobic conditions greatly enhance the rate of biodegradation of xylenes and ethylbenzene (Howard, 1989; USDHSS, 1993).

The model results indicate that, if there were no degradation of xylenes at the site, the expected concentration of xylenes 320 feet downgradient of the source (at MW-25) would be approximately 135 mg/L (see Figure 11b); actual concentrations observed were less than 0.001 mg/L. As shown on Figure 11c, the actual dissolved xylene mass in the plume is estimated to be 85 kg, compared to a calculated value of 4,075 kg if there were no biodegradation. Thus, the model results indicate that approximately 98 percent of the dissolved to high anylenes have biodegraded over the past 30 years, resulting in the low concentrations at the downgradient portion of the attenuated plume that occurs today.

The results from Parameter Set 2 for xylenes are similar to those from Parameter Set 1, as shown on Figures 14b and 14c. The Parameter 2 model results indicate that the calculated dissolved mass of xylenes in the plume is 87 kg, compared to a calculated dissolved mass of xylenes of approximately 3,600 kg if there were no biodegradation. As with Parameter Set 1, the model results indicate that approximately 98 percent of the dissolved mass of xylenes was biodegraded over the past 30 years.

The results of Bioscreen modeling of ethylbenzene using Parameter Set 1, shown on Figures 12b and 12c, indicate that an actual dissolved mass of ethylbenzene in the groundwater plume is approximately 10 kg, compared to a calculated dissolved mass of 832 kg if there were no biodegradation. A 99 percent removal of dissolved ethylbenzene by biodegradation is estimated. Ethylbenzene modeling using Parameter Set 2 is summarized in Table 4 and on Figures 15a, 15b, and 15c. Similar to the results using Parameter Set 1, a dissolved plume mass of 16 kg ethylbenzene is calculated, compared to an estimated 500 kg of dissolved ethylbenzene after 30 years, if there were no biodegradation. The results for Parameter Set 2 indicate a 97 percent removal of dissolved ethylbenzene by biodegradation, compared to 99 percent using Parameter Set 1.

DEHP model results are shown on Figures 13b and 13c, for Parameter Set 1. With an infinite source assumed, a plume mass of approximately 3,600 kg of DEHP is calculated by the model after a 30-year period, if there were no biodegradation. The actual dissolved mass of DEHP in the plume is calculated to be 88 kg, a reduction of 98 percent that is attributed to biodegradation. Using Parameter Set 2 with a finite source mass of 100,000 kg, the model results indicate that 1,860 kg of DEHP would be dissolved in the plume after 30 years, if there were no biodegradation (see Figures 16a, 16b, and 16c). The actual mass of DEHP dissolved in the plume is calculated to be 54 kg, a reduction of 97 percent attributed to biodegradation.

Bioscreen modeling was also conducted along a second potential flowline, between MW-6 and MW-2. Chemical data from the 1992a Weston report was used to evaluate the degree of biodegradation that may be occurring in the aquifer. Model input and output details for xylenes, ethylbenzene, and DEHP are summarized on Figures 17a-c, 18a-c, and 19a-c. Similar to the MW6/MW22/MW25 flowpath, the reduction in concentrations of xylenes, ethylbenzene, and DEHP between MW6 and MW2 is substantial, such that concentrations are reduced to 1/100 or less of the source concentration over a distance of approximately 100 feet.

Current conditions at the site were also simulated, using the same Parameter Set 1 values that were used to simulate 1990-1992 conditions. The reaction rates from the previous simulations were held constant, along with all other Parameter Set 1 values; only the source concentration terms were adjusted to simulate current concentrations at MW-6R. Concentrations of the major constituents of concern (xylenes, ethylbenzene, and DEHP) showed a familiar pattern of sharply decreasing values in April 2000, along the MW6R/MW-22R/MW-25R flowline, as seen previously.

The results of the Bioscreen modeling of current concentrations are shown on Figures 20a-c, 21a-c, and 22a-c. The results indicate that a reasonable fit to the current data was obtained using Parameter Set 1 values. Although an improved fit to the observed data could be derived by adjusting the reaction rates slightly, the fit is reasonably good using the Parameter Set 1

values. This indicates that the rates of reaction at the site are relatively stable over time. Additional simulations of April 2000 conditions for the alternate flowline (MW-6R/MW-2R) are presented on Figures 23a-c, 24a-c, and 25a-c for xylenes, ethylbenzene, and DEHP. Using the same Parameter Set 1 values, the model was again able to simulate current conditions at the site reasonably well.

The Bioscreen modeling results indicate that a substantial degree of biodegradation is occurring in the aquifer, resulting in the attenuated plume of xylenes, ethylbenzene, and DEHP that is observed today. Using two different sets of aquifer parameter values that bracket likely conditions at the site, the model results indicate approximately 97 percent or more of the dissolved mass of xylenes, ethylbenzene, and DEHP has been removed by biodegradation. Along a second possible flowpath, Bioscreen modeling also indicates that approximately 98 percent of the dissolved mass of xylenes, ethylbenzene, and DEHP has been biodegraded.

Independent evidence that the groundwater contaminant plume is attenuated before it reaches the drainage ditch located downgradient of the plume is provided by analyses of surface water samples. Analytical results from samples from locations SW-5-1, SW-7-1, and SW-8-1 (see Figure 2) indicate that there were no detectable concentrations of DEHP, xylenes, or ethylbenzene in any of the three samples (RMT, 1998).

Prediction modeling that simulates future plume concentrations was conducted, based on the calibrated model presented above. This modeling tests whether the plume is likely to advance or be stable over time. Concentrations in the plume 20 years into the future were calculated using a conservative assumption of an infinite source. For the MW6/MW22/MW25 flowpath, the model results for xylenes are summarized on Figures 26a-c. Even with the conservative assumption of an infinite source, the results indicate that the xylene plume would be stable, and would not advance. Similar results were obtained for ethylbenzene and DEHP. Actual decreases in source mass, which are occurring due to source recovery actions, would tend to decrease the predicted concentrations in the plume over time, once the source mass became depleted such that dissolved concentrations at the source decrease significantly.

These results are consistent with findings at numerous other sites and laboratory studies, where substantial biodegradation of xylenes, ethylbenzene, and DEHP (under aerobic conditions) have been documented. While there is some uncertainty in the values of several of the aquifer parameters, the fact that similar estimates of biodegradation were obtained using two parameter sets that bracket the range of likely conditions at the site strongly suggests that substantial biodegradation is occurring, and is removing more than 95 percent of the dissolved mass of xylenes, ethylbenzene, and DEHP.

Section 8 Refined Conceptual Model of the Behavior of Constituents of Concern

The conceptual model of the transport and fate of xylenes, ethylbenzene, and DEHP in the aquifer at the L.E. Carpenter site can now be refined based on evidence from biodegradation indicators and biodegradation modeling. The evidence indicates that substantial reductions to nondetectable or near nondetectable concentrations of all three of the targeted constituents occur over flowpaths of 300 feet or less. There is evidence that the rate of biodegradation increases down the flowpath, as conditions become less reducing chemically and oxygen becomes more available to degrading microbes. Although there is a substantial source in the form of free product at the site, the source is being removed with EFR methods, and the dissolved plume is attenuating over short distances. A conservative Bioscreen assessment of the plume using an assumption of an infinite source indicates that more than 95 percent of the dissolved xylenes, ethylbenzene, and DEHP are removed from the plume over flowpaths of 300 feet or less. Furthermore, concentration trends downgradient of the plume are generally decreasing over time. In total, the evidence from groundwater monitoring and modeling indicates that natural attenuation processes, especially biodegradation, are effective at reducing plume concentrations such that the plume is very restricted in length, and is stable or decreasing over time.

Section 9 References

- ASTM, 1998. Standard Guide for Remediation of Groundwater by Natural Attenuation at Petroleum Release Sites. ASTM E 1943-98.
- Barker, J.F., G. Patrick, and D. Major, 1987. Natural attenuation of aromatic hydrocarbons in a shallow sand aquifer. Groundwater Monitoring Review, Winter 1987, pp. 64-71.
- Freeze, A. and J. Cherry, 1979. Groundwater. Prentice-Hall, Inc. New Jersey.
- GeoEngineering, Inc., 1990. Report of Revised Remedial Investigation Findings, L.E. Carpenter & Company, Wharton, New Jersey Site.
- Graves, D.A., C.A. Lang, and J.N. Rightmyer, 1994. Biodegradation of bis(2-ethylhexyl)phthalate, ethylbenzene, and xylenes in groundwater: Treatability study supporting *in situ* aquifer bioremediation. *In* Proceedings of Second International Conference on *In situ* and On Site Bioreclamation.
- Howard, P.H., 1989. Handbook of Environmental Fate and Exposure Data for Organic Chemicals., Vol. I. Large Production and Priority Pollutants. Chelsea, MA, Lewis Publishing.
- Newell, C., R. K. McLeod, and J. Gonzales, 1996. Bioscreen Natural Attenuation Decision Support System, User's Manual, ver. 1.3. EPA/600/R-96/087.
- NJDEP (New Jersey Department of Environmental Protection), 1994. Superfund Record of Decision, L.E. Carpenter/Dayco Corporation Site, Wharton Borough, Morris County, New Jersey. April, 1994.
- Rice, D.W., R.D. Grose, J.C. Michaelsen, B.P. Dooher, D.H. MacQueen, S.J. Cullen, W.E. Kastenberg, L.G. Everett, and M.A. Marino, 1995. California Leaking Underground Fuel Tank (LUFT) Historical Case Analyses: California State Water Resources Control Board.
- RMT, 1998. Second Quarter 1998 Quarterly Monitoring Report. September, 1998.
- RMT, 1999. Quarterly Monitoring Report, 2nd Quarter, 1999, L.E. Carpenter. July 1999.
- USDHHS (United States Department of Health and Human Services, Public Health Services, Agency for Toxic Substances and Disease Registry), 1993. Toxicological Profile for Di (2-Ethylhexyl) Phthalate.

- USEPA, 1999. Use of Monitored Natural Attenuation at Superfund, RCRA Corrective Action, and Underground Storage Tank Sites. OSWER Directive 9200.4-17P.
- Weston, 1992a. Final Supplemental Remedial Investigation Addendum for L.E. Carpenter and Company. July, 1992.
- Weston, 1992b. Baseline Risk Assessment, L.E. Carpenter and Company, Wharton, New Jersey. January 1992.
- Weston, 1995. Quarterly Progress Report, L.E. Carpenter Site, Wharton, New Jersey. Vol. 1 of 2. April, 1995.
- Wiedemeier, T., J. T. Wilson, D. Kampbell, R. Miller, and J. Hansen, 1995. Technical protocol for implementing intrinsic remediation with long-term monitoring for natural attenuation of fuel contamination dissolved in groundwater: U.S. Air Force Center for Environmental Excellence, San Antonio, TX.
- Wiedemeier, T., M. Swanson, D. Montoux, E.K. Gordon, J. T. Wilson, B. Wilson, D. Kampbell, P. Haas, R. Miller, J. Hansen, and F. Chappelle, 1998. Technical protocol for evaluating natural attenuation of chlorinated solvents in ground water: U.S. Environmental Protection Agency National Risk Management Research Laboratory, Cincinnati, OH.
- Wolfe, N. L., D.F. Paris, W.C. Steen et al., 1980. Correlation of microbial degradation rates with chemical structure. Environmental Science and Technology, Vol. 14, pp. 1143-1146.

Table 1
Groundwater Monitoring Data
L. E. Carpenter - Wharton, New Jersey

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,		3	ND	9.7	ND ND	13 8.7	46,000		
		4	ND	8.8	ND ND	11	NS		
	1996	1	ND	24		47	17,000		
	1770	2	NS	NS NS	ND		NS NS		
_		3	ND	6.8	NS ND	NS 4.3	NS NS		
		4	ND	2.3	ND	ND	NS		
. •	1997	1	ND	3.5	ND	1.8	11,000		
	1557	2	ND	1.2	ND	4.2	NS 100		
		3	ND	2.2	ND	12.6	120		
		4	NS	NS NS	NS	NS NS	NS NS		
	1998	1	ND	ND	ND	ND ND	NS NS		
		2	ND	1.0	ND	1.4	710		
		3	ND	1.9	ND	1.2	NS NS		
-		4	ND	9.3	ND	3.3	650		
	1999	1	ND	1.1	ND	2.5	NS		
,		2	ND	0.66	ND	ND ND	3,000		
	,	2 duplicate	ND	0.43	ND	ND			
		3	ND	3.10	ND	2.9	4,400		
٠.		4	ND	0.51	ND	ND	NS 4 000		
	2000	1	ND	0.54	ND	1.6	4,000 NS		
4		2	ND	0.30	ND	ND			
MW-11I	1989	3	ND	ND	88	700	480		
	1990	1	ND	ND ND	ND		ND		
MW-11(IR)	1999	1	ND	ND ND	ND	0.8	ND ND		
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Table 1
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L. E. Carpenter - Wharton, New Jersey

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	1995 1995	3	ND	ND	ND	ND	1.6		
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	1996	1	ND	ND ND	ND	ND	2.6		
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		3	ND	ND	ND	ND ND	NS NS		
		4	ND	ND	ND	ND ND	2.7		
	1997	1	ND	ND	ND	ND	NS NS		
	1777	2	ND	ND	ND	ND	1.6		
		3	1.2	22.1	ND	176	NS		
		4	NS	NS	NS	NS			
	1998	1	ND	ND	ND	ND	NS NS		
		2	ND	0.34	ND	2	24		
		3	ND	ND ND	ND	ND	NS NS		
		4	ND	ND	ND	ND	ND ND		
	1999	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	ND		
		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	ND		
	2000	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	ND		
MW-15S	1989	3	ND	1.9	ND	ND	ND		
	1990	1	ND	ND	ND	ND	ND		
[1995	1	ND	ND	ND	ND	2.4		
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		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	ND		
	1996	1	ND	33	ND	83	NS		
		2	NŠ	NS	NS	NS	NS		
[3	ND	ND	ND	ND	NS		
		4	ND	0.21	ND	1.7	ND		
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Table 1
Groundwater Monitoring Data
L. E. Carpenter - Wharton, New Jersey

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	1995	1	ND	ND	ND	ND	250		
		2	ND	ND	ND	ND	7.2		
		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	2.8		
	1996	1	ND	ND	ND	ND	NS NS		
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		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	1.7		
		4 ^{duplicate}	ND	ND	ND	ND	1.9		
	1997	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	2.2		
•		3	ND	ND	ND	ND	NS		
		4	NS	NS	NS	NS	NS		
	1998	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	1.9		
		2 ^{duplicate}	ND	ND	ND	ND	3.8		
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		4 ^{duplicate}	ND	0.2	ND	0.8	9.8		
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		3	NS	NS	NS	NS	NS
		4	ND	ND	ND	0.63	ND
	1996	1	NS	NS	NS	NS	NS
		2	NS	NS	NS	NS	NS
		3	NS	NS	NS	NS	NS
		4	ND	ND	ND	ND	1.5
	1997	1	NS	NS	NS	NS	NS
		2	ND	ND	ND	ND	NS
		3	NS	NS	NS	NŠ	NS
		4	NS	NS	NS	NS	NS
	1998	1	NS	NS	NS	NS	NS
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		3	NS	NS	NS	NS	NS
		4	ND	ND	ND	ND	6
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		3	NS	NS	NS	NS	NS
		4	ND	ND	ND	ND	40
	2000	1	NS	NS	NS	NS	NS
		2	ND	ND	ND	ND	ND
MW-21	1991	3	ND	ND	ND	ND	ND
	1999	1	ND	ND	ND	ND	ND
		2	ND	ND	ND	ND	ND
		3	ND	ND	ND	ND	ND
		4	ND	ND	ND	ND	ND
	2000	1	ND	ND	ND	ND	6
		1 ^{duplicate}	NS	NS	NS	NS	ND
		2	ND	ND	ND	ND	ND
MW-22	1992	1	ND	ND	3,200	18,000	NS
	1995	1	ND	57	ND	260	6,500
		2	ND	311	ND	955	380
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Table 1
Groundwater Monitoring Data
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(contained)	 	3	ND	359	ND		NS NG		
		4	ND	320		1,320	NS		
	1997	1	NS	NS NS	ND NS	1,330	ND NG		
	1757	2	ND	5,730	ND ND	NS 22,000	NS		
	<u> </u>	3	ND		 	32,900	7,500		
		4	NS	11,400 NS	348	66,000	NS.		
	1998	1	ND		NS 248	NS NS	NS		
	1770	2	ND	4,070	348	20,600	NS		
·		3	ND	2,260	ND	11,300	5,800		
	-	3 ^{duplicate}	ND	ND	ND	ND	NS		
				2,510	ND	11,000	NS		
	1000	4	ND	1,650	ND	7,230	1,100		
	1999	1	ND	18	ND	84	NS		
		2	ND	1,600	ND	7,600	670		
•		3	ND	1,200	42	5,200	NS		
		4	ND	810	ND	3,300	1,200		
•		4 ^{duplicate}	ND	840	ND	3,400	1,600		
	2000	1	ND	360	ND	1,400	NS		
		2	ND	820	ND	3,600	92		
MW-25	1992	1	ND	ND	ND	ND	8 J B		
	1995	1	NS	NS	NS	NS	NS		
		2	ND	ND	ND	ND	1.6		
		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	68		
	1996	1	NS	NS	NS	NS	NS		
		2	NS	NS	NS	NS	NS		
		3	ND	0.34	ND	2.2	NS		
	100=	4	ND	ND	ND	ND	ND		
	1997	1	ND	ND	ND	ND	NS		
	<u> </u>	2	ND	13.5	ND	89	63		
		3	ND	4.1	ND	30.7	NS		
	1000	4	NS	NS	NS	NS	NS		
	1998	1 1 duplicate	ND	0.33	ND	1.5	NS		
			ND	0.39	ND	0.94	NS		
		2	ND	ND	ND	ND	5.3		
		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	1.9		

Table 1
Groundwater Monitoring Data
L. E. Carpenter - Wharton, New Jersey

	SAMPING DATE CHEMICALANALYSIS RESULTS								
	SAME	L		1	FRANCAS	<u>INALYSISIKESI</u>			
		e e Kari	Renzena	Ethylbenzen	Talvana	Total Xylenes	bis-2- Ethylhexylphthalate (DEHP)		
MONITORING			.0-14g/L0	Segue/L)	lim to pg/L	Pg/L *		
NJ			1	700	1,000	40			
RODDISCI	0 10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	ate and 1 Total 1 To 31, 290, 22 and 3	1	350	500	20	30		
MW-25	1999	1	ND	ND		-	30		
(continued)	1977	2	ND	ND	ND ND	ND 14	ND ND		
(contained)		3	ND	0.39	ND	1.4	ND 9.6		
		4	ND	ND	ND	ND ND	ND		
	2000	1	ND	ND	ND	ND	ND		
		2	ND	ND	ND	ND	ND		
MW-30	1995	1	NS	NS	NS	NS	NS		
		2	ND	17	ND	13	45,000		
		3	ND	ND	ND	ND	NS NS		
		4	ND	ND	ND	ND	ND		
	1996	1	ND	ND	ND	ND	NS NS		
		2	NS	NS	NS	NS	NS		
		3	ND	ND	ND	ND	NS		
		4	NS	NS	NŠ	NS	NS		
	1997	1	NS	NS	NS	NS	NS		
		2	ND	ND	ND	ND	2.2		
MW-30S	1997	1	ND	0.2	ND	1.0	NS		
		2	NS	NS	NS	NS	NS		
		3	ND	4.1	ND	30.7	NS		
		4	NS	NS	NS	NS	NS		
Trip Blank	1995	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	NS		
		3	ND	ND	ND	ND	NS		
		4	ND	ND	ND	ND	NS		
	1996	1	ND	ND	ND	ND	NS NS		
		2	NS	NS	NS	NS	NS		
		3	ND	ND	ND	ND	NS		
	100-	4	ND	ND	ND	ND	NŠ		
	1997	1	ND	ND	ND	ND	NS		
		2	ND	ND	ND	ND	ND		
		3	ND	ND	ND	ND	ŊŚ		
	1000	4	NS	NS	NS	NS	NS		
	1998	1	ND	ND	ND	ND	NS		
,	1000	2	ND	ND	ND	ND	ND		
	1998	3	ND	ND	ND	ND	NS		
	1000	4	ND	ND	ND	NS	1.3		
	1999	1	ND	ND	ND	NS	ND		
•		3	ND ND	ND	ND	NS	ND		
	-		ND ND	ND ND	ND	ND	ND		
		4	ND	ND	ND	ND	NS		

Table 1 **Groundwater Monitoring Data** L. E. Carpenter - Wharton, New Jersey

	I and			i vilanto			
MUNITORING		NG DATE	Benzene	Ethylbenzene		NALYSIS RESI Total Xylenes	LTS bis-2-Ethylhexylphthalat (DEHP)
WELLS		QUARTER	se pg/L	inglate.	HB/L	SCHALES.	ia μg/L
NJD	till and the second second second	the same of the sa	1	700	1,000	· 40	30
ROD DISCI	IARGE CRI	TERIA	1	350	500	20	30
Trip Blank	2000	1	NS	NS	NS	NS	ND
(continued)		1	NS	NS	NS	NS	ND
		2	ND	ND	ND	ND	NS
Field Blank	1995	1	ND	ND	ND	ND	ND
		2	ND	0.73	ND	ND	1.3
		3	ND	ND	ND	ND	NS
		4	ND	ND	ND	ND	ND
	1996	1	ND	ND	ND	ND	NS
•		2	NS	NS	NS	NS	NS
		3	ND	ND	ND	ND	NS
		4	ND	ND	ND	ND	ND
	1997	1	ND	ND	0.2	ND	NS
4		2	ND	ND	ND	ND	NS
		3	ND	NĎ	ND	ND	NS
		4	NS	NS	NS	NS	NS
	1998	1	ND	ND	ND	ND	NS
		2	ND	ND	ND	ND	NS
		3	ND	ND	ND	ND	NS
		4	ND	ND	ND	ND	1.3
	1999	1	ND	ND	ND	ND	ND
!		2	ND	ND	ND	ND	ND
ļ		3	ND	ND	ND	ND	ND
į		4	ND	ND	ND	ND	ND
	2000	1	ND	ND	ND	ND	ND
		1	ND	ND	ND	ND	NS
		1	NS	ÑS	NS	NS	3.2
		2	ND	ND	ND	ND	ND

Values in BOLD BLUE FONT are above BOTH the NJDEP GWQS and the ROD Discharge Criteria.

- Used when comparison against known standards does not apply as the well was not sampled (NS) for a specific analyte mg/L = micrograms per liter.

NJGWQS = New Jersey Groundwater Quality Standards.

ROD = Record of Decision.

NA = not applicable.

NS = not sampled.

ND = no detection.

 $\frac{duplicate}{duplicate} = duplicate sample.$

J = estimated value.

D = secondary dilution.

B = detected in blank.

Table 2
Preferred Reactions by Energy Potential

ELECTRON ACCEPTOR	TYPE OF REACTION	METABOLIC BY-PRODUCT	REACTION PREFERENCE
Oxygen	Aerobic	CO ₂	Most preferred
Nitrate	Anaerobic	N ₂ , CO ₂	+
Ferric Iron (solid)	Anaerobic	Ferrous iron (dissolved)	+
Sulfate	Anaerobic	H ₂ S	+
Carbon dioxide	Anaerobic	Methane	Least preferred

From Newell et al., 1996

Table 3
Natural Attenuation Indicator Parameter Concentrations

	Charles and the second								
SAMPLE LOCATION	DATE	DISSOLVED OXYGEN (mg/L)	REDOX POTENTIAL (at sample) (mV)	NFIRATE (mg/L)	FERROUS IRON (mg/L)	SULFATE:	CO; (at sampling) (mg/L)	HETEROTROPPIE PLATE COUNT (colonies/1 mt.)	
MW-15S	1-Dec-98	0.10	9.31	0.14	0.56	5.3	40	400	
	22-Jan-99	4.25	-8.00	-	-	0.0	30	400	
	16-Apr-99	3.70	7.00	ND	ND	771	25		
	22-Jul-99	0.98	50.5	0.18	1.50	10 July 10 Jul	30	2 1 1 1 2 2 2 2 2 3 3 3 3 3 3 3 3 3 3 3	
WP-A3	1-Dec-98	1.70	152.5	0.57	ND	13.5	36	600	
	22-jan-99	3.40	191.3	1.00	ND	10.0	35	000	
	16-Apr-99	8,80	185.0	0.60	ND		22	A SACRAMAN AND A SACR	
	23-Jul-99	0.98	220,4	0.38	ND	A Land	40		
MW-17S	1-Dec-98	4.70	146.1	0.15	ND	15.1	24	35	
	22-Jan-99	5.80	210.7	0.17	ND		12	39.	
	16-Apr-99	6.30	166.9	ND	ND		20		
	22-Jul-99	1.23	97.5	ND	ND	110	46	elle de la company	
WP-C4 ⁽¹⁾	1-Dec-98	0.00	-144.1	ND	7.60	ND	85	17,100	
	22-Jan-99	0.00	-61.2	0.52	0.25		40	17,100	
	16-Apr-99	0.42	-39.2	ND	0.79	400	65		
WP-B10	1-Dec-98	0.00	-134.7	ND	32.40	7.6	225	19,000	
	22-Jan-99	0.00	-180.6	ND	22.10		295	17,000	
	16-Apr-99	0.11	-196.5	ND	20.80	100	115	Park Control of the Control	
	23-Jul-99	0.35	-209.6	ND	42.70		360		
MW-2R	1-Dec-98	0.20	-168.1	ND	35.40	ND	170	9,000	
	22-Jan-99	0.00	-202.5	ND	19.00		105	7,000	
	16-Apr-99	0.07	-164.3	ND	37.20		170	ACKERS CO. STORY	
	23-Jul-99	0.11	-167.8	ND	44.70		210		
	13-Apr-00	0.00	-143.0	< 0.1	2:60	1 × 5 0		2,000	
WP-B6	1-Dec-98	0.00	-118.4	ND	59.80	ND	370	1,100	
	22-Jan-99	0.00	-163.1	ND	37.90		180		
ļ	16-Apr-99	0.11	-170.1	ND	36.70		180	of the same of the same	
	23-Jul-99	0.18	-165.2	ND	60.20		320	The second second	

Table 3 **Natural Attenuation Indicator Parameter Concentrations**

SAMPLE ** LOCATION	DATE	DISSOLVED OXYGEN::- (ing/L)	REDOX POTENTIAL (at sample) (mV)	NITRATE _(mg/L)	FERROUS IRÓN (mg/L)	SULFATE (mg/L)	CO ₂ (at sampling) (mg/L)	HETEROTROPH PLATE COUNT (colonies/1 mL)
WP-C1 ⁽¹⁾	22-Jul-99	1.59	-147.9	ND	13.80		165	
MW-6R	1-Dec-98	0.00			. 		100	
	22-Jan-99	0.00				AS DESIGN		
	16-Apr-99	0.07			· •••••			FEBRUARY CO.
	23-Jul-99	0.04				and the second		
	13-Apr-00	0.00	-187.0	< 0.1	2.80	<50		Established Commission
MW-22R	22-Jul-99	0.90				- Company		
	13-Apr-00	0.20	-137.0	< 0.1	2.40	<50°°		340
MW-25R	22-Jul-99	1.00						enter de la companya
ing the second s	13-Apr-00	0.50	-71.0	< 0.1	1.80	esil".		200

⁽i) During third quarter 1999 sampling event WP-C4 was found to be destroyed. WP-C1 will replace this well from this point forward.

Table 4
Bioscreen Model Input Parameters

	Longitudinal Dispersivity (feet)	Retardation Coefficient	1st Order Decay Coefficient (per year)	Source Mass (kg)	Source Concentration (mg/L)
Parameter Set 1					
Xylenes (Run 5j)	14	3.4	1.2	Infinite	120
Ethylbenzene (Run 6c)	14	4.0	0.8	Infinite	16
DEHP (Run 7c)	14	9	0.3	Infinite	62
Parameter Set 2					
Xylenes (Run 5i)	14	3.4	1.2	70,000	127
Ethylbenzene (Run 6b)	14	4.0	0.8	10,000	17
DEHP (Run 7d)	14	9	0.3	100,000	63



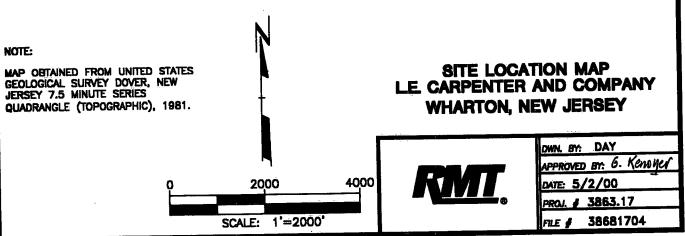
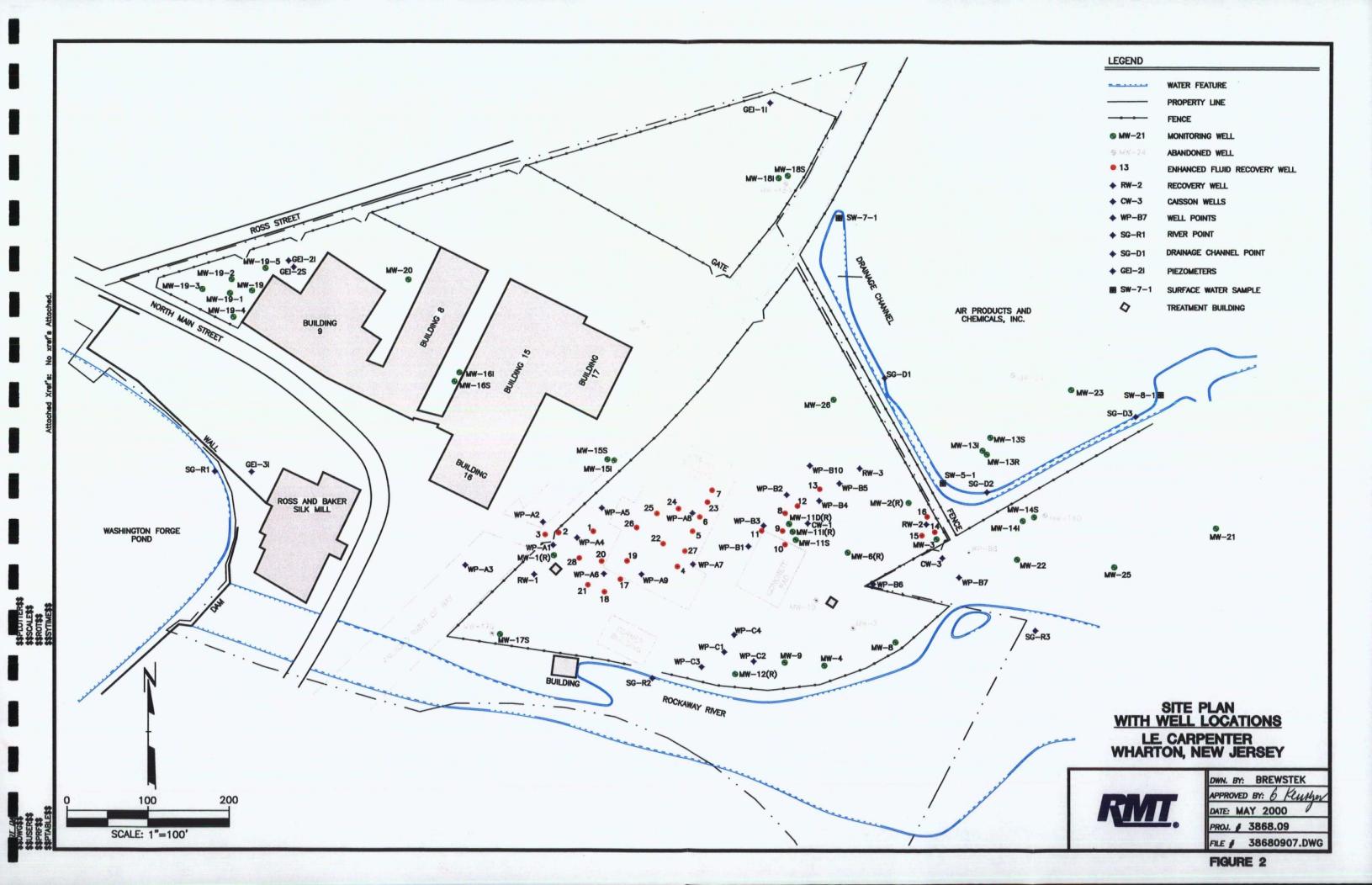
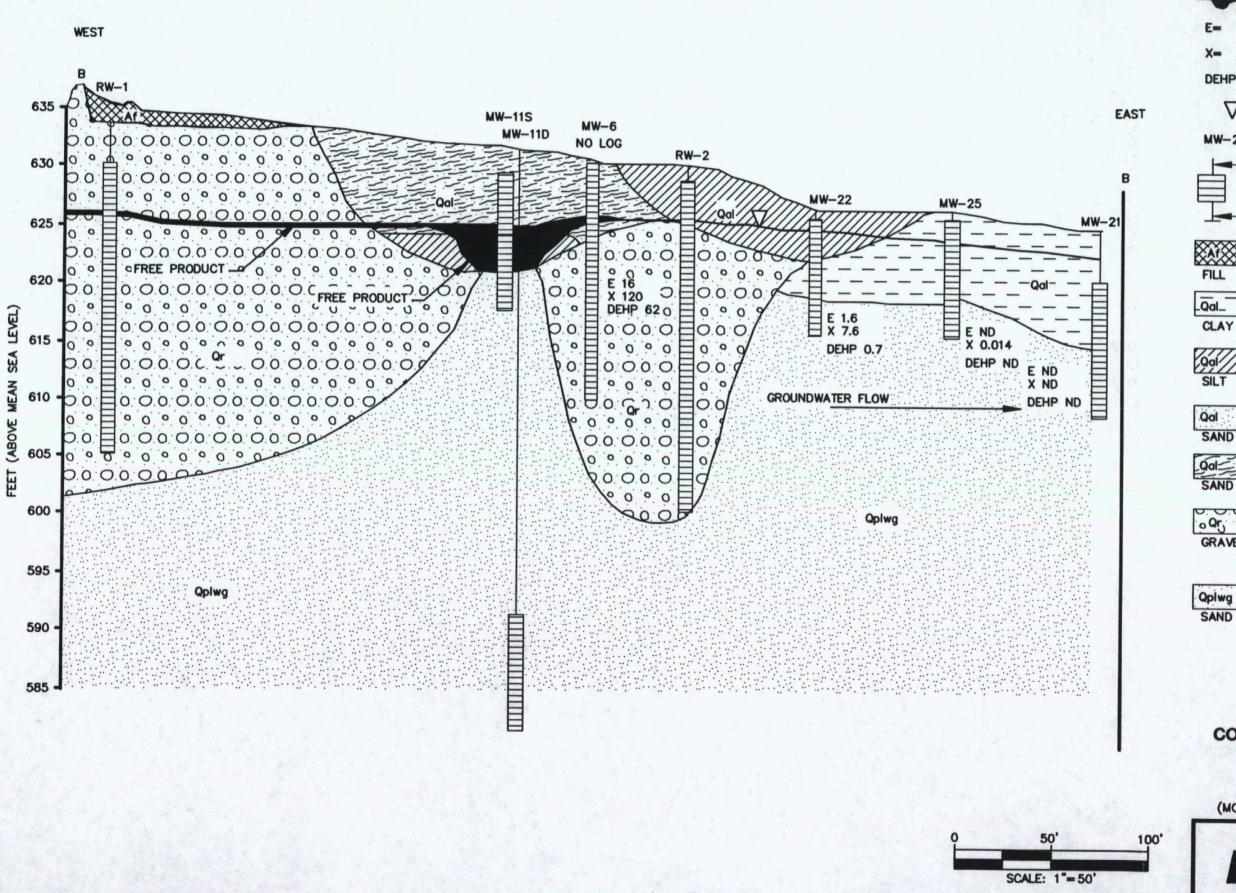


FIGURE 1





LEGEND

FREE PRODUCT LAYER

E= ETHYLBENZENE (mg/L)

X= XYLENE (mg/L)

DEHP - DIETHYL HEXYL PTHALATE (mg/L)

7 WATER TABLE

MW-21 WELL NUMBER

CASED INTERVAL

SCREENED INTERVAL

- BACKFILLED INTERVAL

BROWN, COARSE TO FINE , SOME SILT, LITTLE GRAVEL, SOME ORGANIC MATERIAL

ALLUMUM-GREY CLAY, SOME SILT,
TRACE FINE SAND, SOME ROUNDED
PEBBLES

AY PEDBLES

ALLUVIUM-DARK GREY SILT, SOME CLAY TRACE FINE TO COARSE SAND, SOME FINE TO MEDIUM GRAVEL, FREQUENT BOULDERS

SAND ALUMUM-GREY/BROWN, MEDIUM GRAINED SAND, LITTLE SILT, TRACE

SILTY SAND WITH GRAVEL AND BOULDERS, SOME CLAY.

ROCKAWAY RIVER OUTWASH-GREY
COARSE TO FINE GRAVEL, LITTLE
COARSE TO FINE SAND, ABUNDANT
GRAVEL COBBLES AND BOULDERS

COBBLES AND BOULDERS

STRATIFIED DEPOSITS—GREY/BROWN

GRAVEL, TRACE SILT, OCCASIONAL COBBLES AND BOULDERS.

WATER LEVELS FROM 2ND QUARTER 1999 (RMT, 1999)

CHEMISTRY FROM 2ND QUARTER 1999, EXCEPT MW-6 FROM JAN. 1990 (RMT, 1999; WESTON, 1992)

CONCEPTUAL HYDROGEOLOGIC CROSS SECTION

LE CARPENTER AND CO.

WHARTON, NJ (MODIFIED FROM FIG. 3-6 IN WESTON, 1992)



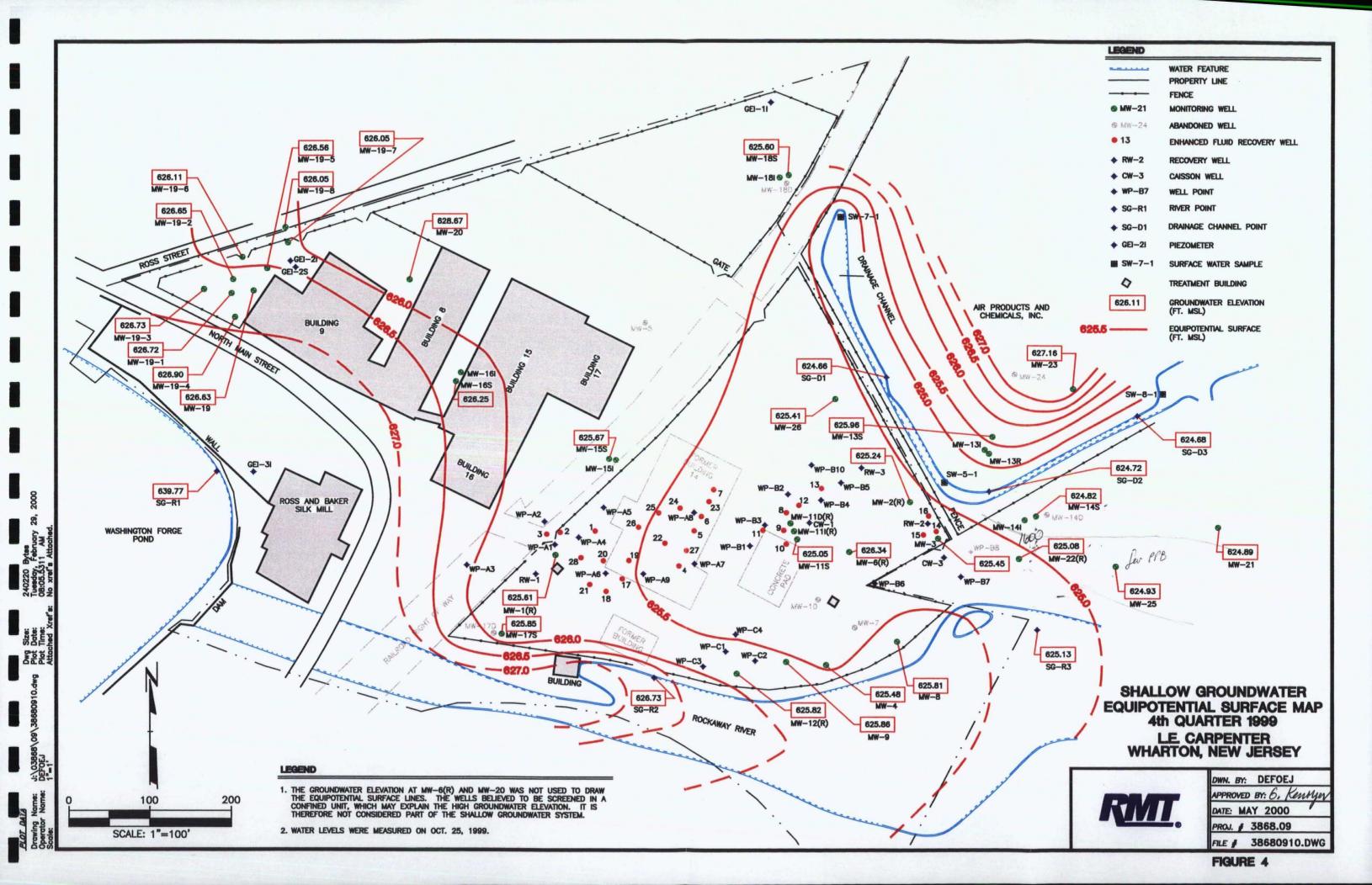
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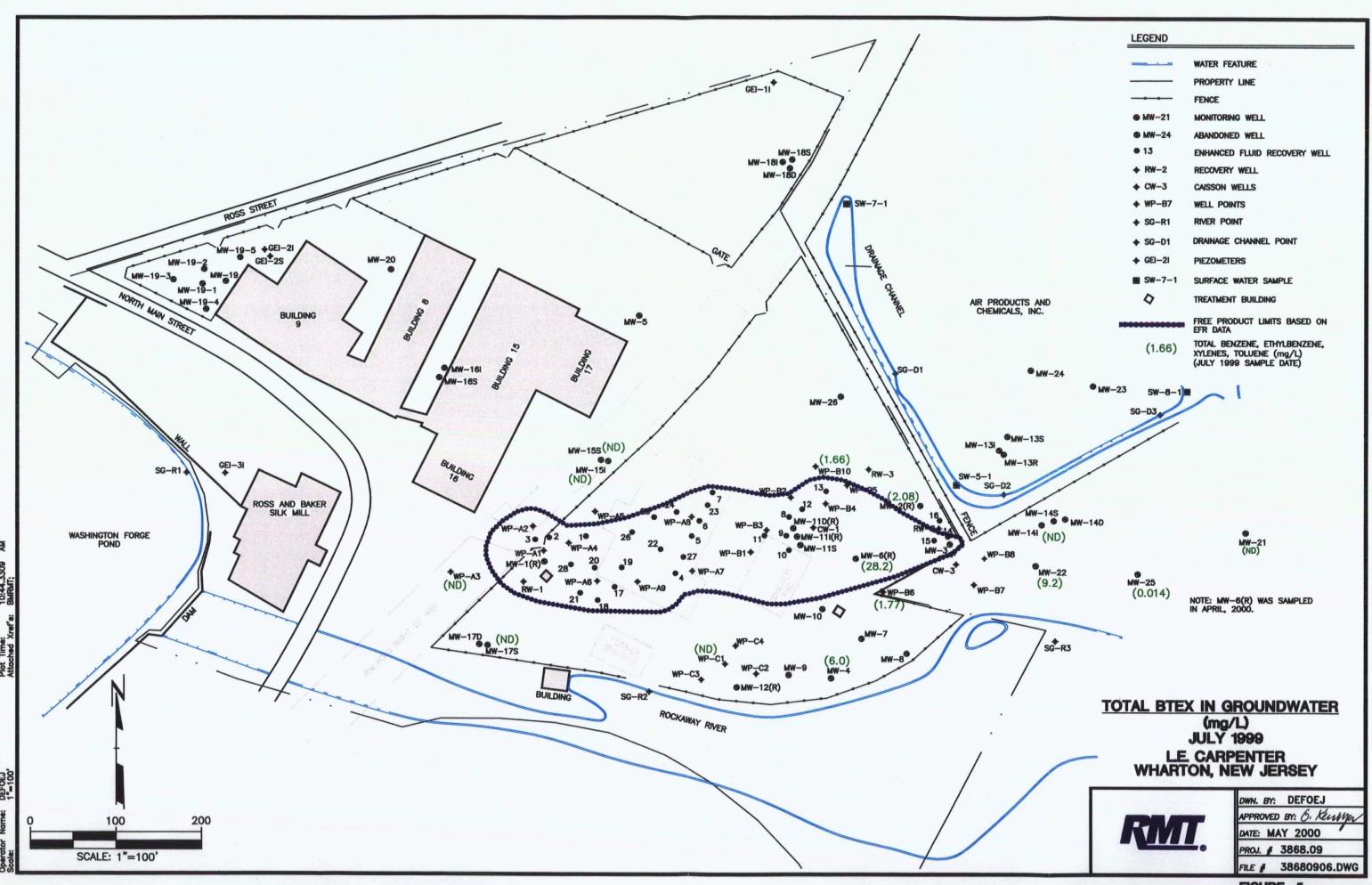
APPROVED BY: 6, KLUMY

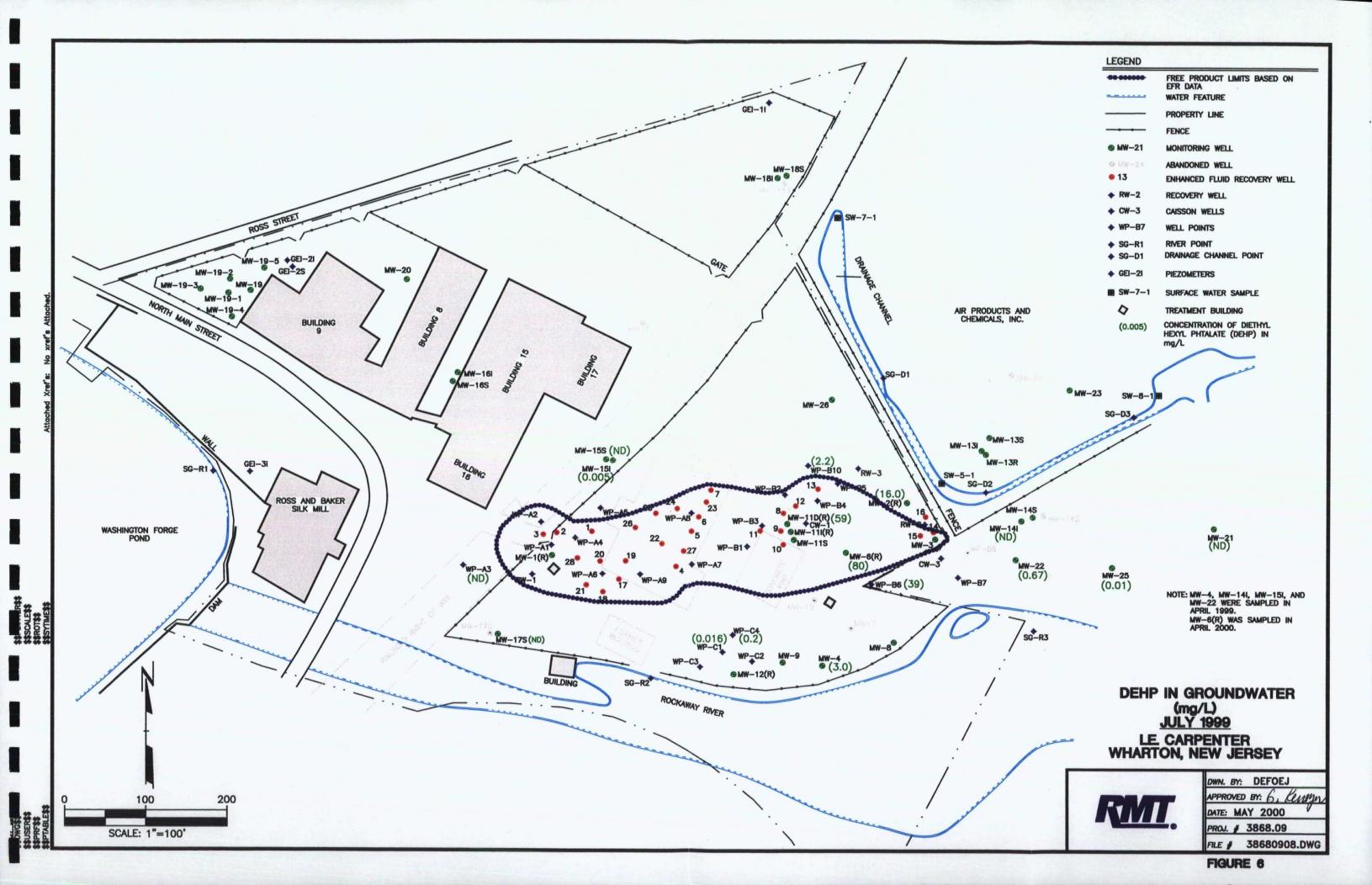
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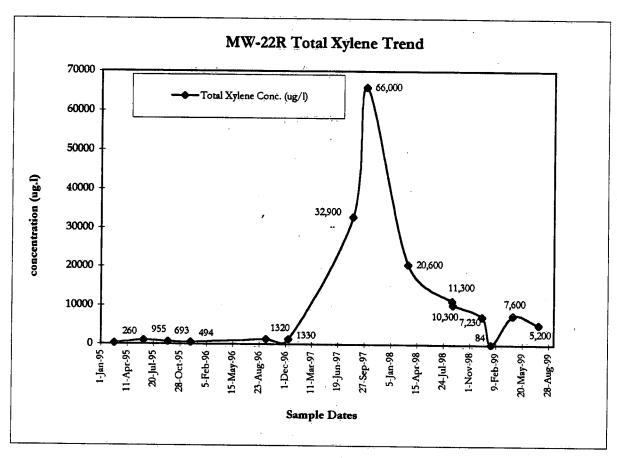
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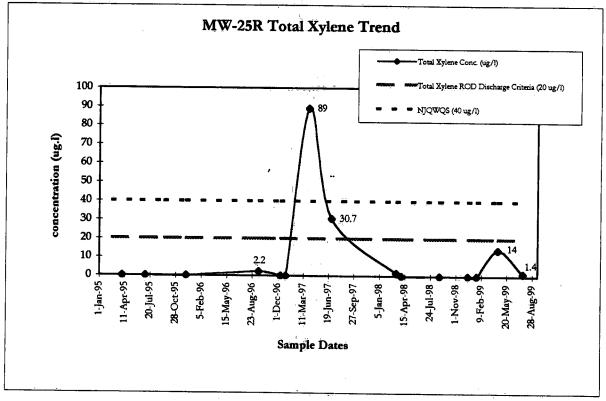
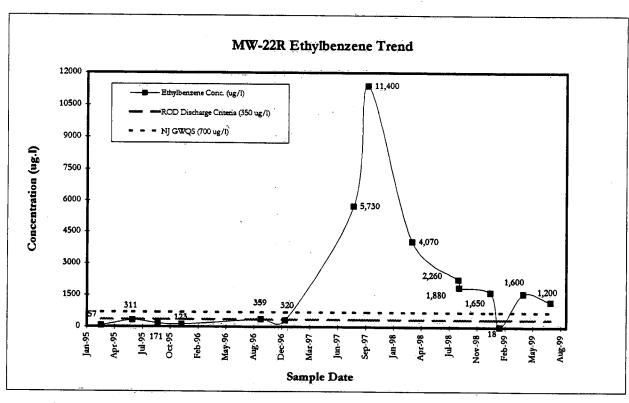


Figure 7a

Concentration of Xylenes in Downgradient Wells Over Time



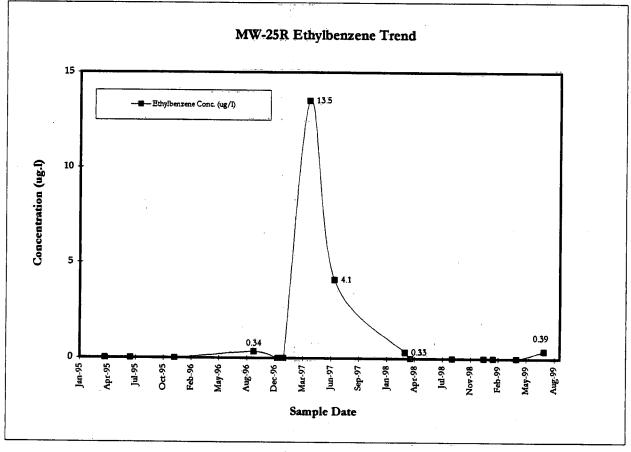
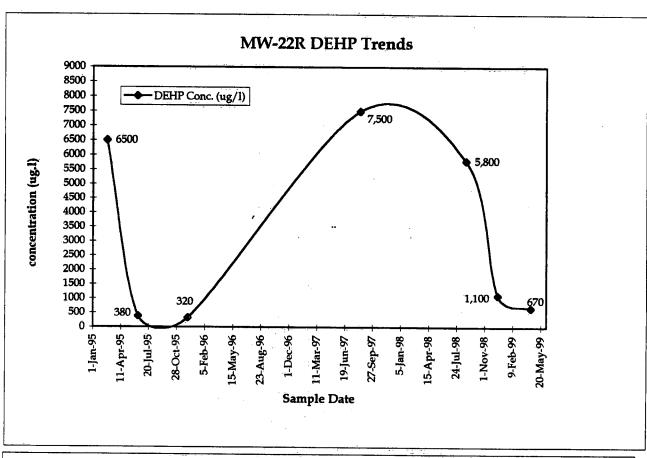


Figure 7b
Concentration of Ethylbenzene in Downgradient Wells Over Time



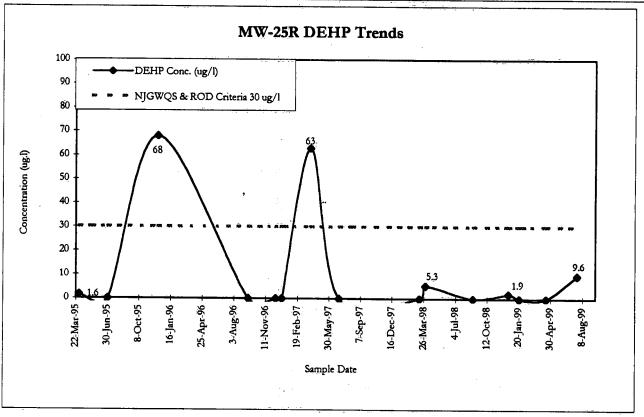
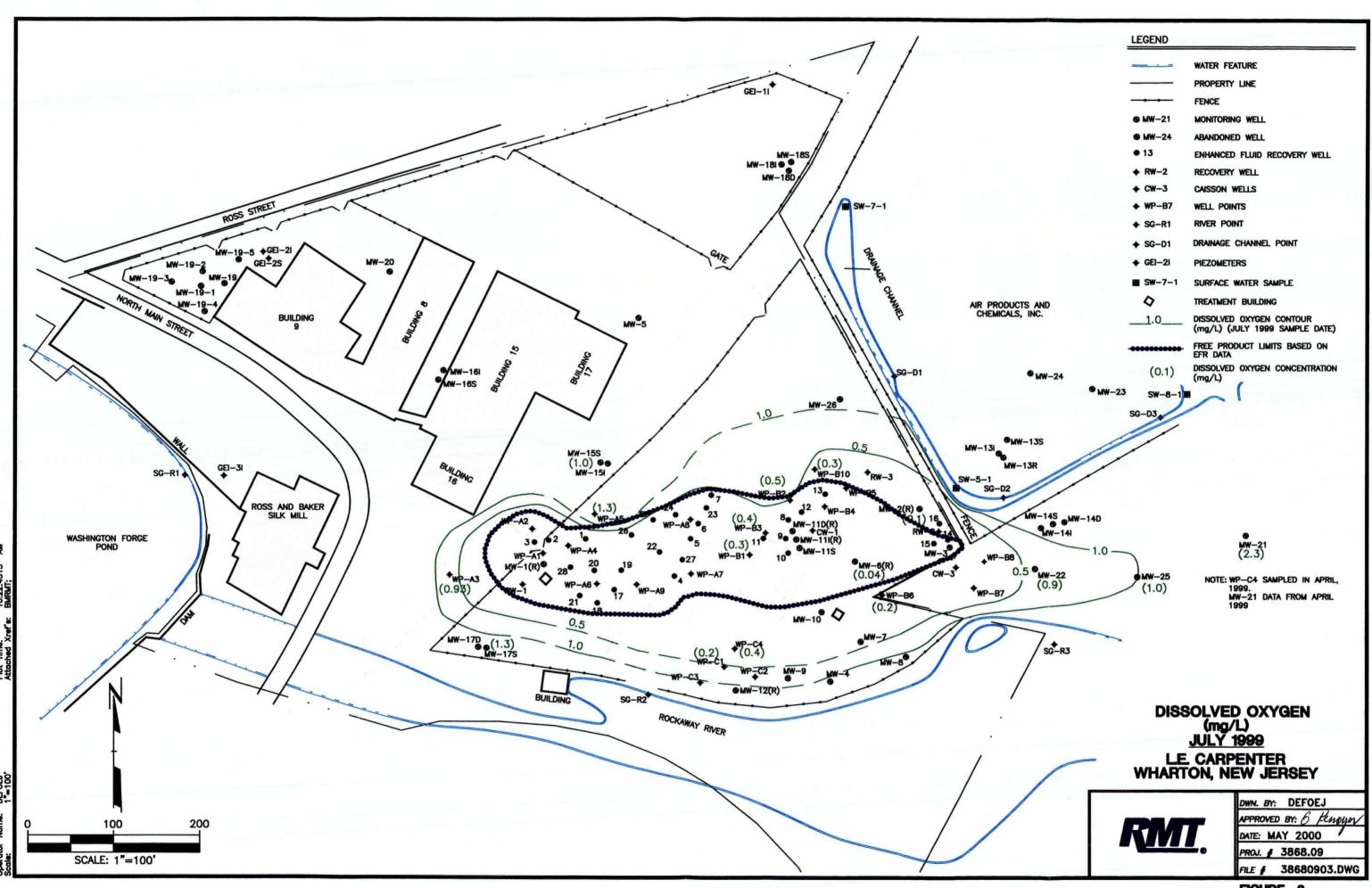
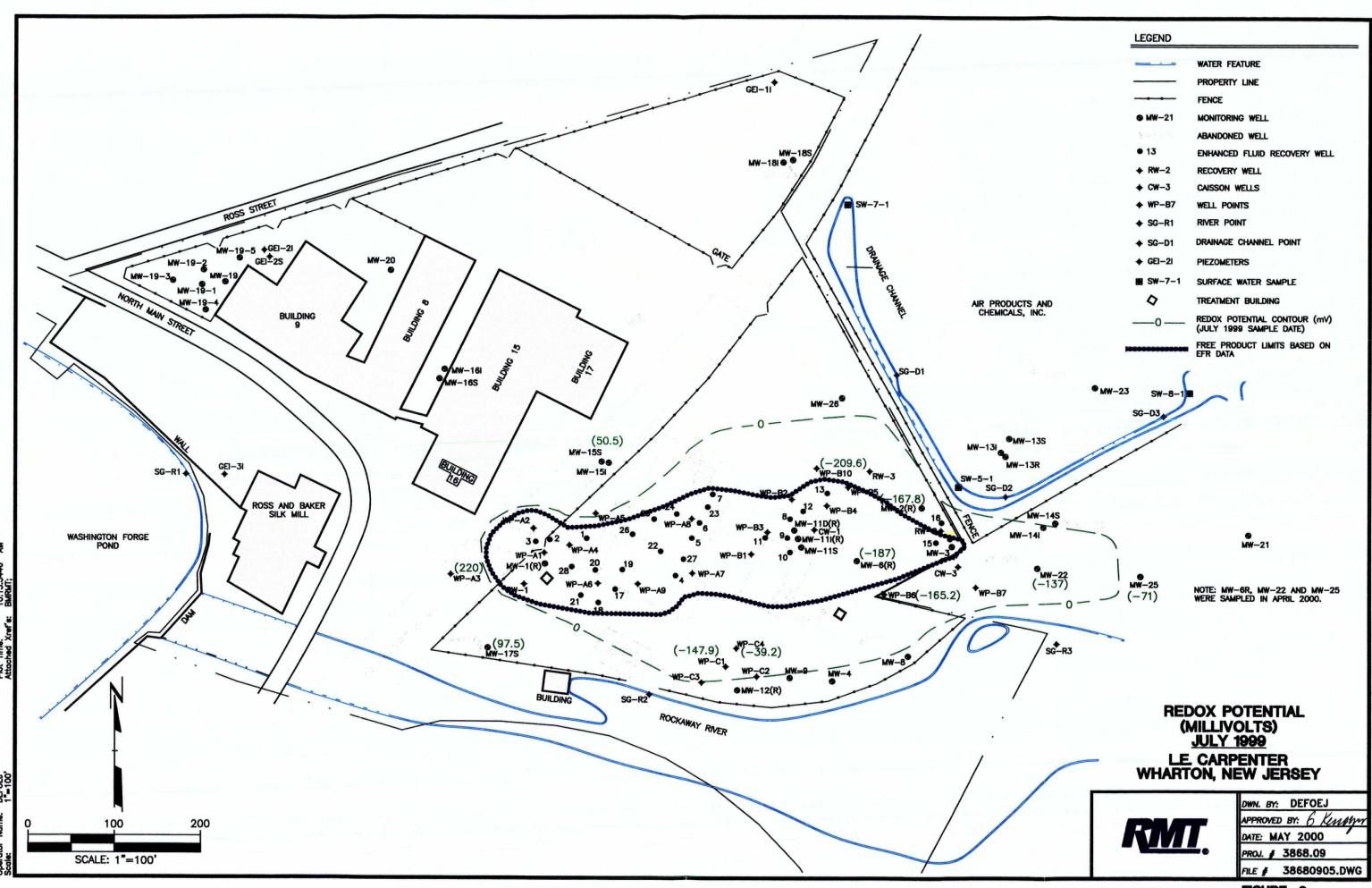
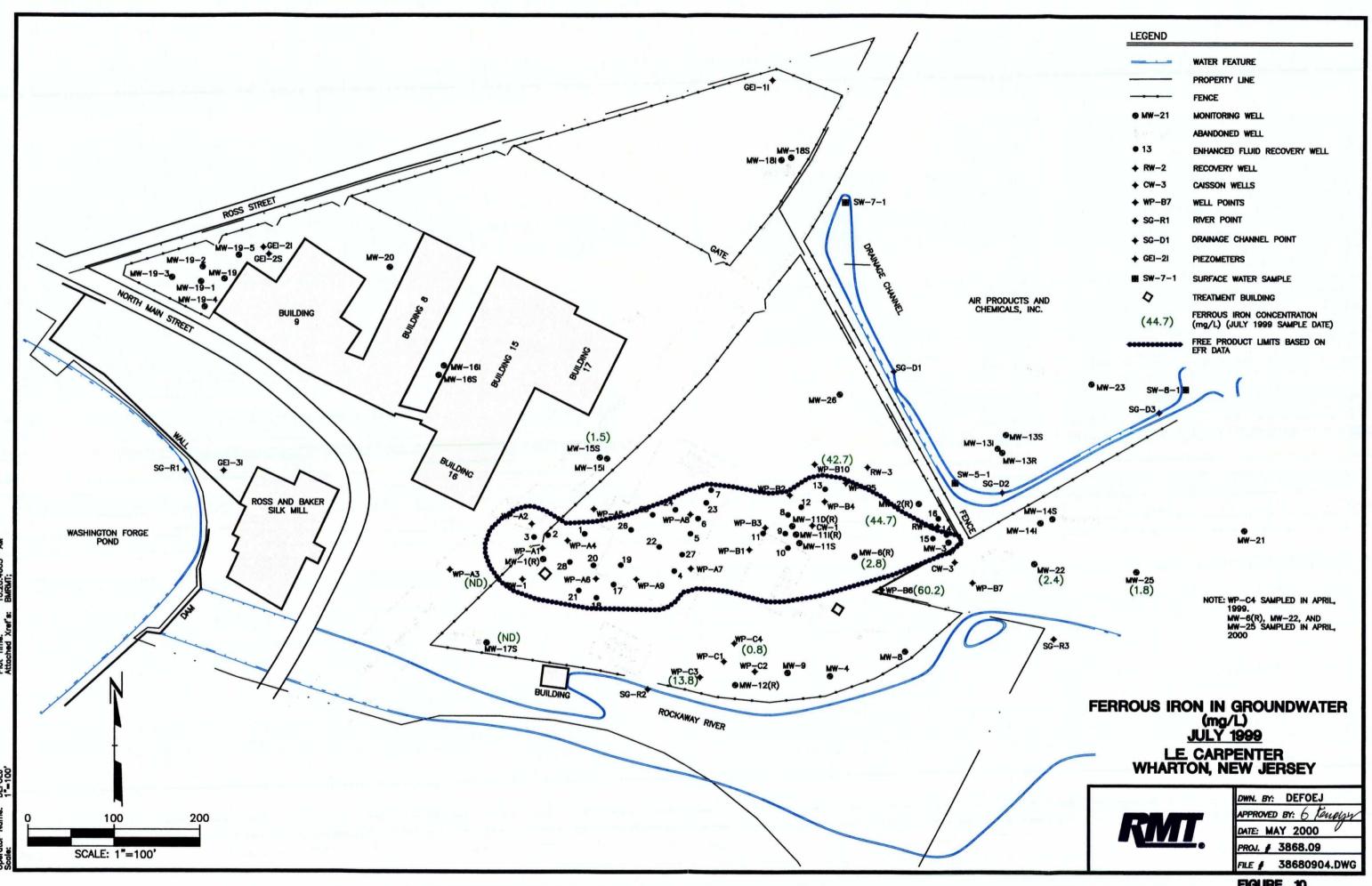
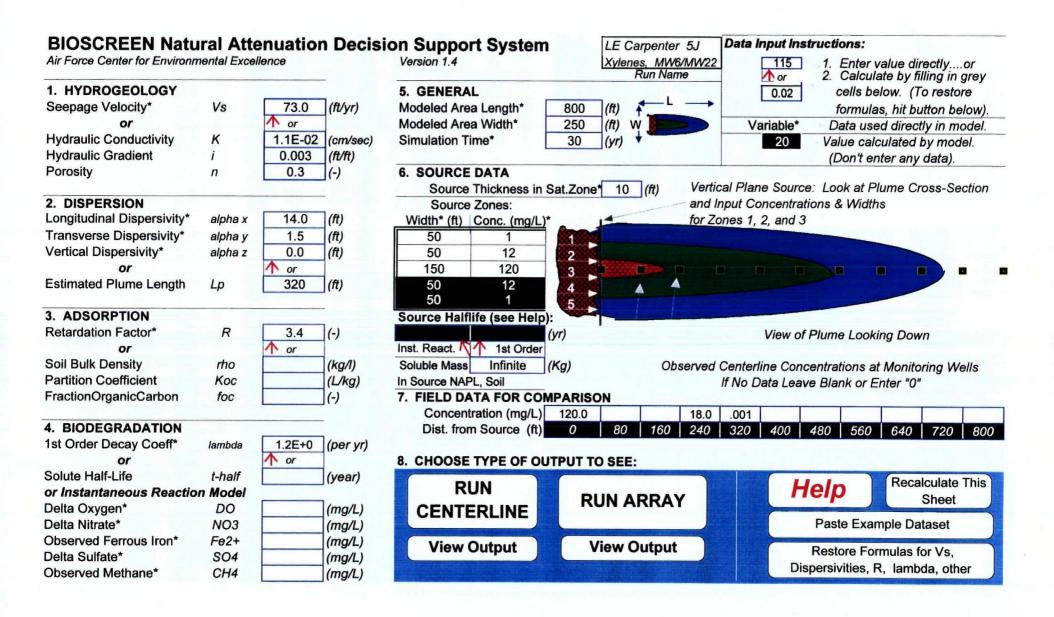


Figure 7c
Concentration of DEHP in Downgradient Wells Over Time

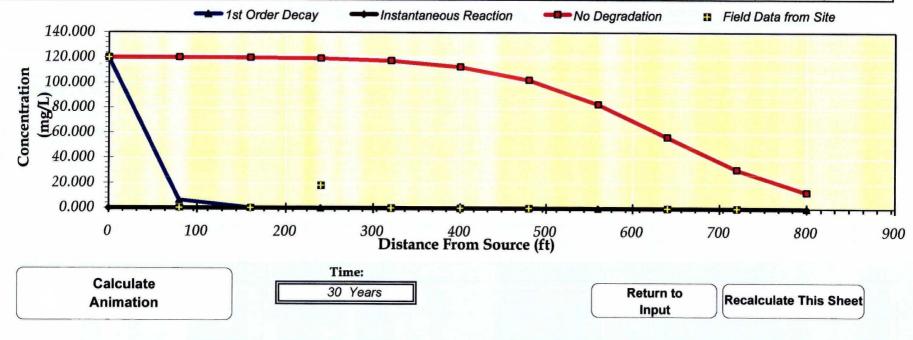








	Distance from Source (ft)										
TYPE OF MODEL	0	80	160	240	320	400	480	560	640	720	800
No Degradation	120.000	119.998	119.914	119.283	117.391	112.683	102.070	82.780	56.632	31.008	13.061
1st Order Decay	120.000	6.286	0.329	0.017	0.001	0.000	0.000	0.000	0.000	0.000	0.000
Inst. Reaction	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!
Field Data from Site	120.0			18.000	0.001						



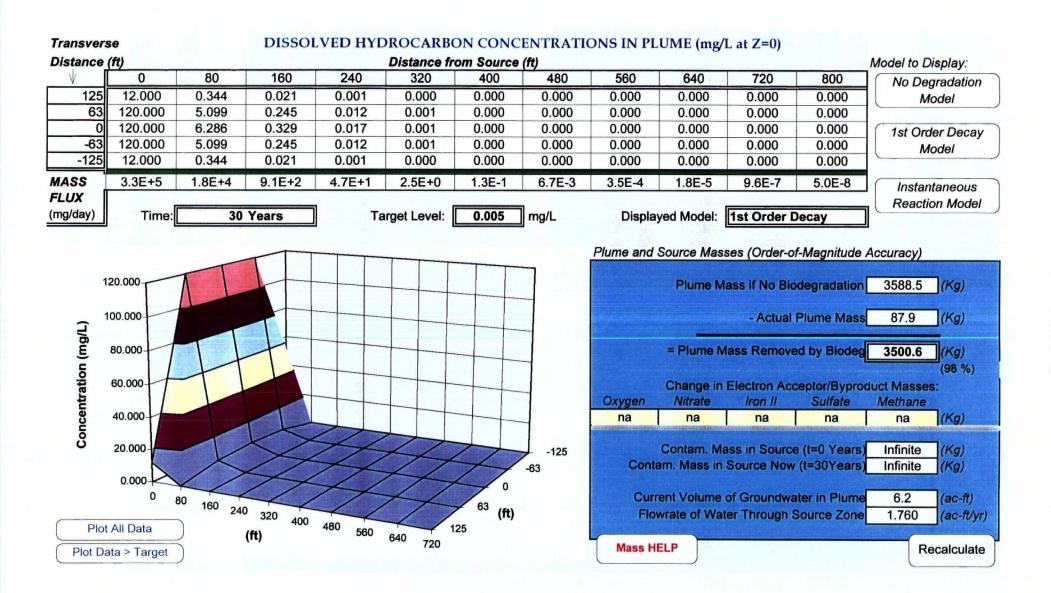
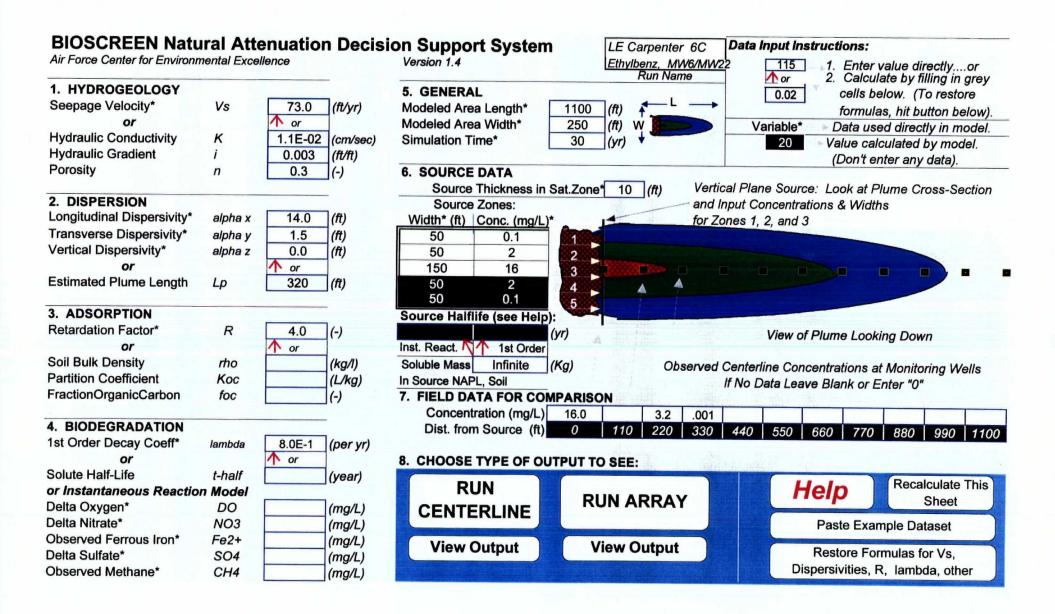


Figure 11c – Xylenes Simulation, Parameter Set 1, Plume Output



	Distance from Source (ft)										
TYPE OF MODEL	0	110	220	330	440	550	660	770	880	990	1100
No Degradation	16.000	15.996	15.886	15.138	12.476	7.423	2.673	0.518	0.051	0.002	0.000
1st Order Decay	16.000	0.548	0.019	0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Inst. Reaction	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!
Field Data from Site	16.0		3.200	0.001							

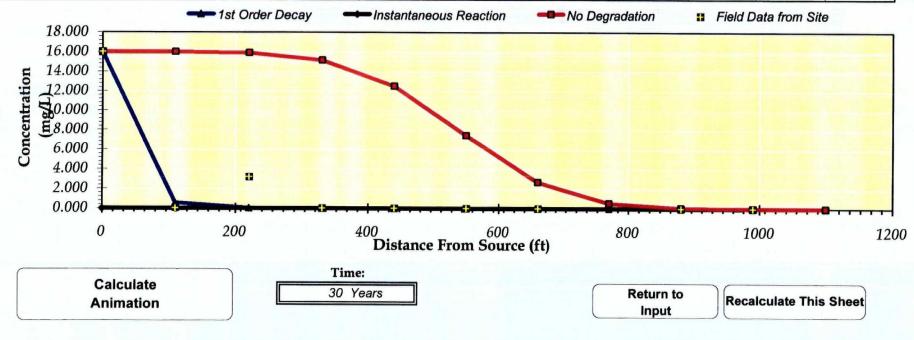


Figure 12b – Ethylbenzene Simulation, Parameter Set 1, Centerline Output

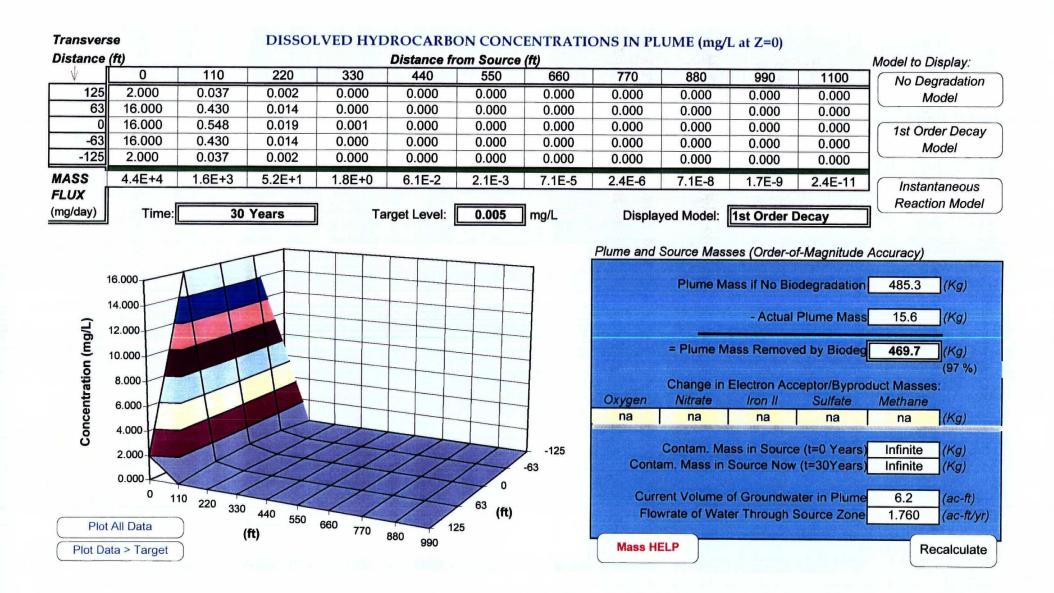
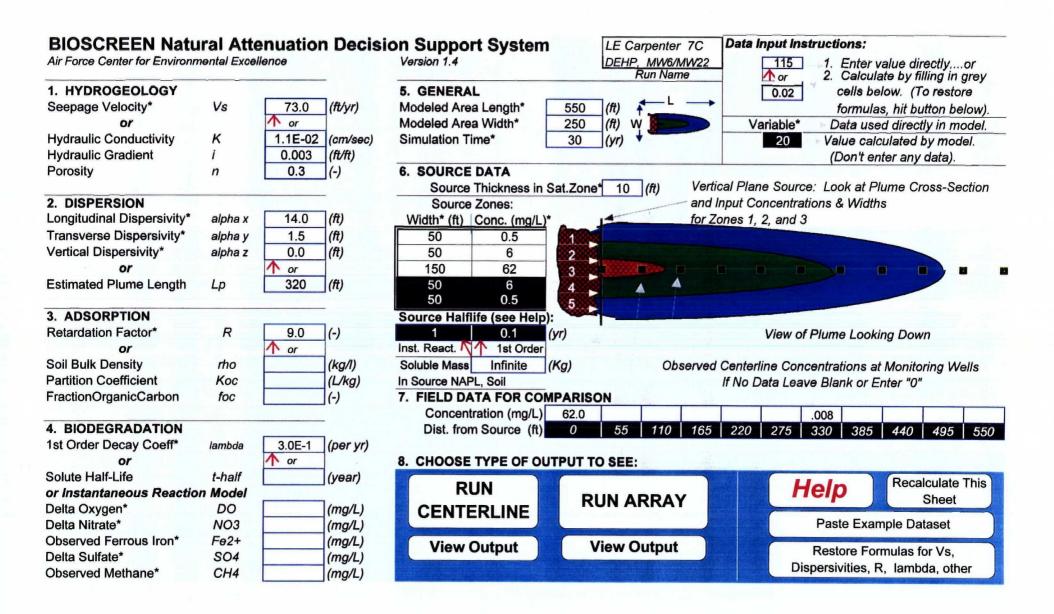


Figure 12c – Ethylbenzene Simulation, Parameter Set 1, Plume Output



Distance from Source (ft) **TYPE OF MODEL** 0 55 110 165 220 275 330 385 440 495 550 No Degradation 62.000 61.302 37.780 58.705 51.344 21.561 8.965 0.514 2.604 0.006 0.068 1st Order Decay 14.140 3.225 0.734 62.000 0.166 0.037 0.008 0.001 0.000 0.000 0.000 Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 62.0 0.008

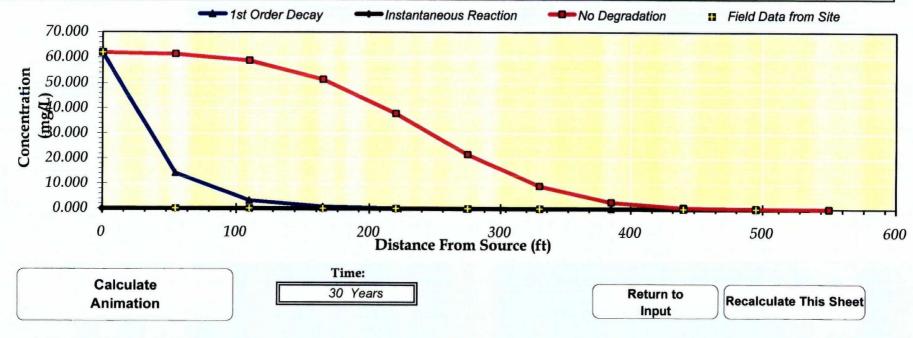
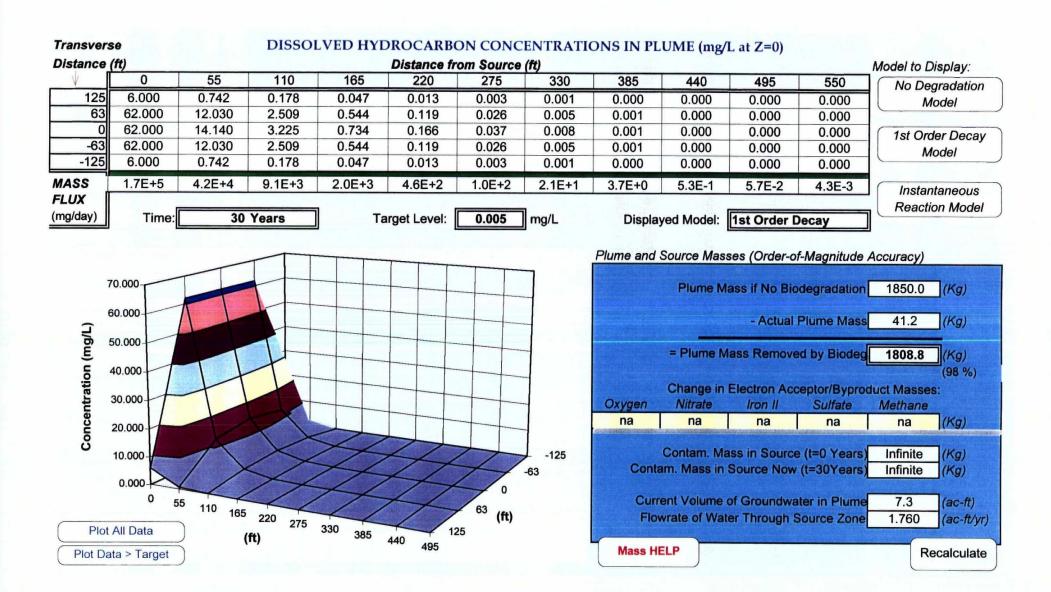
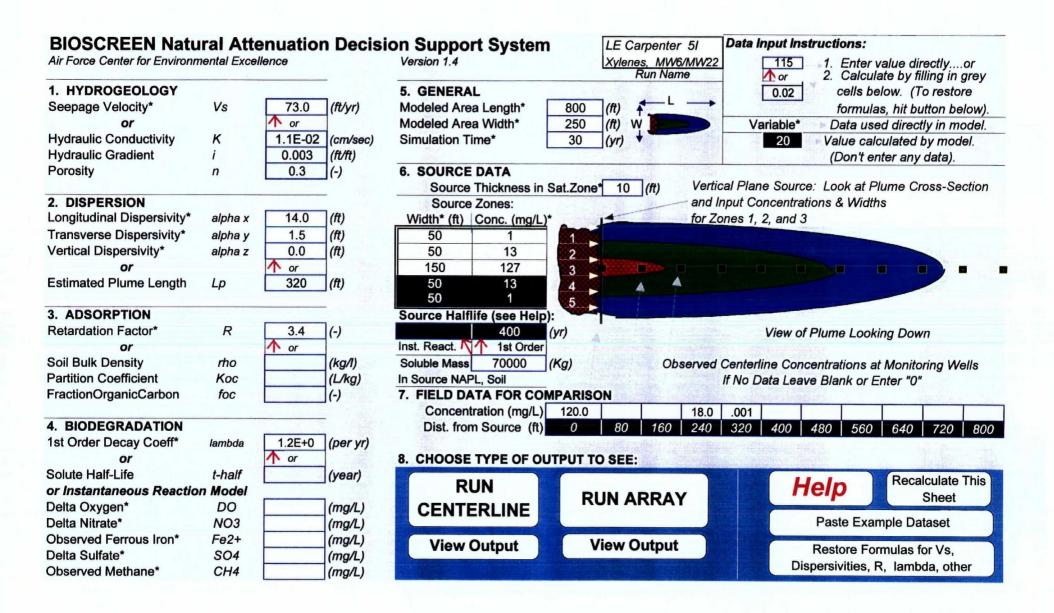
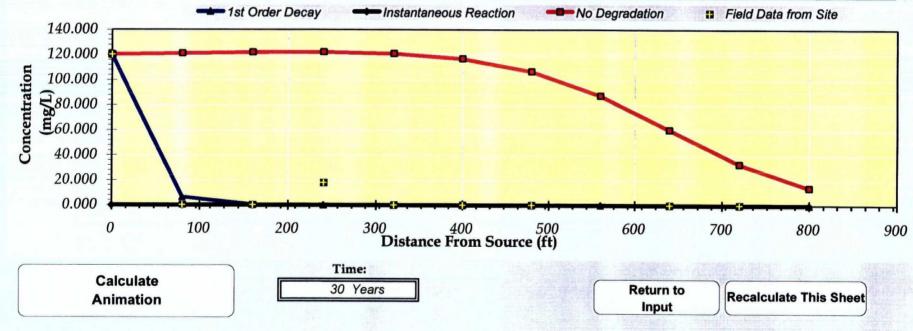


Figure 13b – DEHP Simulation, Parameter Set 1, Centerline Output





Distance from Source (ft) TYPE OF MODEL 0 80 160 240 320 480 400 560 640 720 800 No Degradation 120.286 121.098 121.833 122.013 120.894 116.834 106.551 87.004 59.928 32.825 13.827 1st Order Decay 6.344 120.286 0.334 0.018 0.001 0.000 0.000 0.000 0.000 0.000 0.000 **#VALUE!** Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 120.0 18.000 0.001



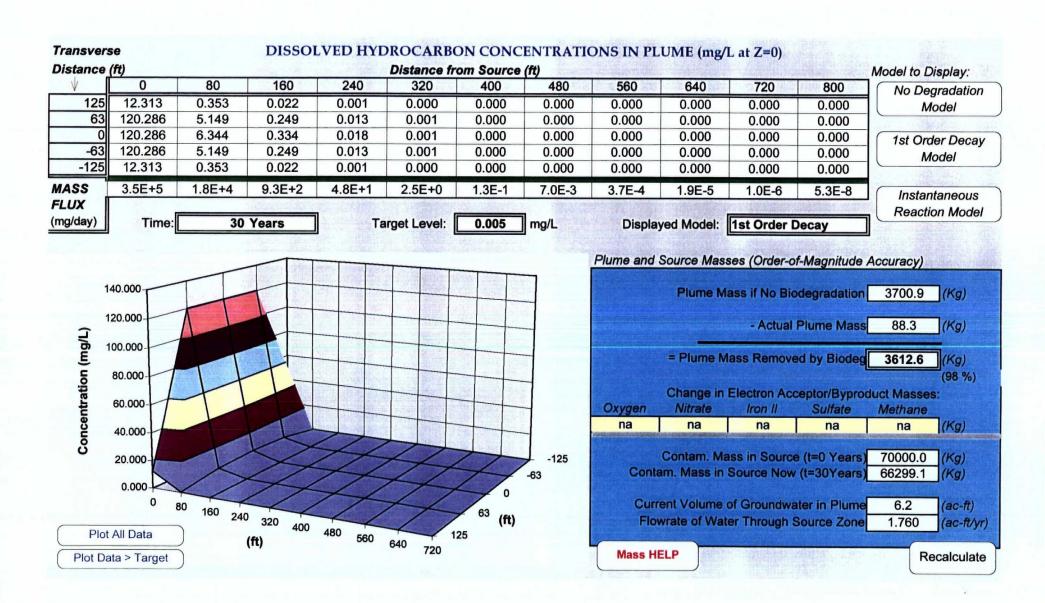


Figure 14c – Xylenes Simulation, Parameter Set 2, Plume Output

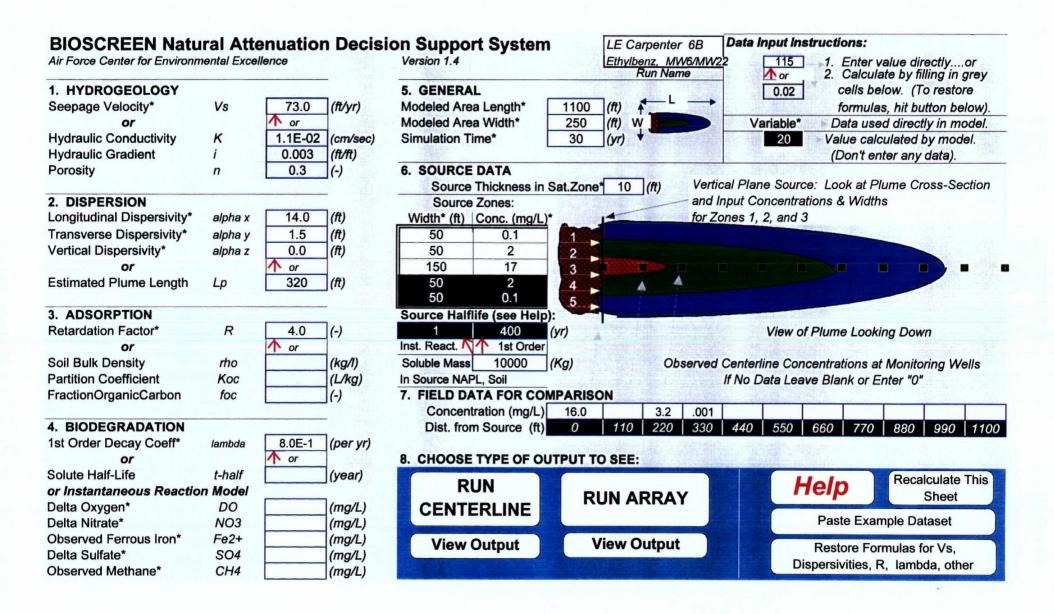


Figure 15a - Ethylbenzene Simulation, Parameter Set 2, Bioscreen Input

Distance from Source (ft) **TYPE OF MODEL** 0 110 220 330 440 550 660 770 880 990 1100 No Degradation 16.150 16.313 16.368 15.757 13.119 7.883 2.838 0.054 0.003 0.550 0.000 1st Order Decay 16.150 0.559 0.019 0.001 0.000 0.000 0.000 0.000 0.000 0.000 0.000 **#VALUE! #VALUE! #VALUE! #VALUE!** Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 16.0 3.200 0.001

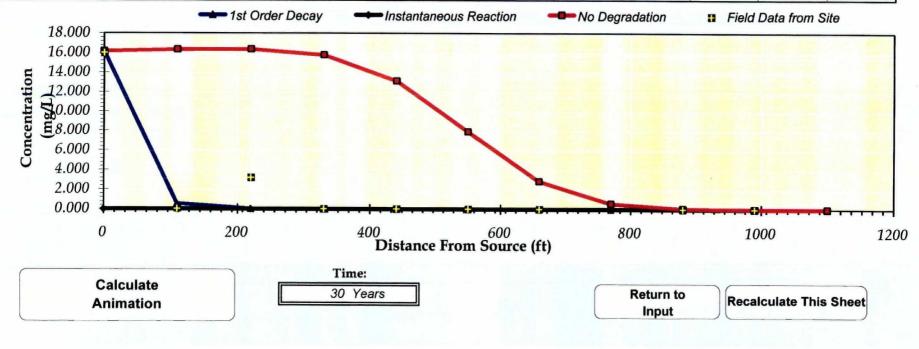


Figure 15b - Ethylbenzene Simulation, Parameter Set 2, Centerline Output

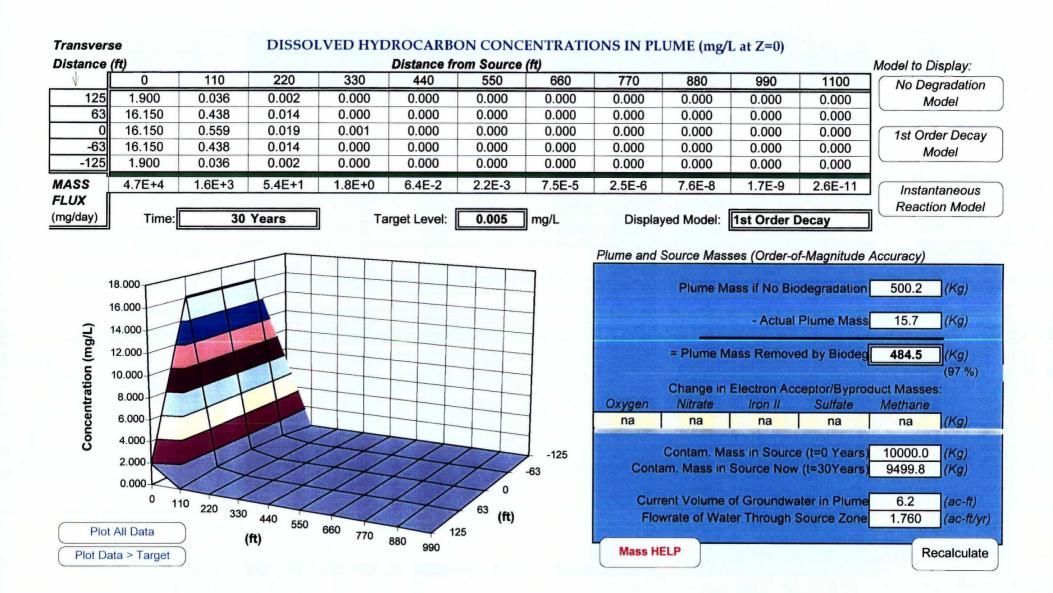


Figure 15c - Ethylbenzene Simulation, Parameter Set 2, Plume Output

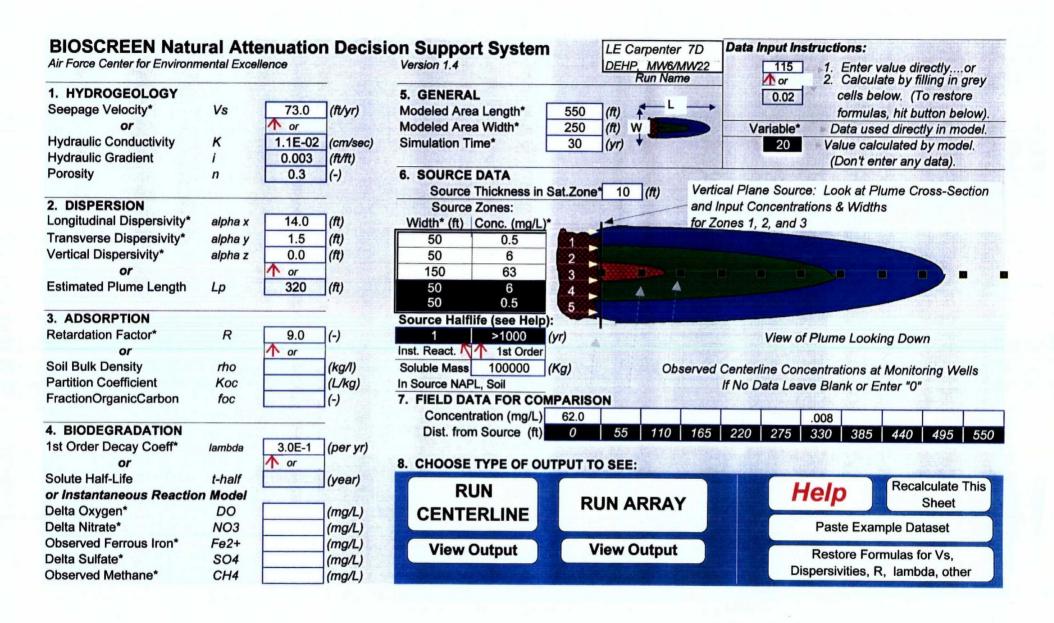


Figure 16a - DEHP Simulation, Parameter Set 2, Bioscreen Input

Distance from Source (ft) TYPE OF MODEL 55 110 0 165 220 275 330 385 440 495 550 No Degradation 61.828 61.392 59.041 51.858 38.320 21.909 9.109 2.645 0.522 0.069 0.006 1st Order Decay 61.828 14.161 3.243 0.742 0.037 0.008 0.001 0.169 0.000 0.000 0.000 Inst. Reaction #VALUE! **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 62.0 0.008

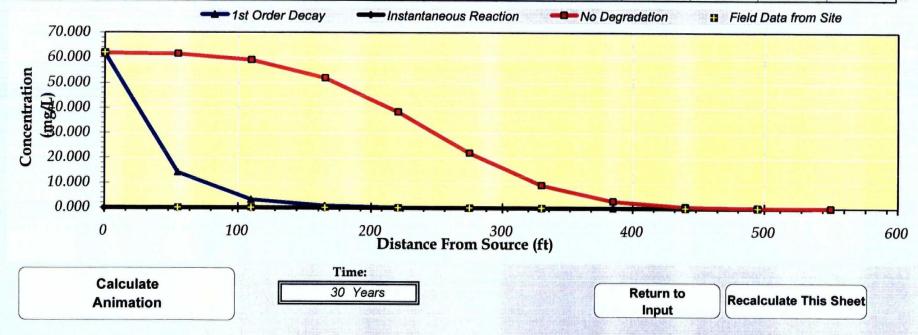


Figure 16b - DEHP Simulation, Parameter Set 2, Centerline Output

18

-125

63

125

495

(ft)

20.000

10.000-

0.000-0

Plot All Data

Plot Data > Target

55

110

165

220

(ft)

275

330

385

440

Figure 16c - DEHP Simulation, Parameter Set 2, Plume Output

100000.0 (Kg)

(Kg)

(ac-ft)

Recalculate

(ac-ft/yr)

98139.6

7.3

1.760

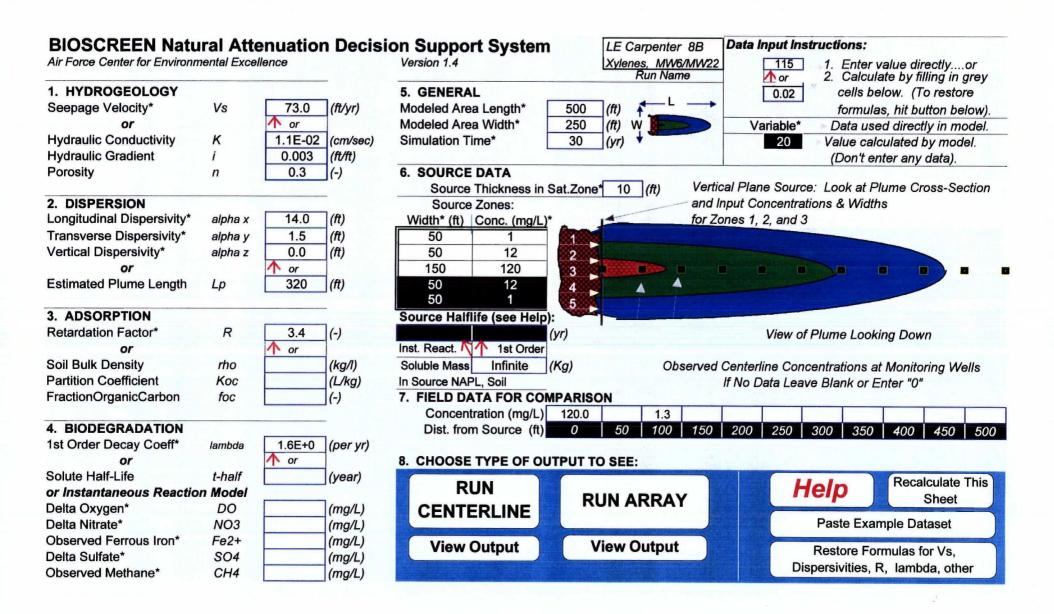
Contam. Mass in Source (t=0 Years)

Contam. Mass in Source Now (t=30Years)

Current Volume of Groundwater in Plume

Flowrate of Water Through Source Zone

Mass HELP



Distance from Source (ft) **TYPE OF MODEL** 0 50 100 150 200 250 300 350 400 450 500 120.000 119.999 119.995 119.942 No Degradation 119.706 119.134 118.042 116.091 112.683 106.971 98.100 **1st Order Decay** 120.000 12.326 1.266 0.130 0.013 0.001 0.000 0.000 0.000 0.000 0.000 **#VALUE! #VALUE! #VALUE! #VALUE!** Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 120.0 1.300

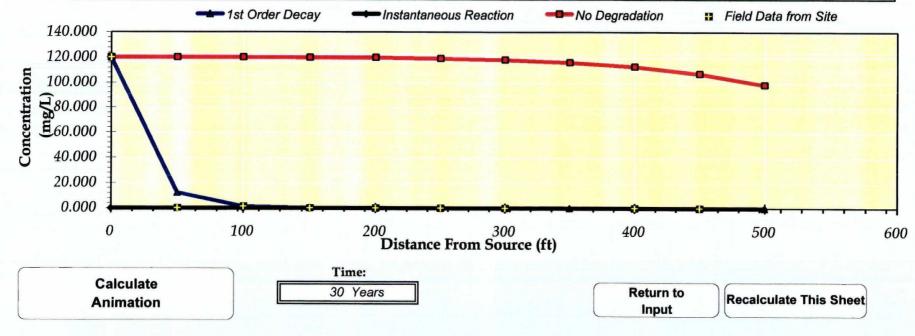


Figure 17b - Xylenes Simulation, Alternate Flowline, Centerline Output

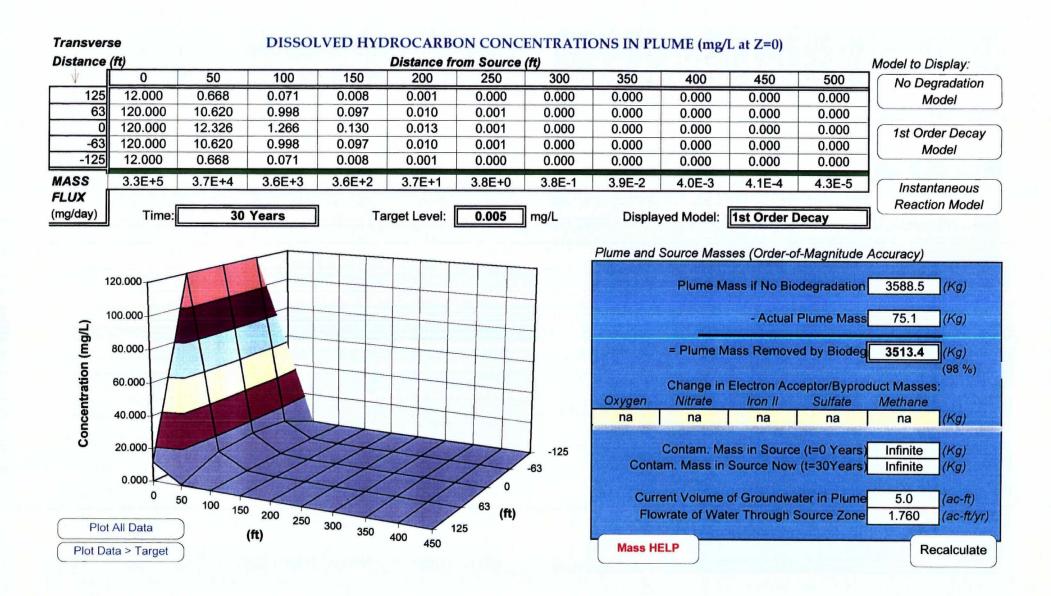


Figure 17c - Xylenes Simulation, Alternate Flowline, Plume Output

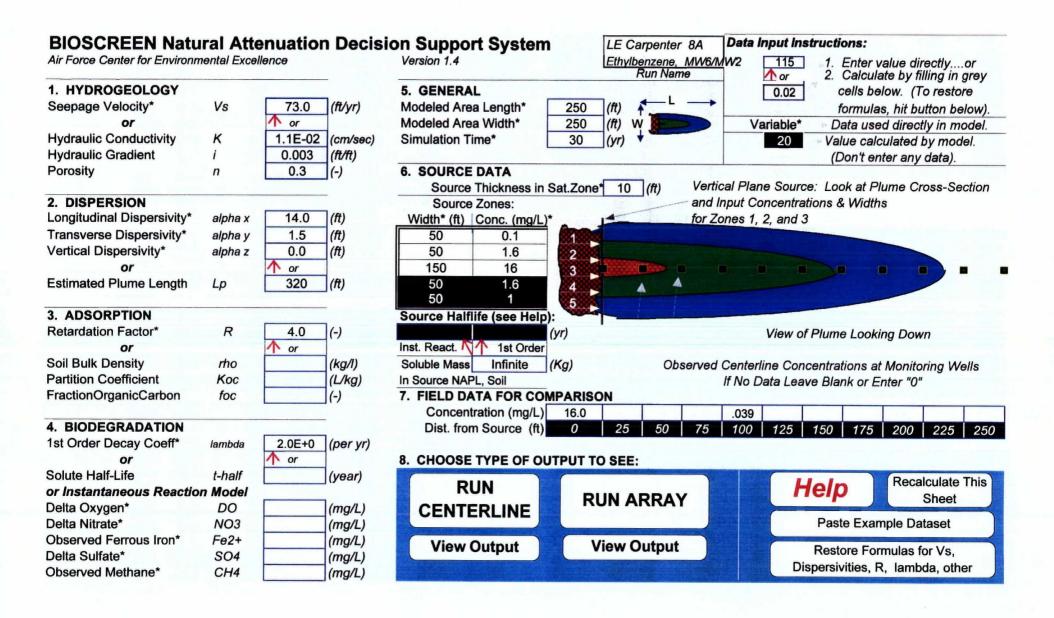
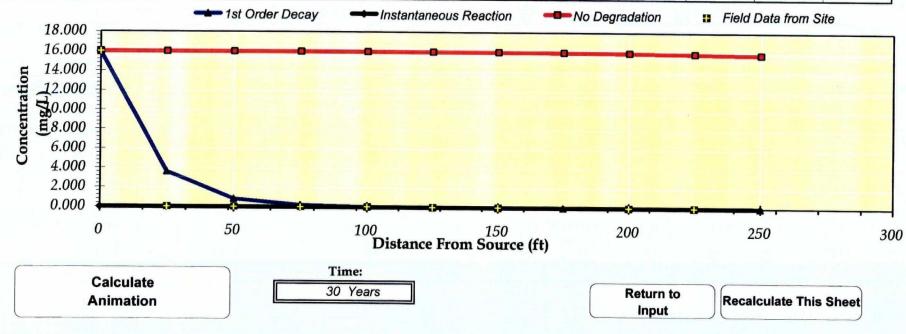


Figure 18a - Ethylbenzene Simulation, Alternate Flowline, Bioscreen Input

	Distance from Source (ft)										
TYPE OF MODEL	0	25	50	75	100	125	150	175	200	225	250
No Degradation	16.000	16.000	16.000	15.999	15.997	15.993	15.984	15.964	15.928	15.871	15.782
1st Order Decay	16.000	3.597	0.809	0.182	0.041	0.009	0.002	0.000	0.000	0.000	0.000
Inst. Reaction	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!
Field Data from Site	16.0				0.039						



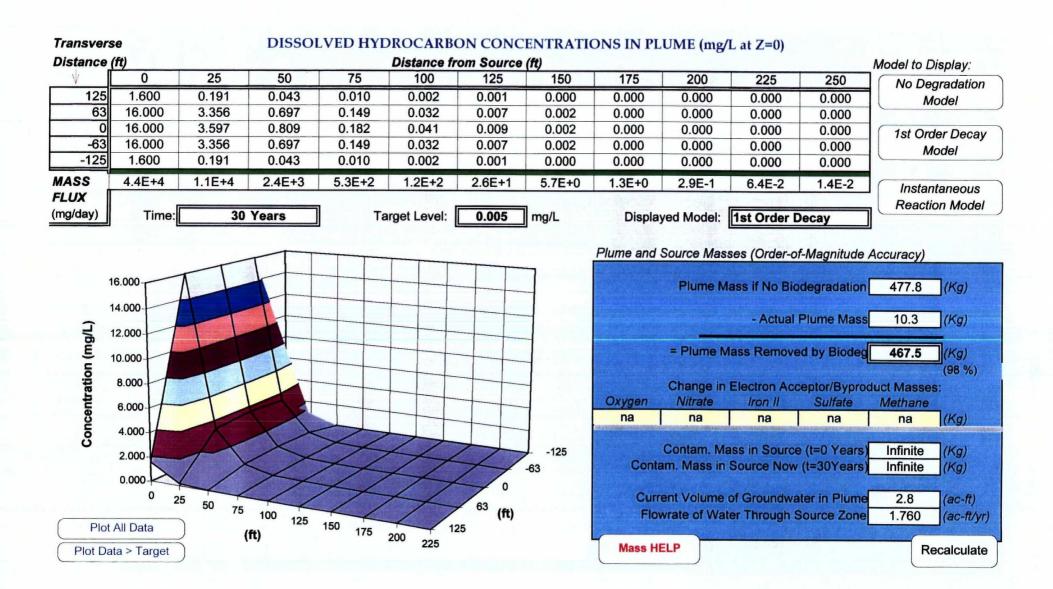


Figure 18c - Ethylbenzene Simulation, Alternate Flowline, Plume Output

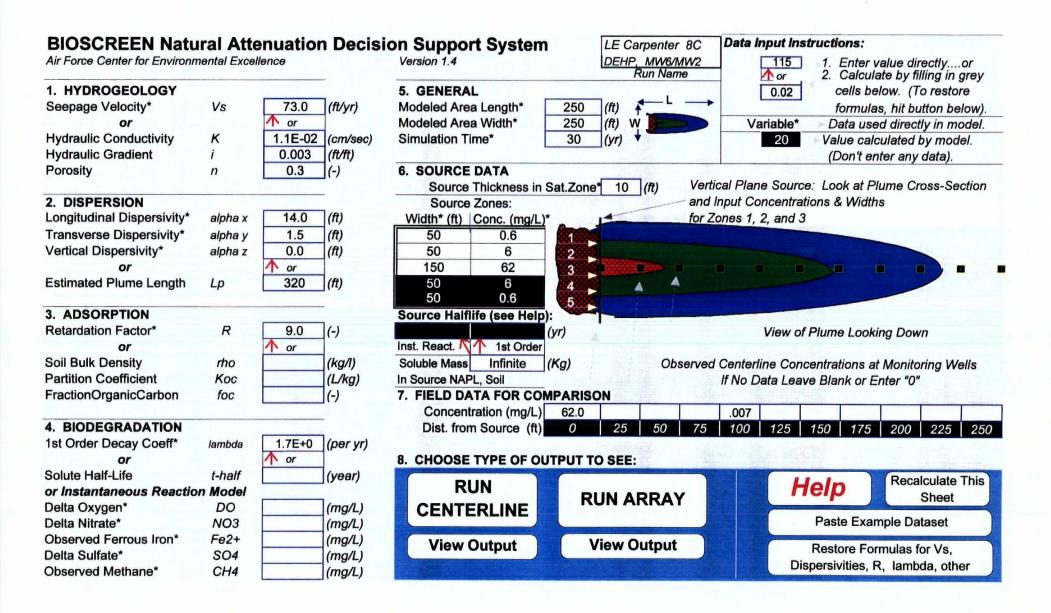
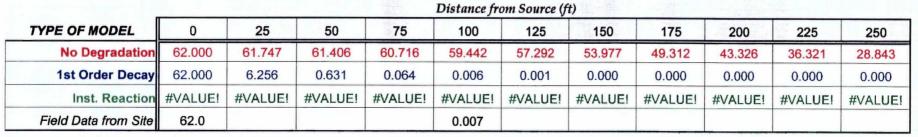


Figure 19a - DEHP Simulation Alternate Flowline, Bioscreen Input



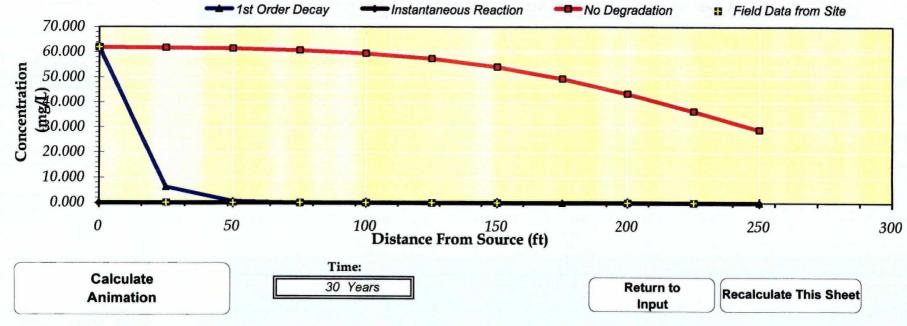


Figure 19b - DEHP Simulation Alternate Flowline, Centerline Output

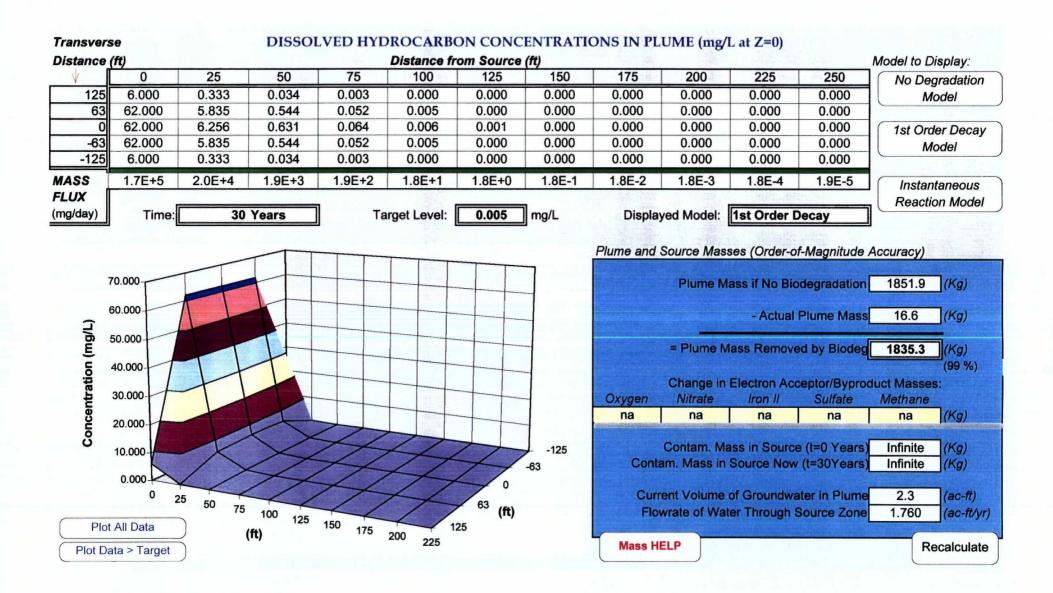
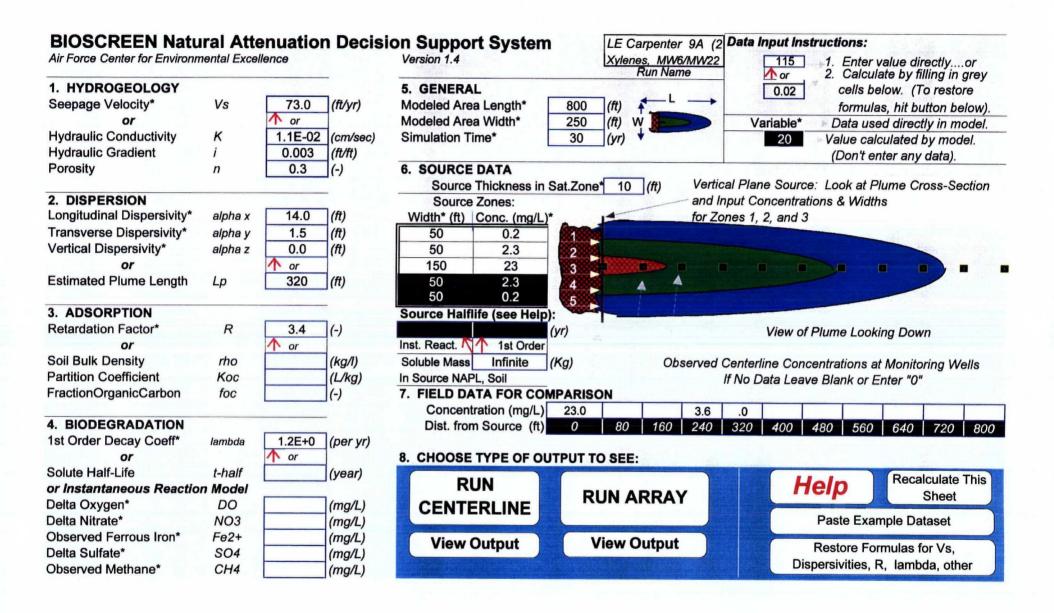
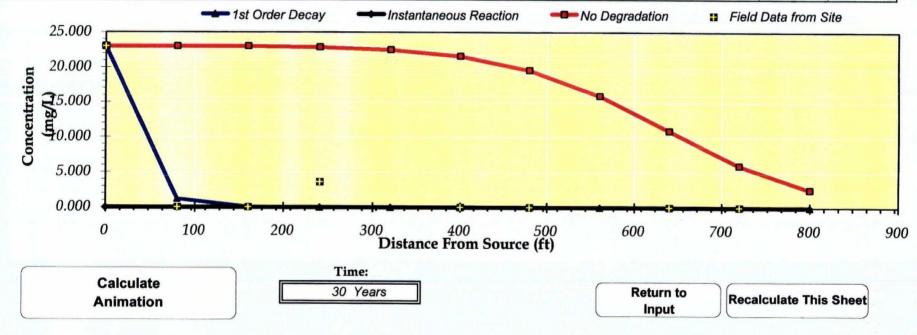


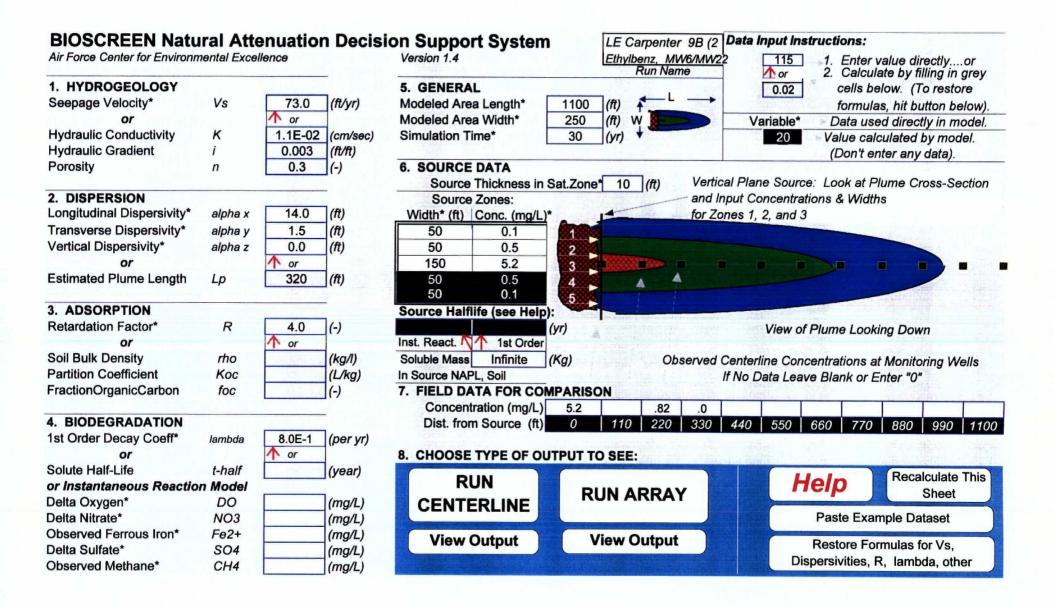
Figure 19c - DEHP Simulation Alternate Flowline, Plume Output



	Distance from Source (ft)										
TYPE OF MODEL	0	80	160	240	320	400	480	560	640	720	800
No Degradation	23.000	23.000	22.984	22.863	22.500	21.598	19.563	15.866	10.855	5.943	2.503
1st Order Decay	23.000	1.205	0.063	0.003	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Inst. Reaction	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!	#VALUE!
Field Data from Site	23.0			3.600	0.000						



DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) **Transverse** Distance (ft) Distance from Source (ft) Model to Display: 0 80 160 240 480 320 560 640 400 720 800 No Degradation 125 2.300 0.066 0.004 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 Model 63 23.000 0.977 0.047 0.002 0.000 0.000 0.000 0.000 0.000 0.000 0.000 23.000 1.205 0.063 0.003 0.000 0.000 0.000 0.000 0.000 0.000 0.000 1st Order Decay -63 23.000 0.977 0.002 0.000 0.047 0.000 0.000 0.000 0.000 0.000 0.000 Model -125 2.300 0.066 0.004 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 3.5E+3 MASS 6.3E+4 1.8E+2 9.1E+0 4.7E-1 2.5E-2 1.3E-3 6.7E-5 3.5E-6 1.8E-7 9.6E-9 Instantaneous **FLUX** Reaction Model (mg/day) Time: 30 Years 0.005 Target Level: mg/L Displayed Model: 1st Order Decay Plume and Source Masses (Order-of-Magnitude Accuracy) 25.000 Plume Mass if No Biodegradation 687.9 (Kg) - Actual Plume Mass 16.8 (Kg) 20.000 Concentration (mg/L) = Plume Mass Removed by Bioded 671.1 (Kg) 15.000 (98 %) Change in Electron Acceptor/Byproduct Masses: 10.000 Oxygen Nitrate Iron II Sulfate Methane na (Kg) na na na na 5.000 Contam. Mass in Source (t=0 Years) (Kg) -125 Infinite Contam. Mass in Source Now (t=30Years) Infinite (Kg) 0.000 0 Current Volume of Groundwater in Plume 4.5 80 (ac-ft) 160 240 (ft) Flowrate of Water Through Source Zone 1.760 320 (ac-ft/yr) 400 480 Plot All Data 125 560 (ft) 640 720 Mass HELP Recalculate Plot Data > Target



Distance from Source (ft) TYPE OF MODEL 0 110 220 330 440 550 660 770 880 990 1100 No Degradation 5.200 5.199 5.162 4.917 4.050 2.408 0.866 0.168 0.016 0.001 0.000 **1st Order Decay** 5.200 0.178 0.006 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 Inst. Reaction #VALUE! **#VALUE! #VALUE! #VALUE!** #VALUE! #VALUE! **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** 5.2 0.820 0.000 Field Data from Site

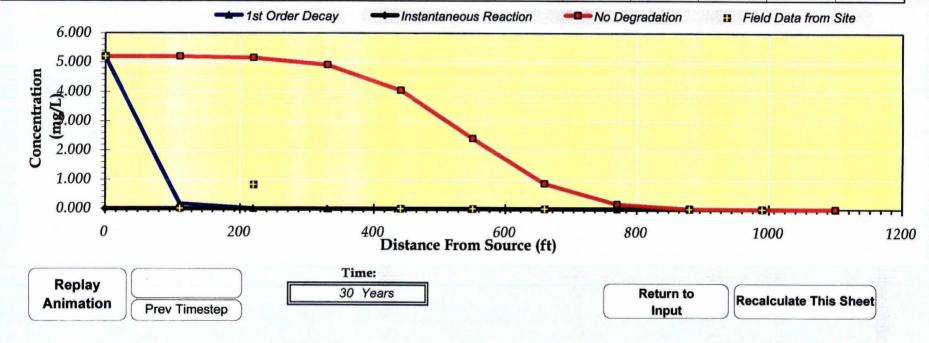
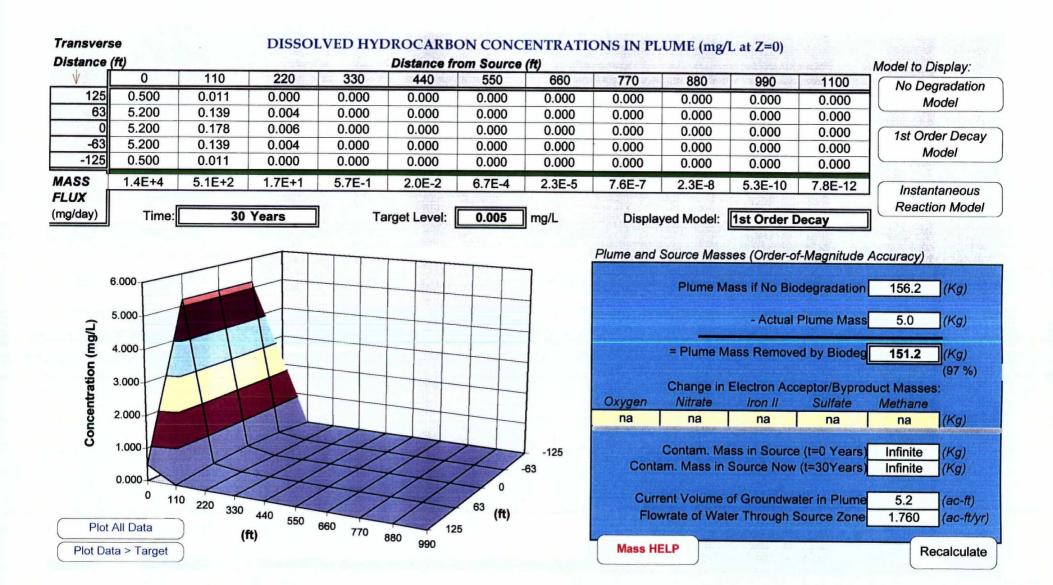
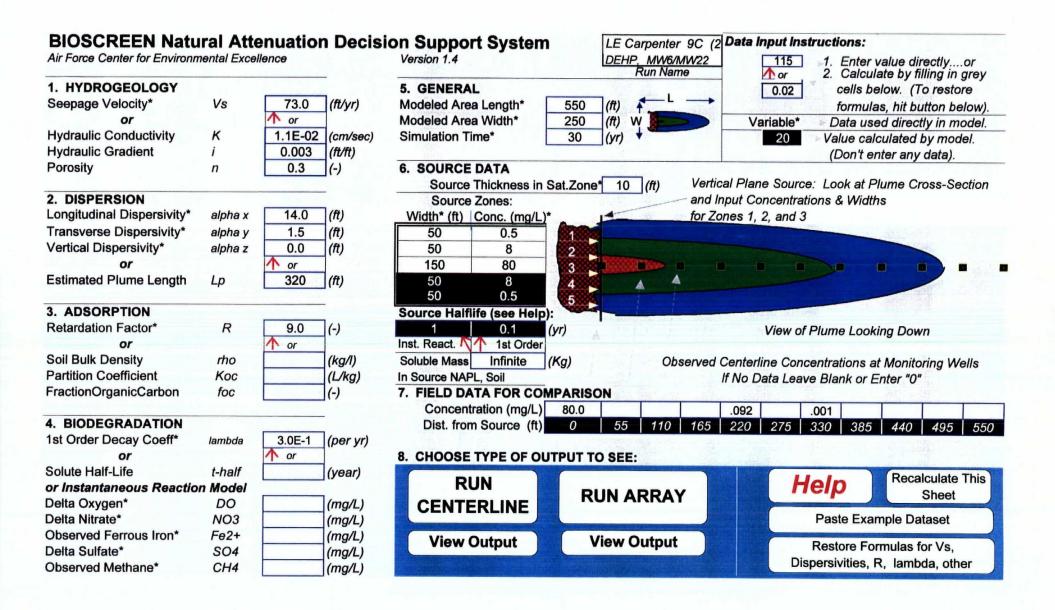
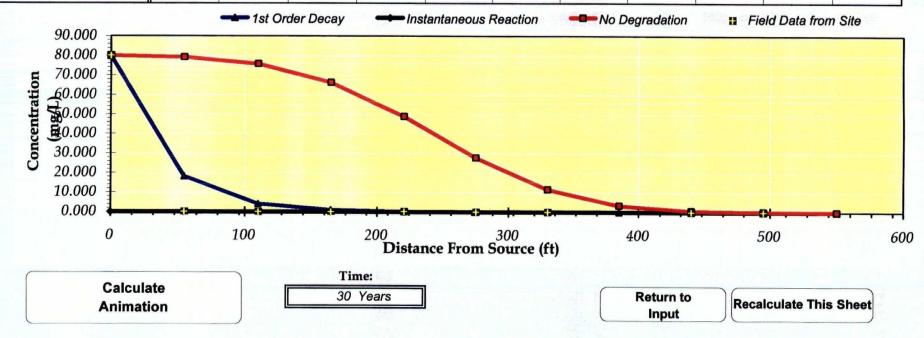


Figure 21b Ethylbenzene Simulation, Parameter Set 1, Centerline Output, Year 2000

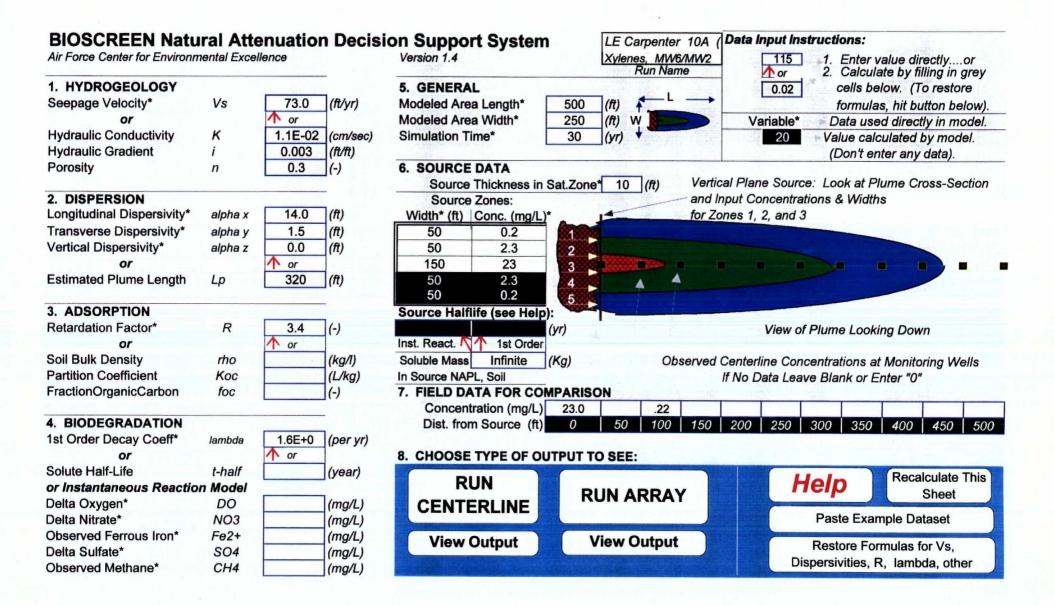




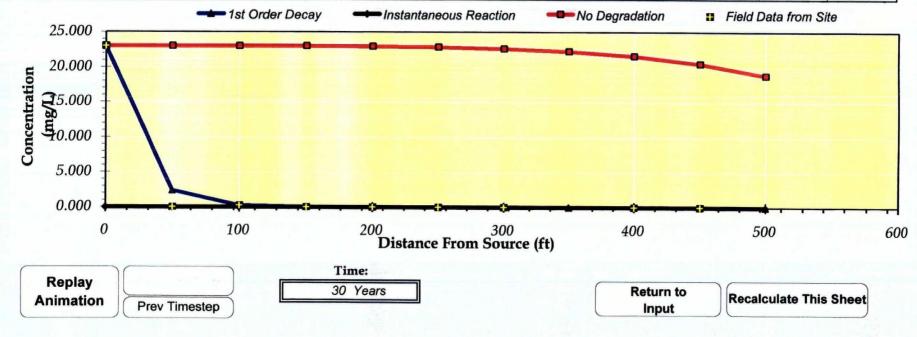
Distance from Source (ft) **TYPE OF MODEL** 55 110 0 165 220 275 330 385 440 495 550 No Degradation 80.000 79.100 75.748 66.251 48.749 27.822 11.568 3.360 0.663 0.088 0.008 1st Order Decay 80.000 18.245 0.214 4.161 0.948 0.010 0.047 0.002 0.000 0.000 0.000 Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 80.0 0.092 0.001



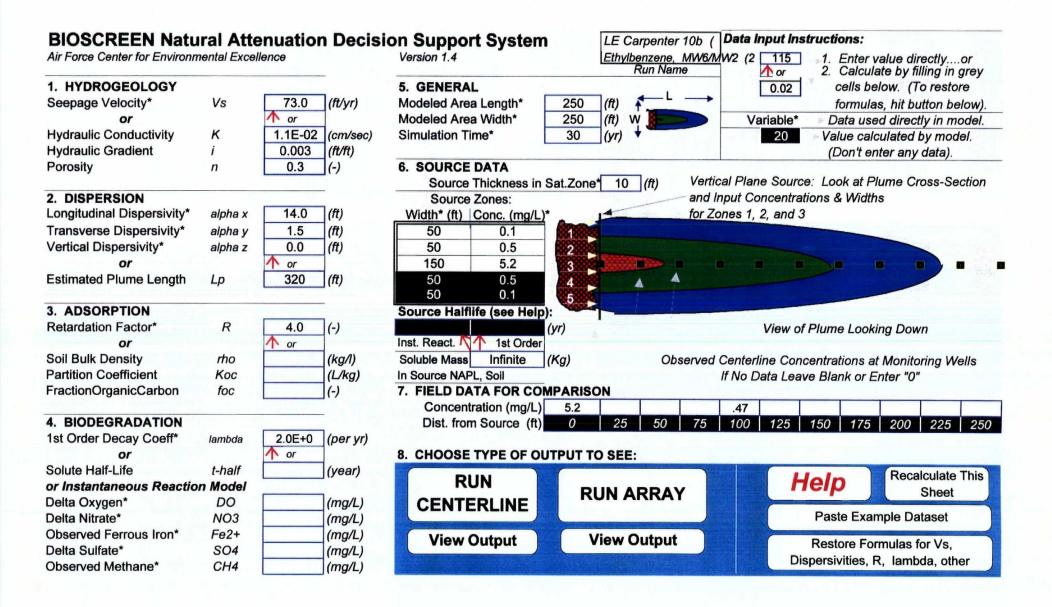
DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) **Transverse** Distance from Source (ft) Distance (ft) Model to Display: 0 55 110 165 220 275 330 440 495 385 550 No Degradation 125 0.970 0.016 8.000 0.232 0.061 0.004 0.001 0.000 0.000 0.000 0.000 Model 63 15.532 80.000 3.241 0.703 0.154 0.033 0.007 0.001 0.000 0.000 0.000 18.245 0.948 80.000 4.161 0.214 0.047 0.010 0.002 0.000 0.000 0.000 1st Order Decay -63 80.000 15.532 3.241 0.703 0.000 0.154 0.033 0.007 0.000 0.001 0.000 Model -125 0.970 0.232 0.061 8.000 0.016 0.004 0.001 0.000 0.000 0.000 0.000 1.2E+4 MASS 2.2E+5 5.4E+4 2.6E+3 5.9E+2 1.3E+2 2.7E+1 4.8E+0 6.9E-1 7.3E-2 5.5E-3 Instantaneous **FLUX** Reaction Model (mg/day) Time: 0.005 30 Years Target Level: mg/L **Displayed Model:** 1st Order Decay Plume and Source Masses (Order-of-Magnitude Accuracy) 80.000 Plume Mass if No Biodegradation 2389.2 (Kg) 70.000 - Actual Plume Mass 53.2 (Kg) Concentration (mg/L) 60.000 = Plume Mass Removed by Biodeo (Kg) 2336.0 50.000 (98 %) 40.000 Change in Electron Acceptor/Byproduct Masses: Nitrate Oxygen Iron II Sulfate Methane 30.000 na (Kg) na na na na 20.000 Contam. Mass in Source (t=0 Years) (Kg) Infinite -125 10.000 Contam. Mass in Source Now (t=30Years) (Kg) Infinite 0.000 0 Current Volume of Groundwater in Plume 7.3 55 (ac-ft) 110 165 63 (ft) Flowrate of Water Through Source Zone 220 1.760 (ac-ft/yr) 275 330 Plot All Data 125 385 (ft) 440 **Mass HELP** Recalculate Plot Data > Target



Distance from Source (ft) TYPE OF MODEL 50 0 100 150 200 250 300 350 400 450 500 No Degradation 23.000 23.000 22.999 22.989 22.944 22.834 22.625 22.251 21.598 20.503 18.803 1st Order Decay 23.000 2.362 0.025 0.003 0.243 0.000 0.000 0.000 0.000 0.000 0.000 Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 23.0 0.220



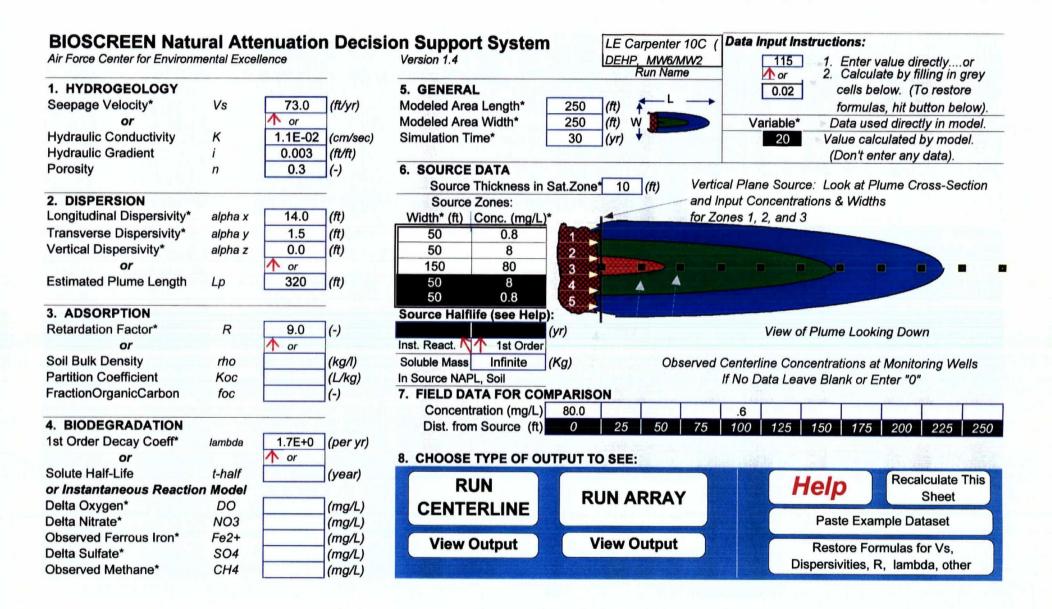
DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) **Transverse** Distance (ft) Distance from Source (ft) Model to Display: 50 0 100 150 200 250 300 350 400 450 500 No Degradation 2.300 125 0.128 0.014 0.002 0.000 0.000 0.000 0.000 0.000 0.000 0.000 Model 63 2.036 23.000 0.191 0.019 0.002 0.000 0.000 0.000 0.000 0.000 0.000 23.000 2.362 0.025 0.003 0.000 0.243 0.000 0.000 0.000 0.000 0.000 1st Order Decay -63 23.000 2.036 0.191 0.019 0.000 0.002 0.000 0.000 0.000 0.000 0.000 Model -125 2.300 0.128 0.014 0.002 0.000 0.000 0.000 0.000 0.000 0.000 0.000 6.3E+4 7.1E+3 MASS 6.9E+2 6.9E+1 7.2E-1 7.4E-2 7.6E-3 7.7E-4 7.0E+0 7.9E-5 8.1E-6 Instantaneous **FLUX** Reaction Model (mg/day) Time: 30 Years Target Level: 0.005 mg/L Displayed Model: 1st Order Decay Plume and Source Masses (Order-of-Magnitude Accuracy) 25.000 Plume Mass if No Biodegradation 687.9 (Kg) - Actual Plume Mass 14.4 (Kg) 20.000 Concentration (mg/L) = Plume Mass Removed by Biodeg (Kg) 673.6 15.000 (98 %) Change in Electron Acceptor/Byproduct Masses: 10.000 Oxygen Nitrate Iron II Sulfate Methane (Kg) na na na na na 5.000 Contam. Mass in Source (t=0 Years) Infinite (Kg) -125 Contam. Mass in Source Now (t=30Years) Infinite (Kg) -63 0.000 0 Current Volume of Groundwater in Plume 3.9 50 (ac-ft) 100 150 (ft) Flowrate of Water Through Source Zone 200 1.760 (ac-ft/yr) 250 300 Plot All Data 125 350 (ft) 400 450 **Mass HELP** Recalculate Plot Data > Target



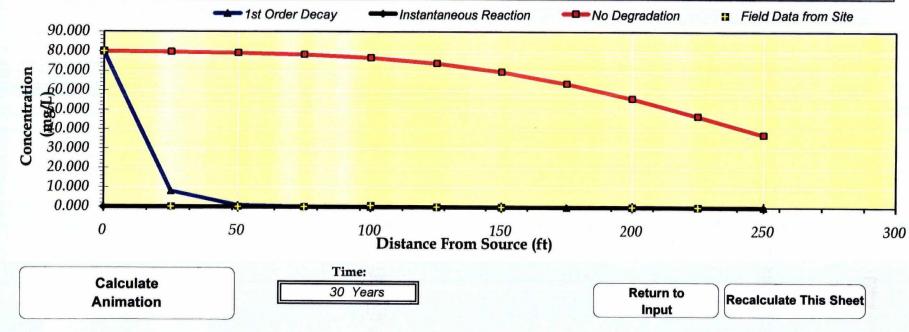
Distance from Source (ft) TYPE OF MODEL 0 25 50 75 100 150 175 125 200 225 250 No Degradation 5.200 5.200 5.200 5.200 5.199 5.195 5.198 5.188 5.129 5.177 5.158 1st Order Decay 5.200 1.169 0.263 0.059 0.013 0.001 0.003 0.000 0.000 0.000 0.000 Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 5.2 0.047



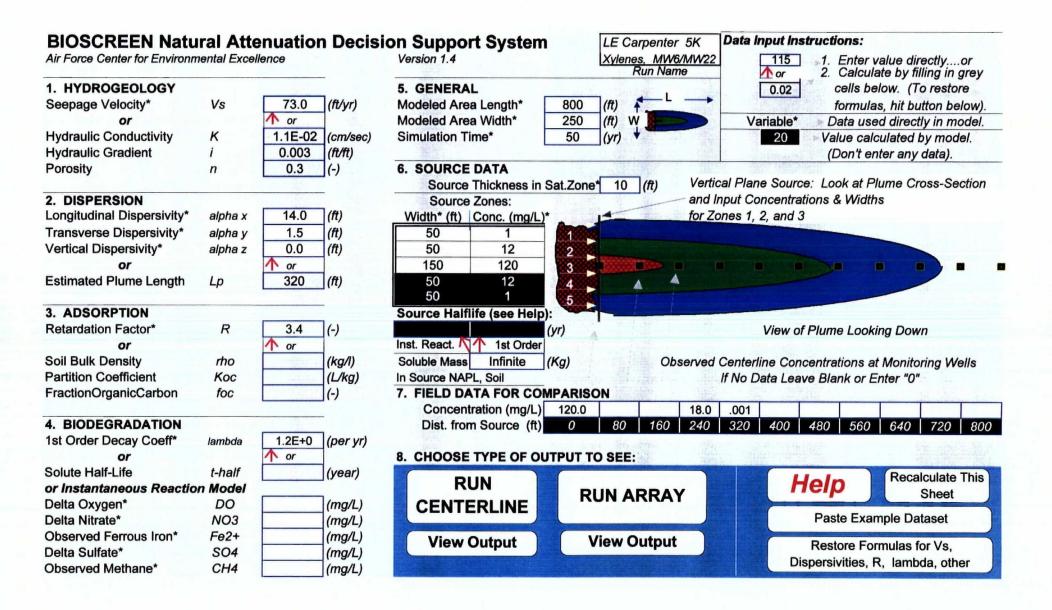
Transverse DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) Distance (ft) Distance from Source (ft) Model to Display: 0 25 50 75 150 175 200 100 225 125 250 No Degradation 125 0.500 0.067 0.000 0.015 0.003 0.001 0.000 0.000 0.000 0.000 0.000 Model 63 5.200 1.090 0.226 0.010 0.048 0.002 0.001 0.000 0.000 0.000 0.000 0.059 0.013 5.200 1.169 0.263 0.003 0.001 0.000 0.000 0.000 0.000 1st Order Decay -63 1.090 0.226 5.200 0.048 0.010 0.002 0.001 0.000 0.000 0.000 0.000 Model -125 0.500 0.067 0.015 0.003 0.001 0.000 0.000 0.000 0.000 0.000 0.000 3.7E+3 7.9E+2 MASS 1.4E+4 1.7E+2 3.8E+1 8.4E+0 4.2E-1 9.4E-2 2.1E-2 1.9E+0 4.7E-3 Instantaneous **FLUX** Reaction Model (mg/day) Time: 30 Years Target Level: 0.005 mg/L Displayed Model: 1st Order Decay Plume and Source Masses (Order-of-Magnitude Accuracy) 6.000 Plume Mass if No Biodegradation 156.2 (Kg) 5.000 - Actual Plume Mass (Kg) 3.4 Concentration (mg/L) 4.000 = Plume Mass Removed by Biodeg (Kg) 152.8 (98 %) 3.000 Change in Electron Acceptor/Byproduct Masses: Nitrate Oxygen Iron II Sulfate Methane 2.000 na na na (Kg) na na 1.000 Contam. Mass in Source (t=0 Years) Infinite (Kg) -125 Contam. Mass in Source Now (t=30Years) Infinite (Kg) -63 0.000 0 Current Volume of Groundwater in Plume 2.3 (ac-ft) 25 50 63 75 Flowrate of Water Through Source Zone 1.760 100 (ac-ft/yr) 125 150 Plot All Data 125 175 (ft) 200 225 **Mass HELP** Plot Data > Target Recalculate



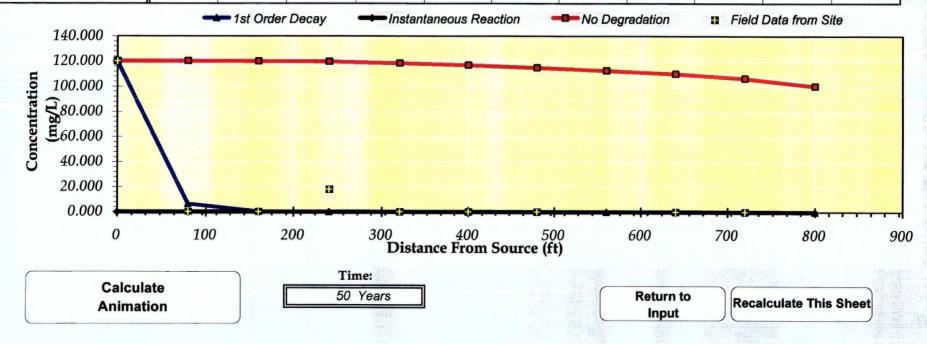
Distance from Source (ft) TYPE OF MODEL 0 25 50 75 100 125 150 175 200 225 250 No Degradation 80.000 79.673 79.233 78.343 76.700 73.925 69.648 63.629 55.905 46.866 37.217 1st Order Decay 80.000 8.072 0.082 0.001 0.814 0.008 0.000 0.000 0.000 0.000 0.000 Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 80.0 0.600



DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) **Transverse** Distance (ft) Distance from Source (ft) Model to Display: 75 100 175 200 0 25 50 125 150 225 250 No Degradation 8.000 125 0.444 0.045 0.005 0.000 0.000 0.000 0.000 0.000 0.000 0.000 Model 63 80.000 7.531 0.702 0.067 0.007 0.001 0.000 0.000 0.000 0.000 0.000 80.000 8.072 0.082 0.000 0.814 0.008 0.000 0.001 0.000 0.000 0.000 1st Order Decay -63 80.000 7.531 0.702 0.067 0.007 0.001 0.000 0.000 0.000 0.000 0.000 Model -125 8.000 0.005 0.444 0.045 0.000 0.000 0.000 0.000 0.000 0.000 0.000 2.2E+5 2.5E+4 2.4E+3 2.4E+2 MASS 2.4E+1 2.4E+0 2.4E-1 2.4E-2 2.4E-3 2.4E-4 2.4E-5 Instantaneous **FLUX** Reaction Model (mg/day) Time: 30 Years Target Level: 0.005 Displayed Model: 1st Order Decay mg/L Plume and Source Masses (Order-of-Magnitude Accuracy) 80.000 Plume Mass if No Biodegradation 2394.8 (Kg) 70.000 21.4 (Kg) - Actual Plume Mass Concentration (mg/L) 60.000 = Plume Mass Removed by Biodeg 2373.4 (Kg) 50.000 (99 %) 40.000 Change in Electron Acceptor/Byproduct Masses: Oxygen Nitrate Iron II Sulfate Methane 30.000 (Kg) na na na na na 20.000 Contam. Mass in Source (t=0 Years) Infinite (Kg) -125 10.000 Contam. Mass in Source Now (t=30Years) Infinite (Kg) -63 0.000 0 Current Volume of Groundwater in Plume 2.3 (ac-ft) 25 50 75 Flowrate of Water Through Source Zone 1.760 100 (ac-ft/yr) 125 150 Plot All Data 125 175 (ft) 200 225 **Mass HELP** Recalculate Plot Data > Target



Distance from Source (ft) **TYPE OF MODEL** 0 80 160 240 320 400 480 560 640 720 800 No Degradation 120.000 116.709 120.000 119.933 119.440 118.325 114.758 112.536 109.875 106.164 100.204 1st Order Decay 120.000 6.286 0.329 0.017 0.001 0.000 0.000 0.000 0.000 0.000 0.000 #VALUE! Inst. Reaction **#VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE! #VALUE!** Field Data from Site 120.0 18.000 0.001



Transverse DISSOLVED HYDROCARBON CONCENTRATIONS IN PLUME (mg/L at Z=0) Distance (ft) Model to Display: Distance from Source (ft) 240 0 80 160 560 640 320 400 480 720 800 No Degradation 125 12.000 0.344 0.021 0.001 0.000 0.000 0.000 0.000 0.000 0.000 0.000 Model 0.245 63 0.012 0.001 0.000 120.000 5.099 0.000 0.000 0.000 0.000 0.000 120.000 6.286 0.329 0.017 0.001 0.000 0.000 0.000 0.000 0.000 0.000 1st Order Decay -63 120.000 5.099 0.245 0.012 0.001 0.000 0.000 0.000 0.000 0.000 0.000 Model -125 12.000 0.344 0.021 0.001 0.000 0.000 0.000 0.000 0.000 0.000 0.000 2.5E+0 MASS 3.3E+5 1.8E+4 9.1E+2 4.7E+1 6.7E-3 3.5E-4 1.8E-5 1.3E-1 9.6E-7 5.0E-8 Instantaneous FLUX Reaction Model (mg/day) Time: 50 Years Target Level: 0.005 **Displayed Model:** 1st Order Decay mg/L Plume and Source Masses (Order-of-Magnitude Accuracy) 120.000 Plume Mass if No Biodegradation 5980.8 (Kg) 100.000 (Kg) - Actual Plume Mass 113.4 Concentration (mg/L) 80.000 = Plume Mass Removed by Biodeg 5867.3 (Kg) (98 %) 60.000 Change in Electron Acceptor/Byproduct Masses: **Nitrate** Oxygen Iron II Sulfate Methane 40.000 (Kg) na na na na na 20.000 Contam. Mass in Source (t=0 Years) (Kg) Infinite -125Contam. Mass in Source Now (t=50Years) Infinite (Kg) -63 0.000 Current Volume of Groundwater in Plume 0 6.2 (ac-ft) 80 160 63 240 (ft) Flowrate of Water Through Source Zone 1.760 320 (ac-ft/yr) 400 480 Plot All Data 125 560 (ft) 640 720 Mass HELP Recalculate Plot Data > Target

Appendix A Laboratory Data Sheets

4th Quarter 1998

Baseline Analysis

Client ID: MW15S

Site: RNA

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Extracted: 12/03/98
Date Analyzed: 12/10/98

GC Column: DB-5

Instrument ID: BNAMS2.i
Lab File ID: s9421.d

Lab Sample No: 99777

Lab Job No: J534

Matrix: WATER Level: LOW

Sample Volume: 920 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

<u>Parameter</u>

bis(2-Ethylhexyl)phthalate

Analytical Result Units: ug/l

ND

Method Detection
Limit
Units: ug/l

Client ID: MW15S

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4019.d Lab Sample No: 99777

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

VOLATILE ORGANICS - GC/PID METHOD 602

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene Toluene Ethylbenzene Xylene (Total)	ND ND ND ND	0.20 0.14 0.14

Client ID: MW15S

Site: RNA

Parameter

Methane

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2090.d

Lab Sample No: 99777

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 Final Volume: 0.0 mL 10.0 ml

Dilution Factor:

5.0

METHANE, ETHANE, ETHENE METHOD 3810

> Analytical Result Units: ug/l

> > 80.4

Quantitation Limit Units: uq/l

Client ID: WPA3

Site: RNA

Lab Sample No: 99778 Lab Job No: J534

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Extracted: 12/03/98
Date Analyzed: 12/10/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9422.d Matrix: WATER Level: LOW

Sample Volume: 980 ml Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

<u>Parameter</u>

bis(2-Ethylhexyl)phthalate

Analytical Result <u>Units: ug/l</u>

69

Method Detection Limit <u>Units: ug/l</u>

Client ID: WPA3

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4020.d Lab Sample No: 99778 Lab Job No: J534

Matrix: WATER

Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

VOLATILE ORGANICS - GC/PID METHOD 602

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50

Client ID: MW15S

Site: RNA

Lab Sample No: 99777 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98

Date Analyzed: 12/10/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9421.d

Matrix: WATER Level: LOW

Sample Volume: 920 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>Parameter</u>

bis (2-Ethylhexyl) phthalate

Method Detection Units: ug/l

Limit Units: ug/l ND

Client ID: MW15S

Site: RNA

Lab Sample No: 99777 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4019.d Matrix: WATER Level: LOW

Purge Volume: 5.0 ml

Final Volume: 0.0 mL

Dilution Factor: 1.0

VOLATILE ORGANICS - GC/PID METHOD 602

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit Units: ug/l
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50

Client ID: MW15S

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2090.d

Lab Sample No: 99777

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor:

5.0

METHANE, ETHANE, ETHENE METHOD 3810

Parameter

Methane

Analytical Result <u>Units: ug/l</u>

80.4

Quantitation Limit Units: ug/l

Client ID: WPA3

Site: RNA

Lab Sample No: 99778 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98
Date Extracted: 12/03/98 Date Analyzed: 12/10/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9422.d

Matrix: WATER Level: LOW

Sample Volume: 980 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Parameter

bis(2-Ethylhexyl)phthalate

Analytical Result Units: ug/l

69

Method Detection Limit Units: uq/l

Client ID: WPA3

Site: RNA

Lab Sample No: 99778 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98

Date Analyzed: 12/04/98 GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4020.d Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit Units: ug/l
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50

Client ID: WPA3

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2080.d

Lab Sample No: 99778

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor:

METHANE, ETHANE, ETHENE METHOD 3810

Parameter

Methane

Analytical Result Units: ug/l

ND

Quantitation Limit Units: ug/l

Client ID: MW17S

Site: RNA

Lab Sample No: 99779 Lab Job No: J534

Matrix: WATER

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98 Date Analyzed: 12/10/98

Level: LOW

GC Column: DB-5

Sample Volume: 1000 ml Extract Final Volume: 2.0 ml

Instrument ID: BNAMS2.i Lab File ID: s9423.d

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>Parameter</u> Units: uq/l

Method Detection Limit Units: ug/l

bis(2-Ethylhexyl)phthalate

6.0

Client ID: MW17S

Site: RNA

Lab Sample No: 99779 Lab Job No: J534

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4021.d Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50

Client ID: MW17S

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2081.d

Lab Sample No: 99779

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml

Final Volume: 0.0 mL

Dilution Factor: 1.0

METHANE, ETHANE, ETHENE METHOD 3810

<u>Parameter</u>

Methane

Analytical Result <u>Units: uq/l</u>

ND

Quantitation Limit Units: uq/l

Client ID: WPC4

Site: RNA

Lab Sample No: 99780 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98 Date Analyzed: 12/12/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9463.d

Matrix: WATER Level: LOW

Sample Volume: 1000 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 5.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

<u>Parameter</u>

bis (2-Ethylhexyl) phthalate

Analytical Result Units: ug/l

540

Method Detection Limit Units: uq/l

Client ID: WPC4

Site: RNA

Lab Sample No: 99780 Lab Job No: J534

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4009.d Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 2.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.40
Toluene	0.87	0.28
Ethylbenzene	26.1	0.28
Xylene (Total)	82.4	1.0

Client ID: WPC4

Site: RNA

Lab Sample No: 99780 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2091.d

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor: 25.0

METHANE, ETHANE, ETHENE METHOD 3810

> Analytical Result Units: ug/l

Quantitation Limit Units: uq/l

996

125

<u>Parameter</u>

Methane

Client ID: WPB10

Site: RNA

Lab Sample No: 99781 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98

Date Extracted: 12/03/98 Date Analyzed: 12/12/98

GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s9464.d

Matrix: WATER Level: LOW

Sample Volume: 970 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 25.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>Parameter</u>

bis(2-Ethylhexyl)phthalate

Units: uq/l

2600

Method Detection Limit

Units: uq/l

28

Client ID: WPB10

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4004.d Lab Sample No: 99781

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume:

5.0 ml

Final Volume: 0.0 mL Dilution Factor:

50.0

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	10.0
Toluene	ND	7.0
Ethylbenzene	478	7.0
Xylene (Total)	2400	25.0

Client ID: WPB10

Site: RNA

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Analyzed: 12/02/98

GC Column: GS-Q

Instrument ID: VSCREEN3.i Lab File ID: scrc2092.d Lab Sample No: 99781

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml

Final Volume: 0.0 mL

Dilution Factor: 200.0

METHANE, ETHANE, ETHENE METHOD 3810

<u>Parameter</u>

Methane

Analytical Result <u>Units: ug/l</u>

6740

Quantitation
Limit
Units: ug/l

1000

Client ID: MW2R

Site: RNA

Lab Sample No: 99782 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98

Date Analyzed: 12/12/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9465.d

Matrix: WATER Level: LOW

Sample Volume: 990 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 50.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

<u>Parameter</u>

bis(2-Ethylhexyl)phthalate

Analytical Result Units: ug/l

9800

Method Detection Limit Units: ug/l

54

Client ID: MW2R

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624
Instrument ID: VOAGC3.i Lab File ID: ipid4010.d Lab Sample No: 99782

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml

Final Volume: 0.0 mL

Dilution Factor: 100.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	20.0
Toluene	ND	14.0
Ethylbenzene	679	14.0
Xylene (Total)	3230	50.0

Client ID: MW2R

Site: RNA

Parameter

Methane

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q

Instrument ID: VSCREEN3.i Lab File ID: scrc2093.d

Lab Sample No: 99782

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor: 200.0

METHANE, ETHANE, ETHENE METHOD 3810

Analytical Result Units: uq/l

6690

Quantitation Limit Units: uq/l

1000

Client ID: WPB6

Site: RNA

Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98 Date Analyzed: 12/12/98

GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s9466.d

Matrix: WATER Level: LOW

Sample Volume: 960 ml

Lab Sample No: 99783

Extract Final Volume: 2.0 ml

Dilution Factor: 250.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>Parameter</u>

bis(2-Ethylhexyl)phthalate

<u>Units: uq/l</u>

42000

Method Detection Limit

Units: ug/l

280

Client ID: WPB6

Site: RNA

Lab Sample No: 99783 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/04/98

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4011.d Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 50.0

25.0

VOLATILE ORGANICS - GC/PID METHOD 602

1310

Method Detection Analytical Result Limit <u>Parameter</u> Units: ug/l Units: ug/l Benzene ND 10.0 Toluene ND 7.0 Ethylbenzene 276 7.0 Xylene (Total)

Client ID: WPB6

Site: RNA

<u>Parameter</u>

Methane

Date Sampled: 12/01/98
Date Received: 12/01/98
Date Analyzed: 12/02/98

GC Column: GS-Q

Instrument ID: VSCREEN3.i Lab File ID: scrc2094.d Lab Sample No: 99783

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor: 250.0

METHANE, ETHANE, ETHENE METHOD 3810

Analytical Result <u>Units: ug/l</u>

9910

Quantitation Limit Units: ug/l

1250

Client ID: Field Blank

Site: RNA

Lab Sample No: 99784 Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Extracted: 12/03/98

Date Analyzed: 12/11/98

GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s9428.d

Matrix: WATER Level: LOW

Sample Volume: 960 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result Parameter <u>Units: uq/l</u>

bis(2-Ethylhexyl)phthalate

5.7

Method Detection Limit

Units: uq/l

Client ID: Field Blank

Site: RNA

Lab Sample No: 99784

Lab Job No: J534

Date Sampled: 12/01/98 Date Received: 12/01/98
Date Analyzed: 12/02/98
GC Column: DB624
Instrument ID: VOAGC3.i

Lab File ID: ipid3993.d

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit Units: ug/l
Benzene Toluene Ethylbenzene	ND ND ND	0.20 0.14 0.14
Xylene (Total)	ND	0.50

Client ID: Field_Blank

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: GS-Q Instrument ID: VSCREEN3.i Lab File ID: scrc2086.d

Lab Sample No: 99784

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 10.0 ml Final Volume: 0.0 mL

Dilution Factor:

1.0

METHANE, ETHANE, ETHENE METHOD 3810

Parameter

Methane

Analytical Result Units: ug/l

ND

Quantitation Limit Units: uq/l

Site: RNA

Lab Job No: J534

Date Sampled: 12/1/98

Date Received: 12/1/98

Date Analyzed: 12/8/98 QA Batch: 1190

Matrix: WATER

FERROUS IRON

Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
99777	MW15s	1.0	0.56
99778	WPA3	1.0	ND
99779	MW17S	1.0	ND
99780	WPC4	5.0	7.6
99781	WPB10	25	32.4
99782	MW2R	25	35.4
99783	WPB6	50	59.8
99784	Field_Blank	1.0	ND

Quantitation Limit for ferrous iron is 1.0 mg/l for an undiluted sample.

Site: RNA

Lab Job No: J534

Date Sampled: 12/1/98
Date Received: 12/1/98

Matrix: WATER

Date Analyzed: 12/2/98

QA Batch: 0153

NITRATE

Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
99777	MW15S	1.0	0.14
99778	WPA3	1.0	0.57
99779	MW17s	1.0	0.15
99780	WPC4	1.0	ND
99781	WPB10	1.0	ND
99782	MW2R	1.0	ND
99783	WPB6	1.0	ND
99784	Field_Blank	1.0	ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.

Site: RNA

Lab Job No: J534

Date Sampled: 12/1/98
Date Received: 12/1/98

Matrix: WATER

Date Analyzed: 12/8/98

QA Batch: 1223

SULFATE

Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
99777	MW15s	1.0	5.3
99778	WPA3	1.0	13.5
99779	MW17S	1.0	15.1
99780	WPC4	1.0	ND
99781	WPB10	1.0	7.6
99782	MW2R	1.0	ND
99783	WPB6	1.0	ND
99784	Field_Blank	1.0	ND

Quantitation Limit for Sulfate is 5.0 mg/l for an undiluted sample.

Site: RNA

Lab Job No: J534

Date Sampled: 12/1/98
Date Received: 12/1/98

Matrix: WATER

Date Analyzed: 12/8/98

QA Batch: 1147

TOTAL PHOSPHORUS

Envirotech Sample #	Client ID	Dilution <u>Factor</u>	Analytical Result <u>Units: mg/l</u>
9977 7	MW15S	1.0	ND
99778	WPA3	1.0	0.1
99779	MW17S	1.0	ND
99780	WPC4	1.0	0.23
99781	WPB10	1.0	0.39
99782	MW2R	1.0	0.3
99783	WPB6	10	0.86
99784	Field_Blank	1.0	ND

Quantitation Limit for Total Phosphorus is 1.5 mg/l for an undiluted sample.

Client ID: Trip_Blank

Site: RNA

Date Sampled: 12/01/98 Date Received: 12/01/98 Date Analyzed: 12/02/98

GC Column: DB624
Instrument ID: VOAGC3.i Lab File ID: ipid3994.d Lab Sample No: 99785

Lab Job No: J534

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50

1st Quarter 1999



Client ID: Trip_Blank Site: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Analyzed: 01/29/99

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4834.d Lab Sample No: 108735

Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50



lient ID: WP-A3
lite: Le Carpenter

Date Sampled: 01/22/99
Date Received: 01/22/99
Date Extracted: 01/26/99
Date Analyzed: 01/27/99

3C Column: DB-5

Parameter

Instrument ID: BNAMS3.i
Lab File ID: t2317.d

ois(2-Ethylhexyl)phthalate

Lab Sample No: 108736

Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 970 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: ug/l

Method Detection

Limit

<u>Units: ug/l</u>

5.4



:lient ID: WP-A3 ite: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Analyzed: 01/29/99

C Column: DB624
Instrument ID: VOAGC3.i Lab File ID: ipid4835.d Lab Sample No: 108736

Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result Units: ug/l	Method Detection Limit <u>Units: ug/l</u>
Benzene Foluene Ethylbenzene Kylene (Total)	ND ND ND ND	0.20 0.14 0.14 0.50



Client ID: MW-17s Site: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Extracted: 01/26/99 Date Analyzed: 01/27/99

<u>Parameter</u>

GC Column: DB-5 Instrument ID: BNAMS3.i Lab File ID: t2318.d

Lab Sample No: 108737

Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 1000 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: uq/l

Method Detection

Limit

Units: ug/l

bis(2-Ethylhexyl)phthalate

6.0



llient ID: MW-17s lite: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Analyzed: 01/29/99

Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4836.d Lab Sample No: 108737 Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Foluene	ND	0.14
Sthylbenzene	ND	0.14
<pre>Kylene (Total)</pre>	ND	0.50



lient ID: WP-C4

;ite: Le Carpenter

Date Sampled: 01/22/99
Date Received: 01/22/99 Date Extracted: 01/26/99 Date Analyzed: 01/27/99

3C Column: DB-5

<u>Parameter</u>

Instrument ID: BNAMS3.i Lab File ID: t2319.d

Lab Sample No: 108738

Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 900 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Method Detection Limit Units: ug/l

Units: ug/l

bis(2-Ethylhexyl)phthalate 25



Client ID: WP-C4 3ite: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Analyzed: 01/29/99 GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4837.d Lab Sample No: 108738 Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

1.0

<u>Parameter</u>	Analytical Result Units: ug/l	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	2.2	0.14
Xylene (Total)	9.0	0.50



lient ID: WP-B6

Site: Le Carpenter

)ate Sampled: 01/22/99 Date Received: 01/22/99 Date Extracted: 01/26/99 Date Analyzed: 01/27/99

C Column: DB-5

<u>Parameter</u>

Instrument ID: BNAMS3.i Jab File ID: t2321.d

Lab Sample No: 108739

Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 1000 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 5.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

> Analytical Result Units: uq/l

Method Detection Limit Units: ug/l

ois(2-Ethylhexyl)phthalate

510

20



:lient ID: WP-B6 lite: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Analyzed: 01/29/99

C Column: DB624
Instrument ID: VOAGC3.i Lab File ID: ipid4829.d Lab Sample No: 108739

Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume:

Final Volume: 0.0 mL

Dilution Factor: 100.0

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit Units: ug/l
3enzene	ND	20
<pre>Foluene</pre>	ND	14
Ethylbenzene	680	14
Kylene (Total)	2700	50



:lient ID: MW-2R

lite: Le Carpenter

Pate Sampled: 01/22/99 Pate Received: 01/22/99 Pate Extracted: 01/26/99 Pate Analyzed: 01/28/99

C Column: DB-5
Instrument ID: BNAMS3.i ab File ID: t2335.d

Lab Sample No: 108740 Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 1000 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 500.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>'arameter</u>

pis(2-Ethylhexyl)phthalate

Units: uq/l

52000

Method Detection

Limit Units: ug/l

2000



Client ID: MW-2R Site: Le Carpenter

Date Sampled: 01/22/99
Date Received: 01/22/99
Date Analyzed: 01/30/99

GC Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4855.d Lab Sample No: 108740

Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml

Final Volume: 0.0 mL

Dilution Factor: 250.0

<u>Parameter</u>	Analytical Result Units: ug/l	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	50
Toluene	ND	35
Ethylbenzene	1100	35
Xylene (Total)	5700	120



lient ID: WP-B10 3ite: Le Carpenter

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Extracted: 01/26/99 Date Analyzed: 01/28/99

GC Column: DB-5
Instrument ID: BNAMS3.i Lab File ID: t2323.d

Lab Sample No: 108741

Lab Job No: K971

Matrix: WATER Level: LOW

Sample Volume: 920 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 5.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result <u>Parameter</u>

bis(2-Ethylhexyl)phthalate

Units: ug/l

420

Method Detection Limit

Units: ug/l

22



lient ID: WP-B10 ite: Le Carpenter

ate Sampled: 01/22/99 Pate Received: 01/22/99 ate Analyzed: 01/30/99

C Column: DB624
Instrument ID: VOAGC3.i ab File ID: ipid4856.d Lab Sample No: 108741

Lab Job No: K971

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 100.0

<u>'arameter</u>	Analytical Result Units: ug/l	Method Detection Limit <u>Units: ug/l</u>
<pre>Genzene Yoluene Ithylbenzene Lylene (Total)</pre>	ND ND 1000 5000	20 14 14 50



Client ID: Field_Blank Site: Le Carpenter

Lab Sample No: 108742 Lab Job No: K971

Date Sampled: 01/22/99 Date Received: 01/22/99 Date Extracted: 01/26/99 Date Analyzed: 01/27/99

Matrix: WATER Level: LOW

Sample Volume: 960 ml Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

GC Column: DB-5

Parameter

Instrument ID: BNAMS3.i Lab File ID: t2320.d

> SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

> > Analytical Result Units: uq/l

Method Detection Limit

Units: uq/l

bis(2-Ethylhexyl)phthalate

ND

4.2



llient ID: Field Blank

Lab Sample No: 108742 3ite: Le Carpenter Lab Job No: K971

Date Sampled: 01/22/99 Date Received: 01/22/99

Date Analyzed: 01/29/99 3C Column: DB624

Instrument ID: VOAGC3.i Lab File ID: ipid4838.d Level: LOW Purge Volume: 5.0 ml Final Volume: 0.0 mL

Matrix: WATER

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result Units: ug/l	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.20
Toluene	ND	0.14
Ethylbenzene	ND	0.14
Xylene (Total)	ND	0.50



Site: Le Carpenter Lab Job No: K971

Date Sampled: 1/22/99

Date Received: 1/22/99 Date Analyzed: 1/25/99 QA Batch: 1198

Matrix: WATER

FERROUS IRON

Envirotech Sample #	Client ID	Dilution <u>Factor</u>	Analytical Result Units: mg/l
108736	WP-A3	1.0	ND
108737	MW-17s	1.0	ND
108738	WP-C4	1.0	0.25
108739	WP-B6	25	37.9
108740	MW-2R	25	19.0
108741	WP-B10	25	22.1
108742	Field_Blank	1.0	ND

Quantitation Limit for ferrous iron is 1.0 mg/l for an undiluted sample.



Site: Le Carpenter

Lab Job No: K971

Date Sampled: 1/22/99
Date Received: 1/22/99

Date Analyzed: 1/23/99 Matrix: WATER QA Batch: 0164

NITRATE

Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
108736	EA-4M	1.0	1.0
108737	MW-17s	1.0	0.17
108738	WP-C4	1.0	0.52
106739	WP-B6	1.0	ND
108740	MW-2R	1.0	ND
108741	WP-B10	1.0	ND
108742	Field_Blank	1.0	ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.

2nd Quarter 1999



Client ID: WP-A3 Site: LE Carpenter

Lab Sample No: 125615

Lab Job No: N593

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Extracted: 04/20/99 Date Analyzed: 04/22/99

Matrix: WATER Level: LOW

Sample Volume: 970 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

GC Column: DB-5

<u>Parameter</u>

Instrument ID: BNAMS2.i Lab File ID: s2060.d

> SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

> > Analytical Result Units: uq/l

Method Detection Limit

Units: uq/l

ND

4.2

bis(2-Ethylhexyl)phthalate



Client ID: WP-A3 Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Analyzed: 04/22/99

GC Column: DB624 Instrument ID: VOAGC3.i Lab File ID: ipid6049.d Lab Sample No: 125615

Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 m Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.31
Toluene	ND	0.34
Ethylbenzene	ND	0.38
Xylene (Total)	ND	0.40



Client ID: WP-C4 Site: LE Carpenter

Lab Sample No: 125616

Lab Job No: N593

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Extracted: 04/20/99

Matrix: WATER Level: LOW

Sample Volume: 980 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

Parameter

Date Analyzed: 04/22/99 GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s2061.d

> SEMI-VOLATILE ORGANICS - GC/MS. METHOD 625

> > Analytical Result <u>Units: ug/l</u>

Method Detection Limit Units: ug/l

bis(2-Ethylhexyl)phthalate

200 B

4.2



Client ID: WP-C4 Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Analyzed: 04/22/99

GC Column: DB624
Instrument ID: VOAGC3.i
Lab File ID: ipid6050.d

Lab Sample No: 125616 Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 1.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.31
Toluene	0.70	0.34
Ethylbenzene	0.48	0.38
Xylene (Total)	1.4	0.40



Client ID: WP-B10 Site: LE Carpenter

Lab Sample No: 125617 Lab Job No: N593

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Extracted: 04/20/99

Matrix: WATER Level: LOW

Sample Volume: 960 ml Extract Final Volume: 2.0 ml

Date Analyzed: 04/22/99 GC Column: DB-5

Instrument ID: BNAMS2.i Lab File ID: s2063.d

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result Parameter Units: uq/l

Method Detection Limit

Units: uq/l

bis(2-Ethylhexyl)phthalate

230 B

4.2



Client ID: WP-B10 Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Analyzed: 04/23/99

GC Column: DB624
Instrument ID: VOAGC2.i
Lab File ID: hpid0511.d

Lab Sample No: 125617

Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor: 25.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	7.8
Toluene	ND	8.5
Ethylbenzene	200	9.5
Xylene (Total)	930	10



Client ID: MW-2R Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Extracted: 04/20/99 Date Analyzed: 04/22/99

Parameter

GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s2075.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 125618

Lab Job No: N593

Matrix: WATER Level: LOW

Sample Volume: 980 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 100.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: uq/l

Method Detection

Limit

Units: uq/l

13000 B 420

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Client ID: MW-2R Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99
Date Analyzed: 04/23/99
GC Column: DB624
Instrument ID: VOAGC2.i

Lab File ID: hpid0512.d

Lab Sample No: 125618

Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

Method Detection Limit <u>Units: ug/l</u>
7.8 8.5 9.5 10



Client ID: WP-B6 Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Extracted: 04/20/99

Date Analyzed: 04/23/99

GC Column: DB-5

Parameter

Instrument ID: BNAMS2.i Lab File ID: s2076.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 125619

Lab Job No: N593

Matrix: WATER Level: LOW

Sample Volume: 1000 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 10.0

SEMI-VOLATILE ORGANICS - GC/MS. METHOD 625

> Analytical Result Units: uq/l

> > 2000 B

Method Detection Limit Units: uq/l

41

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Client ID: WP-B6 Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Analyzed: 04/23/99 GC Column: DB624

Instrument ID: VOAGC2.i Lab File ID: hpid0513.d Lab Sample No: 125619

Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

50.0

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: uq/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	16
Toluene	ND	17
Ethylbenzene	1000	19
Xylene (Total)	3200	20



Client ID: Trip_Blank Site: LE Carpenter

Date Sampled: 04/16/99
Date Received: 04/16/99
Date Analyzed: 04/22/99
GC Column: DB624
Instrument ID: VOAGC3.i

Lab File ID: ipid6051.d

Lab Sample No: 125620 Lab Job No: N593

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

1.0

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0 21
	ND	0.31
Toluene	ND	0.34
Ethylbenzene	ND	0.38
Xylene (Total)	ND	0.40



Client ID: Field Blank

Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99
Date Extracted: 04/20/99

Date Analyzed: 04/22/99

Parameter

GC Column: DB-5
Instrument ID: BNAMS2.i Lab File ID: s2065.d

Lab Sample No: 125621

Lab Job No: N593

Matrix: WATER Level: LOW

Sample Volume: 900 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: uq/l

bis(2-Ethylhexyl)phthalate

ND

Method Detection

Limit

Units: uq/l

4.5

a part of

12



Client ID: Field Blank

Site: LE Carpenter

Date Sampled: 04/16/99 Date Received: 04/16/99 Date Analyzed: 04/22/99

GC Column: DB624 Instrument ID: VOAGC3.i Lab File ID: ipid6052.d Lab Sample No: 125621

Lab Job No: N593

Matrix: WATER

Level: LOW Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

1.0

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: uq/l</u>
Benzene	ND	0.31
Toluene	ND	0.34
Ethylbenzene	ND	0.38
Xylene (Total)	ND	0.40



Site: LE Carpenter

Lab Job No: N593

Date Sampled: 4/16/99

Date Received: 4/16/99

Matrix: WATER

Date Analyzed: 4/23/99 QA Batch: 0187

NITRATE

STL-Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
125615	WP-A3	1.0	0.6
125616	WP-C4	1.0	ND
125617	WP-B10	1.0	ND
125618	MW-2R	1.0	ND
125619	WP-B6	1.0	ND
125621	Field_Blank	1.0	ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.



Site: LE Carpenter

Lab Job No: N593

Date Sampled: 4/16/99
Date Received: 4/16/99

Date Analyzed: 4/22/99 QA Batch: 1231

Matrix: WATER

FERROUS IRON

STL-Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
125615	WP-A3	1.0	ND
125616	WP-C4	1.0	0.79
125617	WP-B10	50	20.8
125618	MW-2R	50	37.2
125619	WP-B6	50	36.7
125621	Field_Blank	1.0	ND

Quantitation Limit for ferrous iron is 0.1 mg/l for an undiluted sample.

3rd Quarter 1999



Client ID: WP-A3

Site: LE Carpenter-RNA

Date Sampled: 07/23/99
Date Received: 07/23/99
Date Extracted: 07/27/99
Date Analyzed: 07/31/99

GC Column: DB-5

<u>Parameter</u>

Instrument ID: BNAMS2.i
Lab File ID: s3990.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 145804

Lab Job No: R742

Matrix: WATER Level: LOW

Sample Volume: 960 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

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Analytical Result

<u>Units: uq/l</u>

ND

Method Detection
Limit
<u>Units: ug/l</u>

4.2

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Client ID: WP-A3

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Analyzed: 07/30/99

GC Column: DB624
Instrument ID: VOAGC2.i Lab File ID: hpid1621.d Lab Sample No: 145804

Lab Job No: R742

Matrix: WATER

Level: LOW Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	0.31
Toluene	ND	0.34
Ethylbenzene	ND	0.38
Xylene (Total)	ND	0.40



Client ID: WP-B10

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Extracted: 07/27/99

Date Analyzed: 08/02/99

<u>Parameter</u>

GC Column: DB-5
Instrument ID: BNAMS2.i Lab File ID: s4013.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 145805

Lab Job No: R742

Matrix: WATER Level: LOW

Sample Volume: 980 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 25.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

3

Analytical Result Units: uq/l

2200

Method Detection Limit

Units: uq/l

100

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Client ID: WP-B10

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Analyzed: 07/31/99

GC Column: DB624 Instrument ID: VOAGC2.i Lab File ID: hpid1628.d Lab Sample No: 145805

Lab Job No: R742

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL

Dilution Factor: 100.0

<u>Parameter</u>	Analytical Result <u>Units: uo/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	31
Toluene	57	34
Ethylbenzene	200	38
Xylene (Total)	1400	40



Client ID: MW2R

Site: LE Carpenter-RNA

Date Sampled: 07/23/99
Date Received: 07/23/99
Date Extracted: 07/27/99
Date Analyzed: 08/02/99

GC Column: DB-5

<u>Parameter</u>

Instrument ID: BNAMS2.i
Lab File ID: s4014.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 145806

Lab Job No: R742

Matrix: WATER Level: LOW

Sample Volume: 970 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 100.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result Units: ug/l

16000

Method Detection
Limit
Units: uq/l

420

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Severn Trent Services Inc.



Client ID: MW2R

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Analyzed: 08/02/99

GC Column: DB624 Instrument ID: VOAGC2.i Lab File ID: hpid1647.d Lab Sample No: 145806

Lab Job No: R742

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor: 50.0

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	16
Toluene	ND	17
Ethylbenzene	380	19
Xylene (Total)	1700	20



Client ID: WP-B6

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Extracted: 07/27/99

<u>Parameter</u>

Date Analyzed: 08/02/99 GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s4015.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 145807

Lab Job No: R742

Matrix: WATER

Level: LOW Sample Volume: 990 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 500.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: ug/l

Method Detection

Limit

<u>Units: ug/l</u>

39000

2100

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Client ID: WP-B6

Site: LE Carpenter-RNA

Date Sampled: 07/23/99 Date Received: 07/23/99 Date Analyzed: 08/02/99

GC Column: DB624 Instrument ID: VOAGC2.i Lab File ID: hpid1648.d Lab Sample No: 145807

Lab Job No: R742

Matrix: WATER Level: LOW

Purge Volume: 5.0 Final Volume: 0.0 mL 5.0 ml

Dilution Factor:

Parameter	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene	ND	31
Toluene	ND	34
Ethylbenzene	170	38
Xylene (Total)	1600	. 40



Site: LE Carpenter-RNA

Lab Job No: R742

Date Sampled: 7/23/99

Date Received: 7/23/99

Date Analyzed: 7/26/99 QA Batch: 1261

Matrix: WATER

FERROUS IRON

STL-Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
145804	WP-A3	1.0	ND
145605	WP-B10	50	42.7
145806	MW2R	50	44.7
145807	WP-B6	50	60.2

Quantitation Limit for Ferrous Iron is 0.1 mg/l for an undiluted sample.

X-Method 3500-Fe (Standard Methods for the Examination of Water and Wastewater, 18th Edition) specifies that ferrous iron be determined "at sampling site because of the possibility of change in the ferrous-ferric ratio with time in acid solutions p_3-67 .



Site: LE Carpenter-RNA

Lab Job No: R742

Date Sampled: 7/23/99
Date Received: 7/23/99

Date Analyzed: 7/26/99

Matrix: WATER

QA Batch: 0232

NITRATE

STL-Envirotech Sample #	Client ID	Dilution <u>Factor</u>	Analytical Result Units: mg/l
145804	WP-A3	1.0	0.38
145805	WP-B10	1.0	ND
145806	MW2R	1.0	ND
145807	WP-B6	1.0	ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.



Client ID: WP-C1 Site: L.E. Carpenter

Date Sampled: 07/28/99 Date Received: 07/28/99 Date Extracted: 07/31/99 Date Analyzed: 08/05/99

<u>Parameter</u>

GC Column: DB-5 Instrument ID: BNAMS2.i Lab File ID: s4083.d

Lab Sample No: 146533

Lab Job No: R874

Matrix: WATER Level: LOW

Sample Volume: 780 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: uq/l

Method Detection

Limit

Units: uq/l

bis(2-Ethylhexyl)phthalate

16

5.2



Client ID: WP-C1 Site: L.E. Carpenter

Date Sampled: 07/28/99 Date Received: 07/28/99 Date Analyzed: 08/03/99 GC Column: DB624

Instrument ID: VOAGC2.i Lab File ID: hpid1673.d Lab Sample No: 146533

Lab Job No: R874

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Limit Units: ug/l
Benzene Toluene Ethylbenzene Xylene (Total)	ND ND ND ND	0.31 0.34 0.38 0.40



Site: L.E. Carpenter

Lab Job No: R874

Date Sampled: 7/28/99
Date Received: 7/28/99

Date Analyzed: 8/10/99

Matrix: WATER

QA Batch: 1264

FERROUS IRON

STL-Envirotech Sample #	Client ID	Dilution <u>Factor</u>	Analytical Result Units: mg/l
146533	WP-C1	25	13.8

Quantitation Limit for Ferrous Iron is 0.1 mg/l for an undiluted sample.

X-Method 3500-Fe (Standard Methods for the Examination of Water and Wastewater, 18th Edition) specifies that ferrous iron be determined "at sampling site because of the possibility of change in the ferrous-ferric ratio with time in acid solutions" p3-67.



Site: L.E. Carpenter

Date Sampled: 7/28/99 Date Received: 7/28/99

Matrix: WATER

Lab Job No: R874

Date Analyzed: 7/28/99

QA Batch: 0232

NITRATE

STL-Envirotech

Sample #

Client ID

Dilution Factor

Analytical Result

Units: mg/l

146533

WP-C1

1.0

ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.



Client ID: MW15S

Site: L.E. Carpenter-RNA

Site: D.E. Carpenter-RNA

Date Sampled: 07/22/99
Date Received: 07/22/99
Date Extracted: 07/24/99
Date Analyzed: 08/03/99

GC Column: DB-5

Parameter

Instrument ID: BNAMS3.i
Lab File ID: t5798.d

Lab Sample No: 145569

Lab Job No: R707

Matrix: WATER Level: LOW

Sample Volume: 980 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

1

Analytical Result

Units: uq/l

Method Detection

Limit

Units: ug/l

bis(2-Ethylhexyl)phthalate

ND

4.2



Client ID: MW17S

Site: L.E. Carpenter-RNA

Date Sampled: 07/22/99 Date Received: 07/22/99

Date Extracted: 07/24/99 Date Analyzed: 08/03/99

<u>Parameter</u>

GC Column: DB-5 Instrument ID: BNAMS3.i Lab File ID: t5799.d

bis(2-Ethylhexyl)phthalate

Lab Sample No: 145570

Lab Job No: R707

Matrix: WATER Level: LOW

Sample Volume: 970 ml

Extract Final Volume: 2.0 ml

Dilution Factor: 1.0

SEMI-VOLATILE ORGANICS - GC/MS METHOD 625

Analytical Result

Units: uq/l

ND

Method Detection

Limit

Units: ug/l

4.2



Client ID: MW17S

Site: L.E. Carpenter-RNA

Lab Sample No: 145570

Lab Job No: R707

Date Sampled: 07/22/99 Date Received: 07/22/99 Date Analyzed: 07/30/99

GC Column: DB624
Instrument ID: VOAGC2.i
Lab File ID: hpid1620.d

Matrix: WATER Level: LOW

Purge Volume: 5.0 ml Final Volume: 0.0 mL

Dilution Factor:

<u>Parameter</u>	Analytical Result <u>Units: ug/l</u>	Method Detection Limit <u>Units: ug/l</u>
Benzene Toluene Ethylbenzene	ND ND ND	0.31 0.34 0.38
Xylene (Total)	ND ND	0.40



Site: L.E. Carpenter-RNA

Lab Job No: R707

Date Sampled: 7/22/99
Date Received: 7/22/99

Date Received: 7/22/
Matrix: WATER

Date Analyzed: 7/26/99

QA Batch: 1259

FERROUS IRON

STL-Envirotech Sample #	Client ID	Dilution <u>Factor</u>	Analytical Result Units: mg/l
145569	MW15S	1.0	1.5
145570	MW17s	1.0	ND

Quantitation Limit for Ferrous Iron is 0.1 mg/l for an undiluted sample.

X-Method 3500-Fe (Standard Methods for the Examination of Water and Wastewater, 18th Edition) specifies that ferrous iron be determined "at sampling site because of the possibility of change in the ferrous-ferric ratio with time in acid solutions" p3-67.



Site: L.E. Carpenter-RNA

Lab Job No: R707

Date Sampled: 7/22/99
Date Received: 7/22/99

Matrix: WATER

Date Analyzed: 7/23/99 QA Batch: 0234

NITRATE

STL-Envirotech Sample #	Client ID	Dilution Factor	Analytical Result Units: mg/l
145569	MW15s	1.0	0.18
145570	MW17s	1.0	ND

Quantitation Limit for Nitrate is 0.1 mg/l for an undiluted sample.